



Experiments in two and three dimensions

A confocal microscopy study

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MPI Polymer Research, Mainz

Twente, March 2016

Before I start



Physics at Interfaces Max Planck Institute for Polymer Research



Prof. Dr. Hans-Jürgen Butt

Director



Dr. Günter Auernhammer

Project Leader



Prof. Dr. Markus Mezger

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Project Leader [\[more\]](#)



PD Dr. Doris Vollmer

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Dr. Michael Kappl

Project Leader [\[more\]](#)



Dr. Stefan Weber

Project Leader [\[more\]](#)



Dr. Kaloian Koynov

Project Leader [\[more\]](#)



Dr. Si Wu

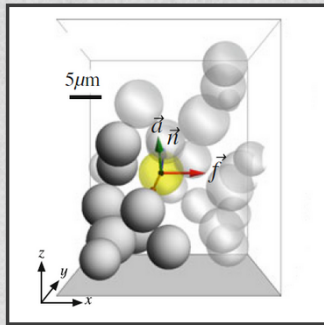
Project Leader

Auernhammer group

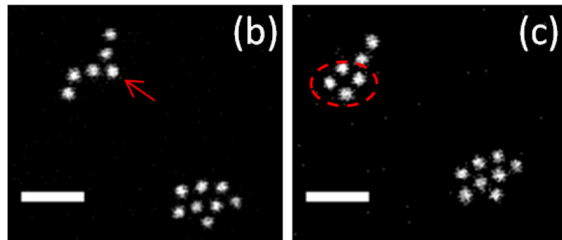
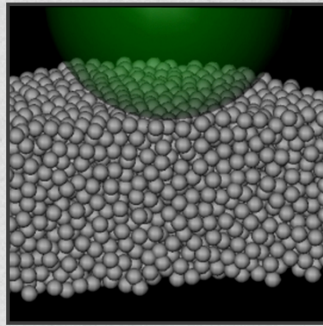


• Colloids

Dynamical processes
in granular matter

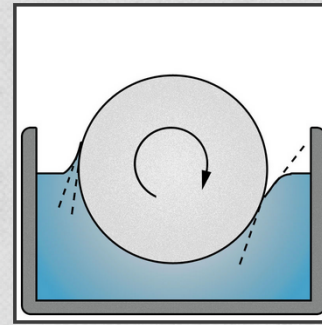


Mechanical properties
of colloidal aggregates

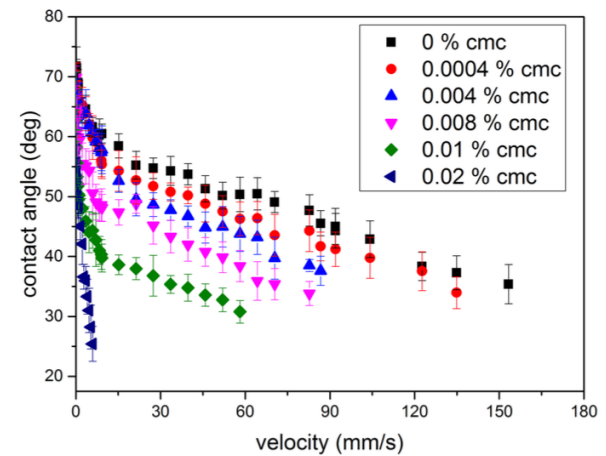
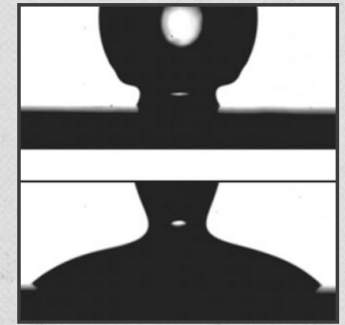


• Dynamic Wetting

Wetting
of surfactant solutions



Fast wetting by miscible
and immiscible liquids





Thanks ...

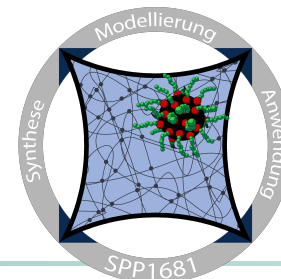
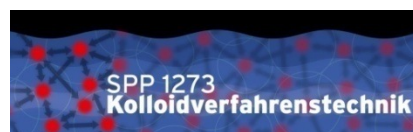
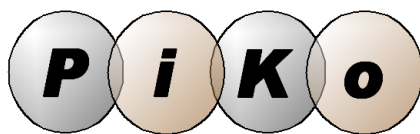
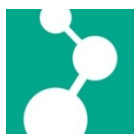
MPI Mainz:

- Laurent Gilson
- Shilin Huang
- Marcel Roth
- Jennifer Wenzl
- Ryohei Seto
- Rene Stangenberg

- Phillip Lellig
- Stefan Geiter
- Gaby Schäfer

Other:

- Carsten Schilde (TU Braunschweig)
- Arno Kwade (TU Braunschweig)
- Heiko Briesen (TU München)
- Stefan Luding (U Twente)
- Andreas Menzel (U Düsseldorf)
- Christian Holm (U Stuttgart)





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Twente, March 2016

Why particles?



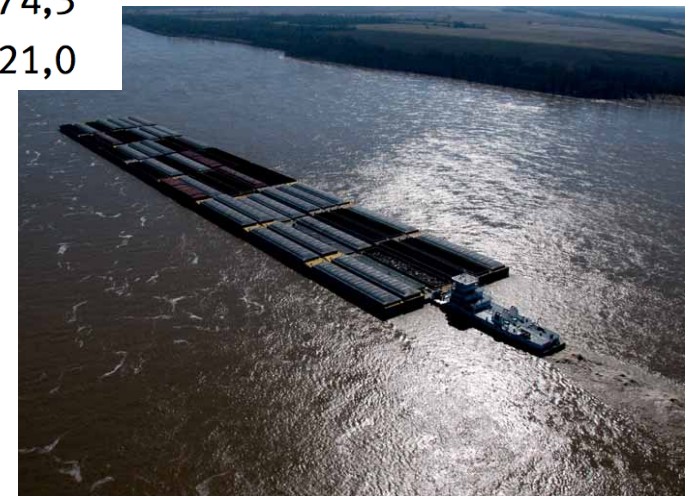
- Transported goods as bulk solids (granulates)
 - > 50% on inland water ways

July 2012

	10 ⁶ t km
Flüssiges Massengut.....	1 087,0
Schüttgut.....	2 939,6
Stückgut.....	574,3
20-Fuß-Container.....	221,0

Statistisches Bundesamt

- Paints, coatings, etc.

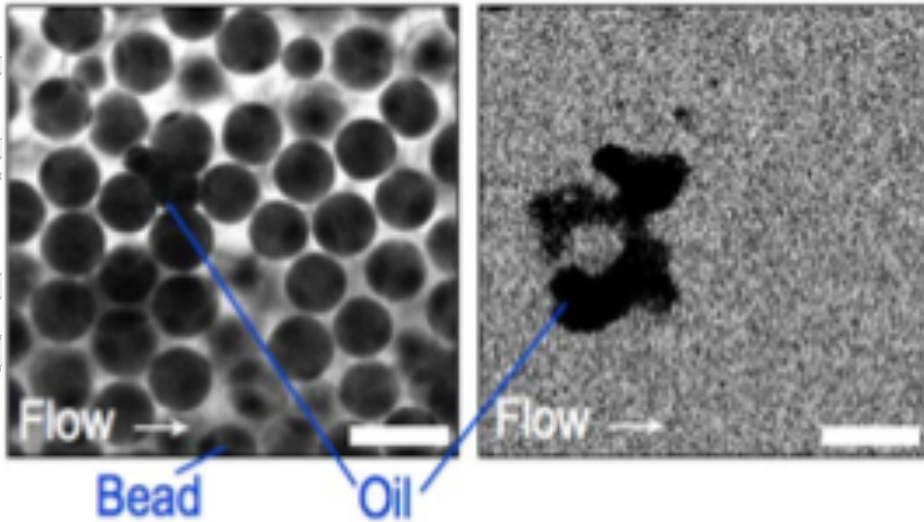


<http://www.drycargomag.com>

Why wet particles?

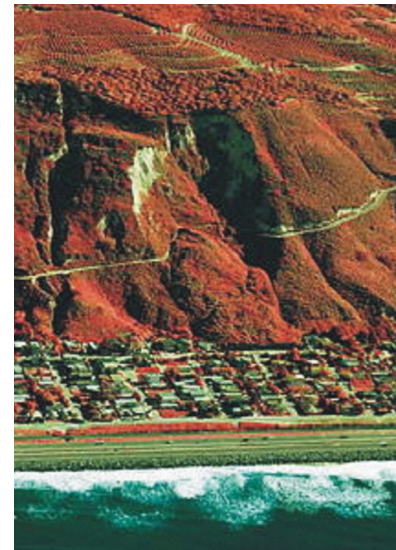


Flow through porous media
Clogging



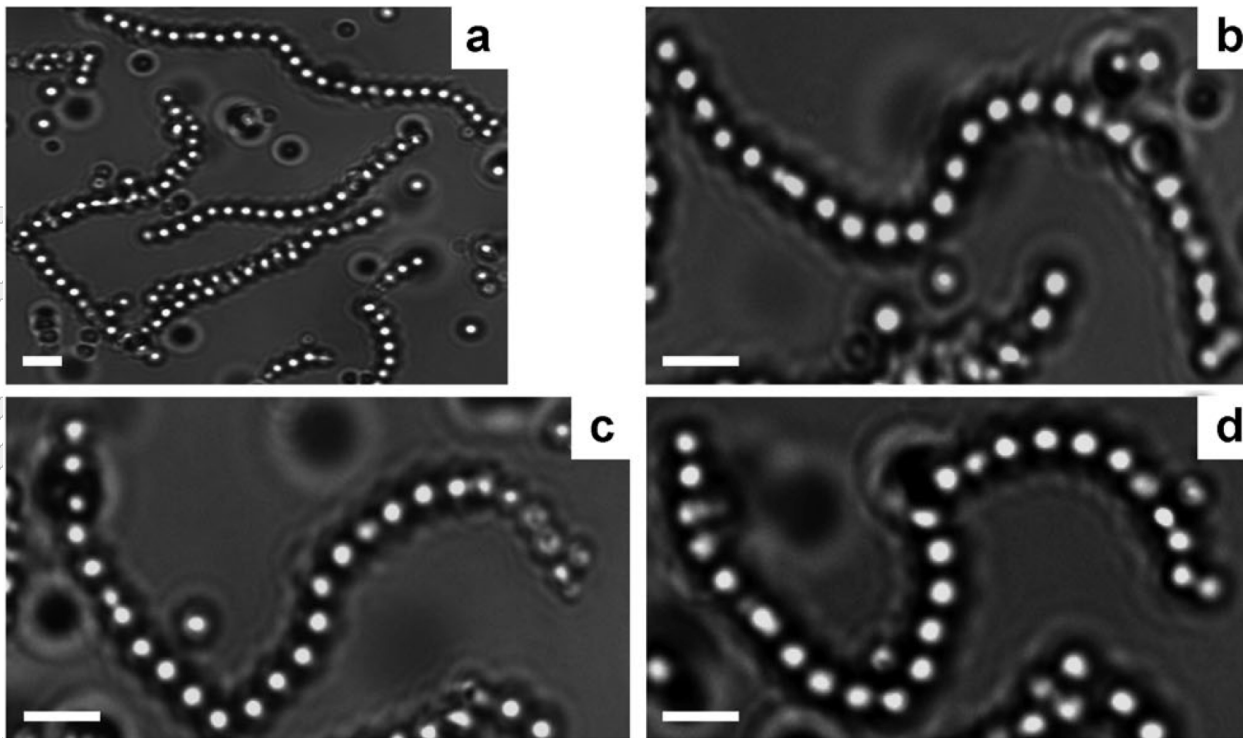
A. T. Krummel, S. S. Datta, S. Münster, D. A. Weitz,
AIChE Journal **59**, 1022 (2013).

oil sand



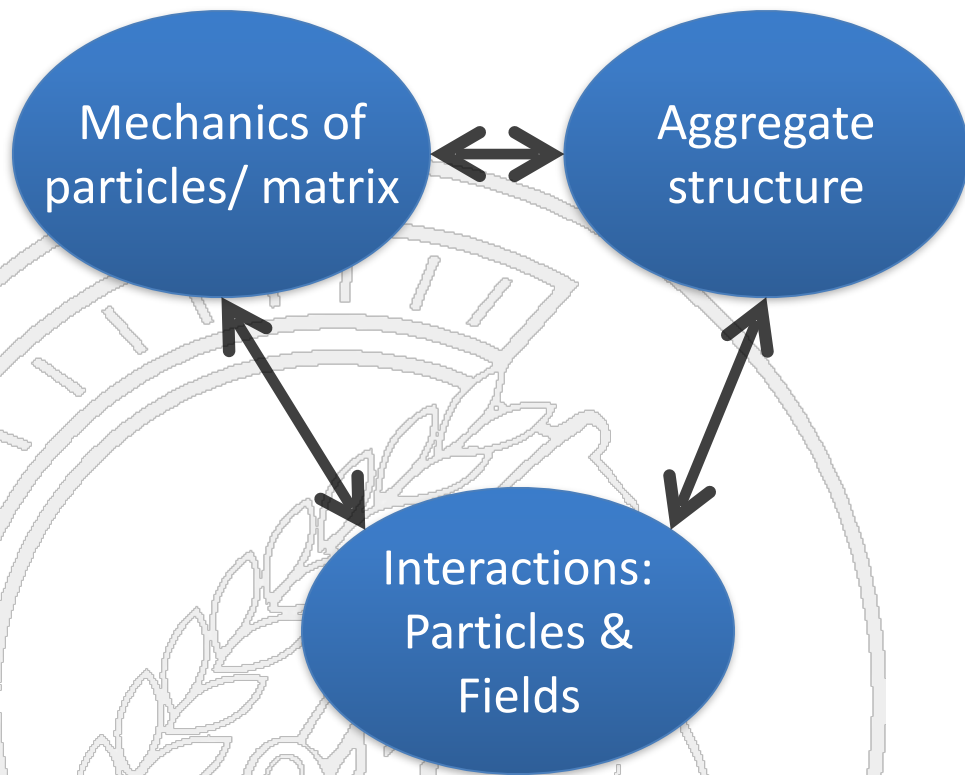
Landslides

Particles as model atoms



Vutukuri, H. R., A. F. Demirörs, B. Peng, P. D. J. van Oostrum, A. Imhof and A. van Blaaderen (2012). "Colloidal Analogues of Charged and Uncharged Polymer Chains with Tunable Stiffness." Angewandte Chemie International Edition **51**(45): 11249-11253.

Bottom up approach to mechanics of particulate systems



... and others

Particles



Colloids

1 μm

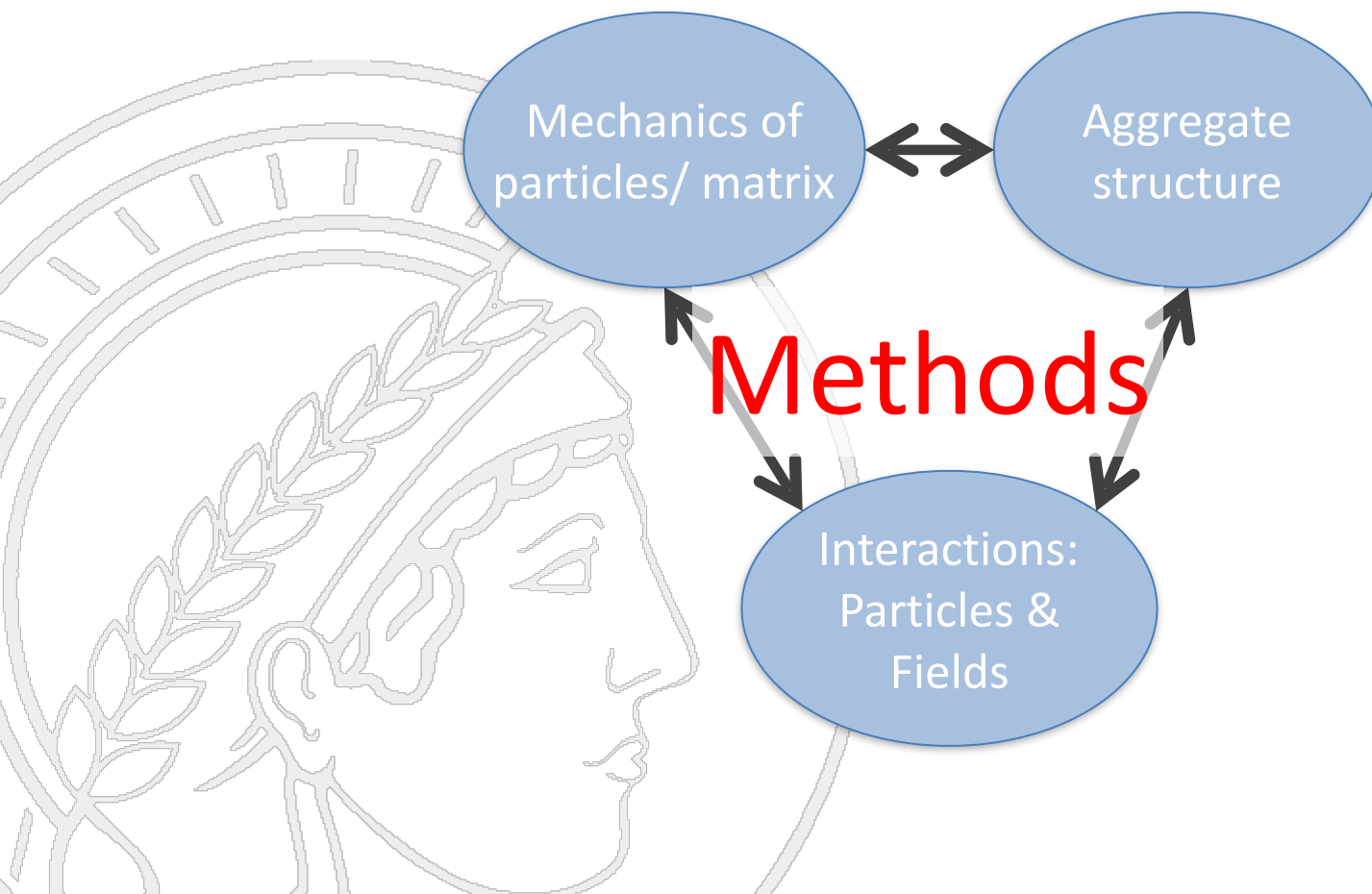


en.wikipedia.org



www.i-c-m.at

?



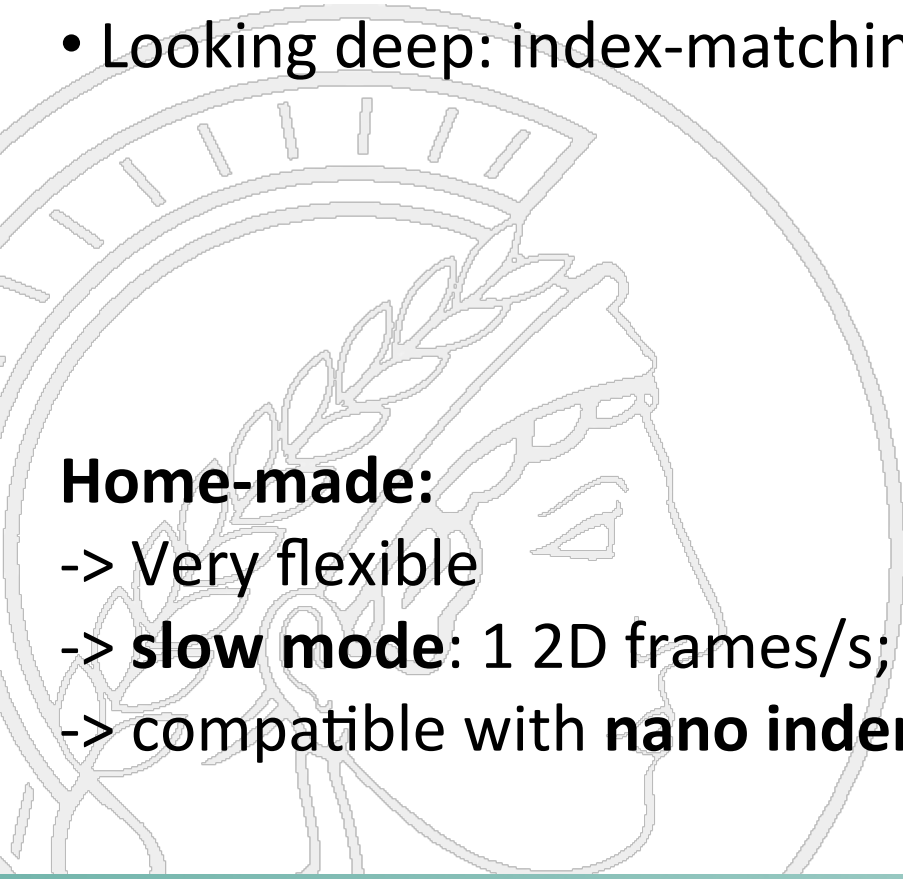
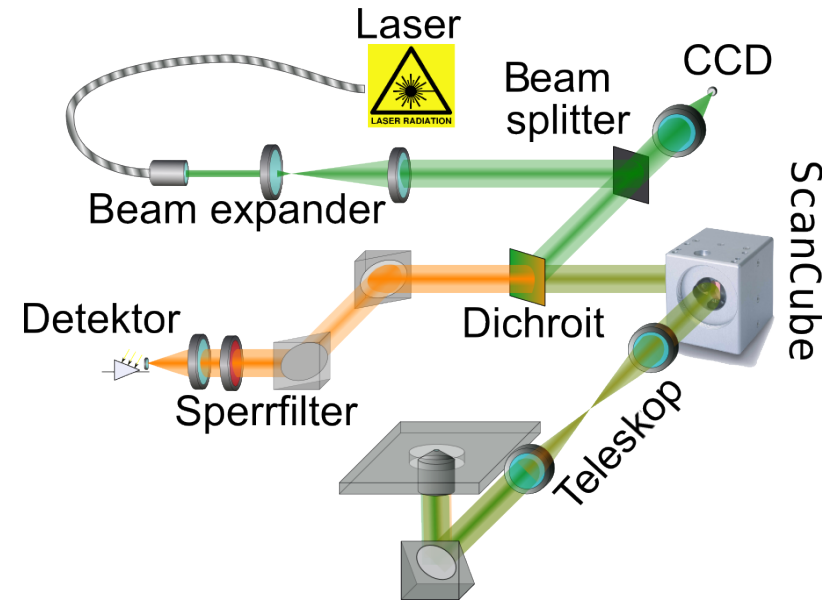
Methods I

Requirements:

- Fluorescent particles
- Looking deep: index-matching

Home-made:

- > Very flexible
- > **slow mode**: 1 2D frames/s; **fast mode**: ~ 1000 2D frames/s
- > compatible with **nano indenter** and **piezo rheometer**



Methods II

Sample

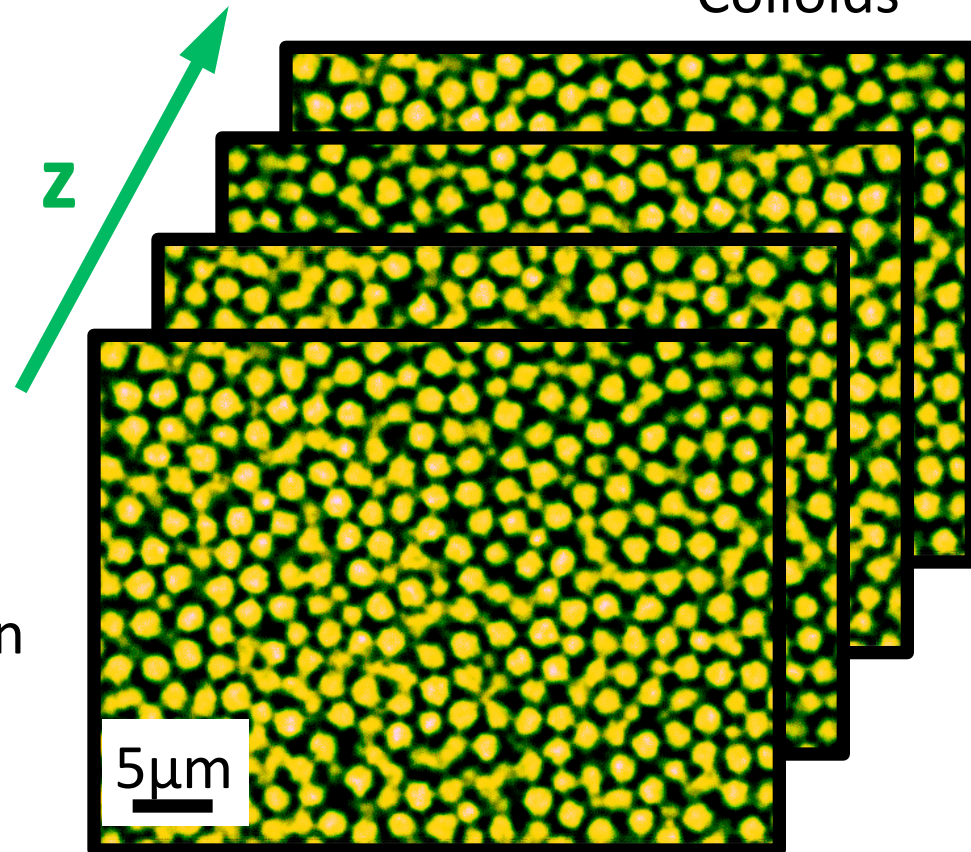
- Fluorescently labeled colloids
- Index-matching liquid
- Imaging layer by layer

Determining the structure

- Colloids = local maxima
 - Finding the colloids
- Better than optical resolution



Colloids

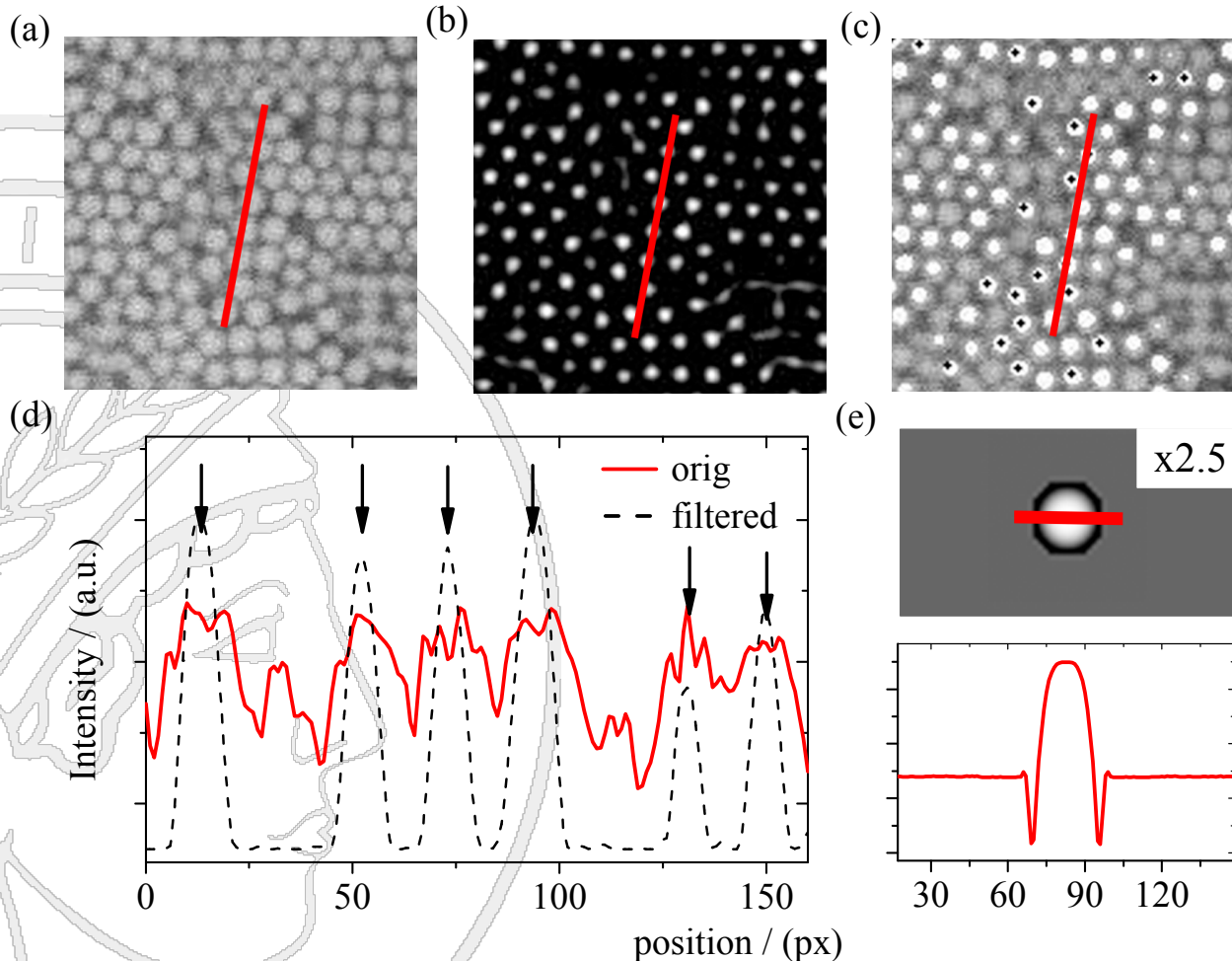


Crocker, Grier, *J. Coll. Interf. Sci.* **(1996)** 179, 298
Wenzl, et al., *Granul Matt* 15 **(2013)**: 391-400.

Reconstructing the structure



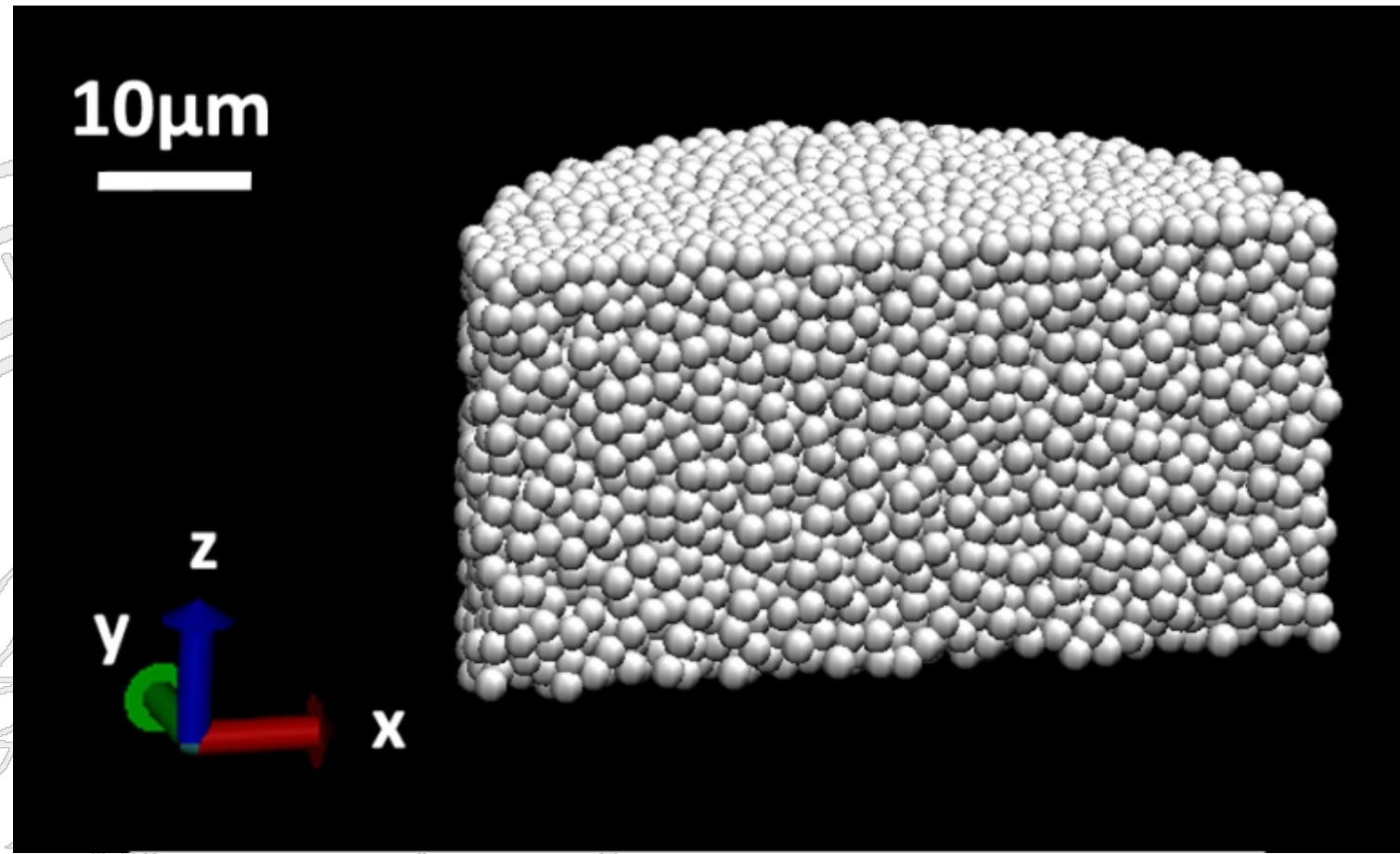
Original \Rightarrow Convolution \Rightarrow Detect max



Profiles

Kernel

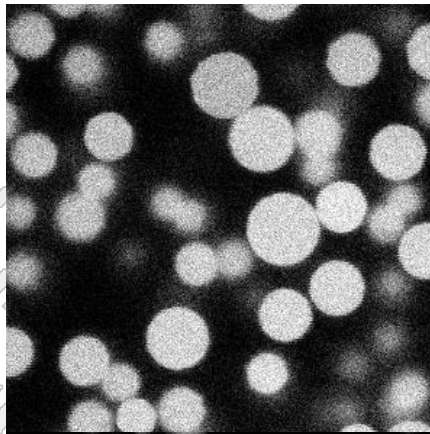
Reconstructing the structure



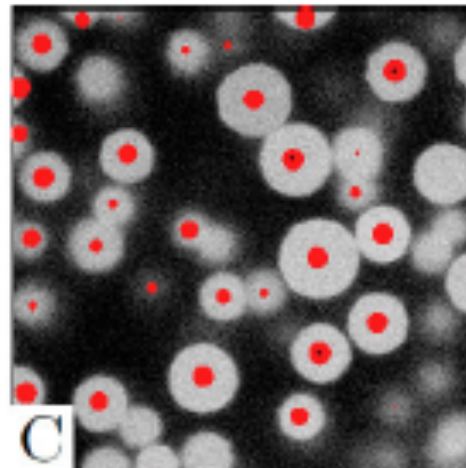
Data analysis: polydispersity



How to find polydisperse particles

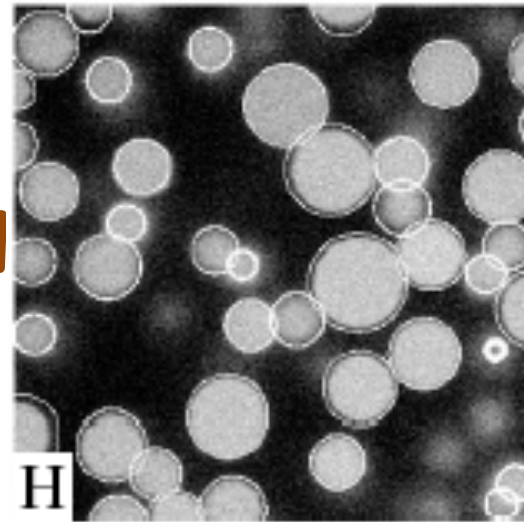
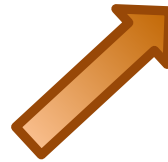


Image



Localization in 3D

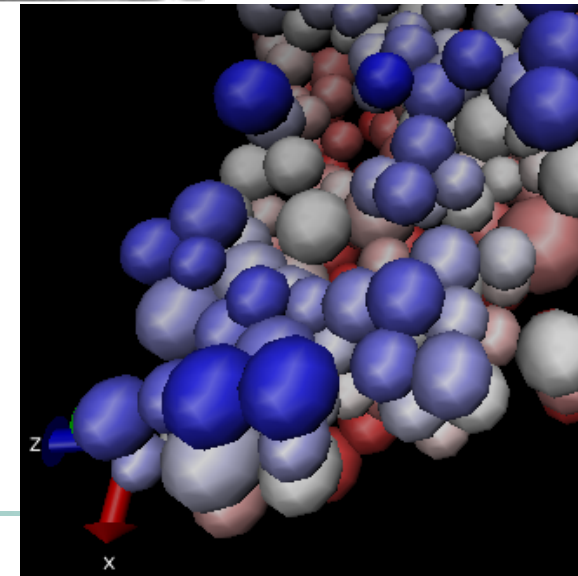
Diameter
of particles

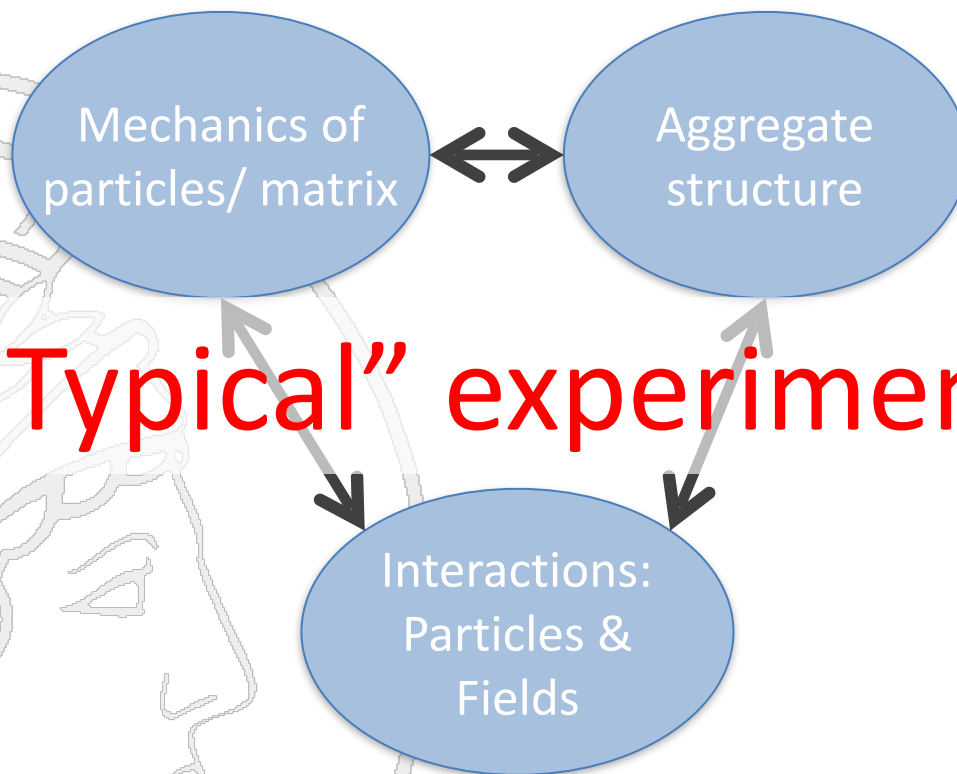


H



Visualization





"Typical" experiment

“Hard” polymer spheres: electrostatics



PMMA colloids:

Diameter: 1.6 μm

Fluorescently labeled

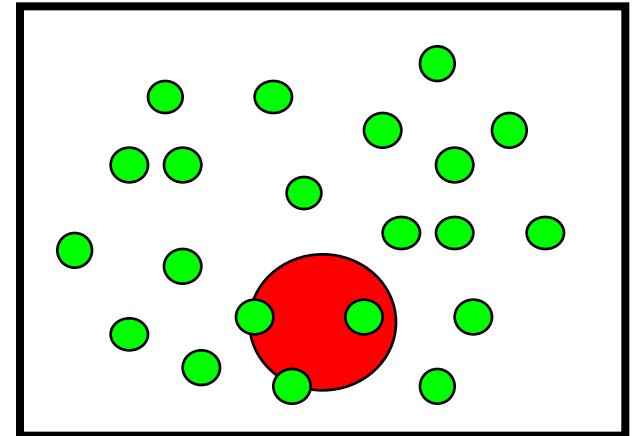
Magnetic colloids:

Diameter 4.4 μm

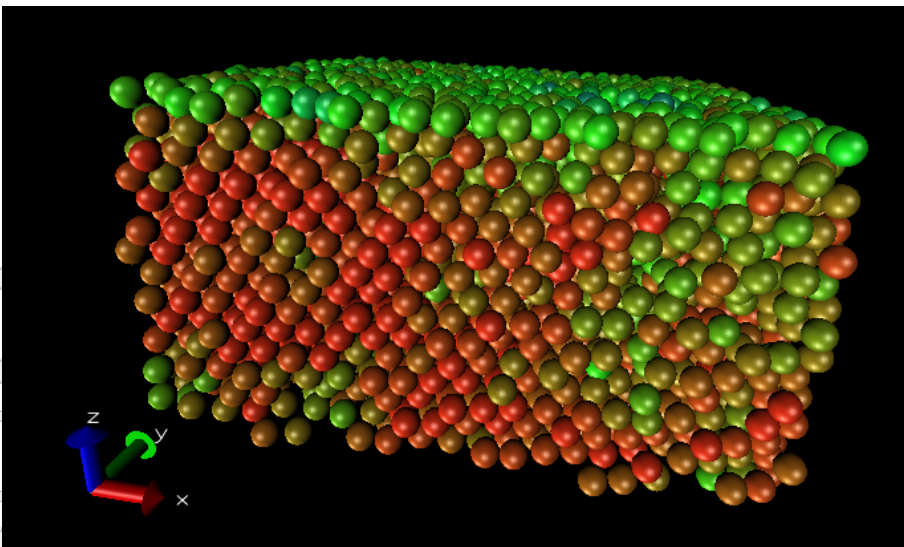
Concentration Magn./PMMA $\sim 10^{-5}$

 Colloidal crystals with defects

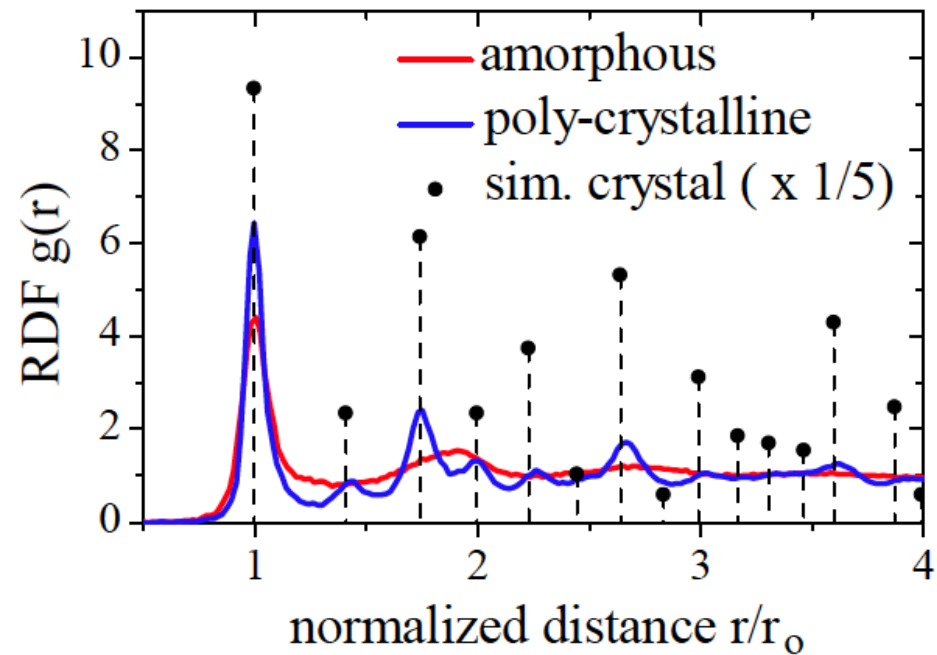
Confocal microscopy => Dynamics of defects



Reconstruction

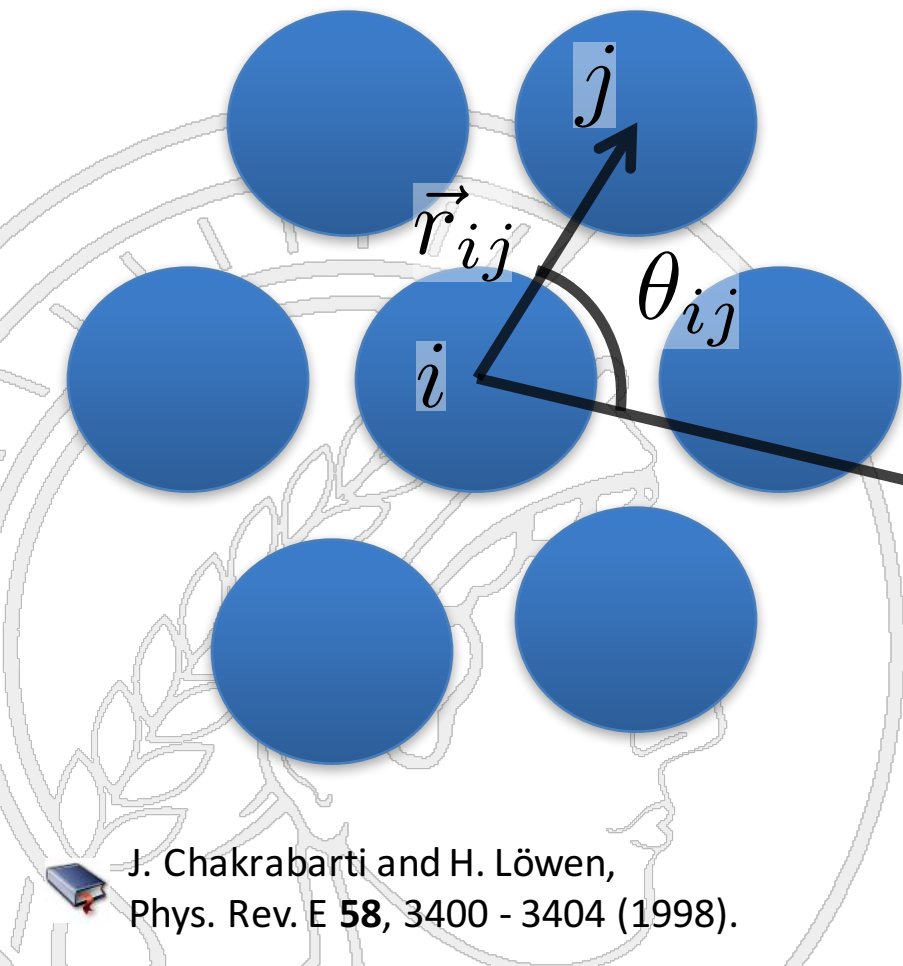


Radial distribution function



Number of nearest neighbors:
closer than $1.2 r_0$

Degree of order in 2D



$$\Psi_6 = \frac{1}{N} \sum_{i=1}^N \underbrace{\frac{1}{6} \sum_j \exp(i6\theta_{ij})}_{\text{Projecting bond vectors to nearest neighbors}}$$

Projecting
bond vectors
to nearest
neighbors

Averaging over small volume

J. Chakrabarti and H. Löwen,
Phys. Rev. E **58**, 3400 - 3404 (1998).

Degree of order in 3D



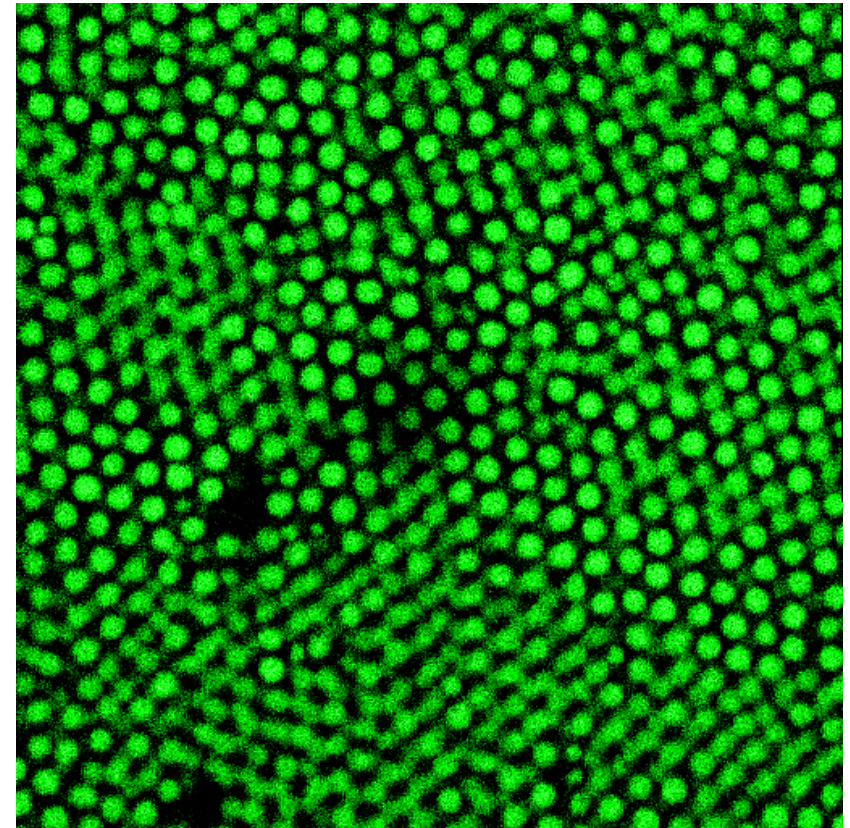
Bond order parameter in 3D

$$Q_{lm}(\vec{r}) \equiv Y_{lm}(\theta(\vec{r}), \phi(\vec{r}))$$

$$Q_l \equiv \left[\frac{4\pi}{2l+1} \sum_{m=-l}^l |\bar{Q}_{lm}|^2 \right]^{1/2}$$



Steinhardt, et al., *Phys Rev B* **28**, 784 (1983).



Generating defects in a crystal

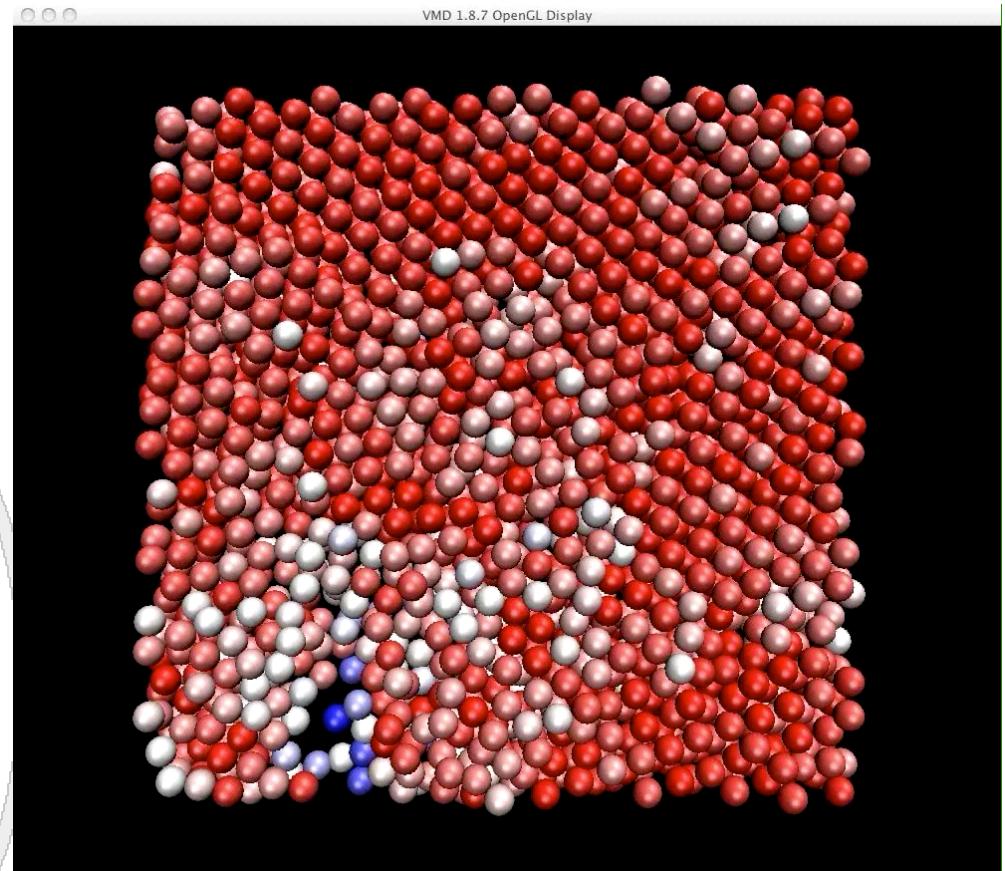


Visualization or experiment

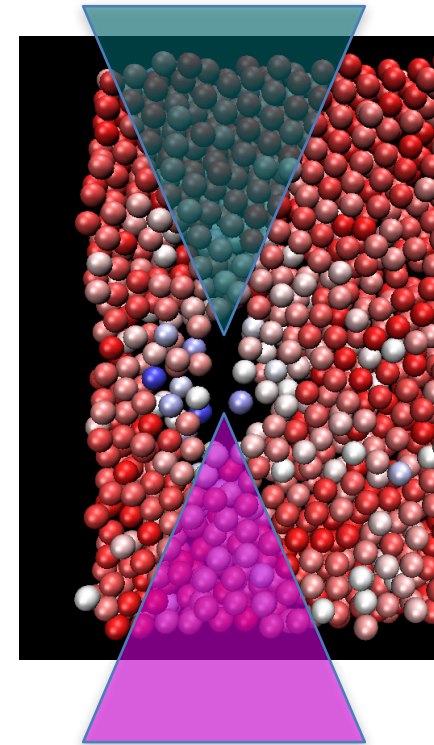
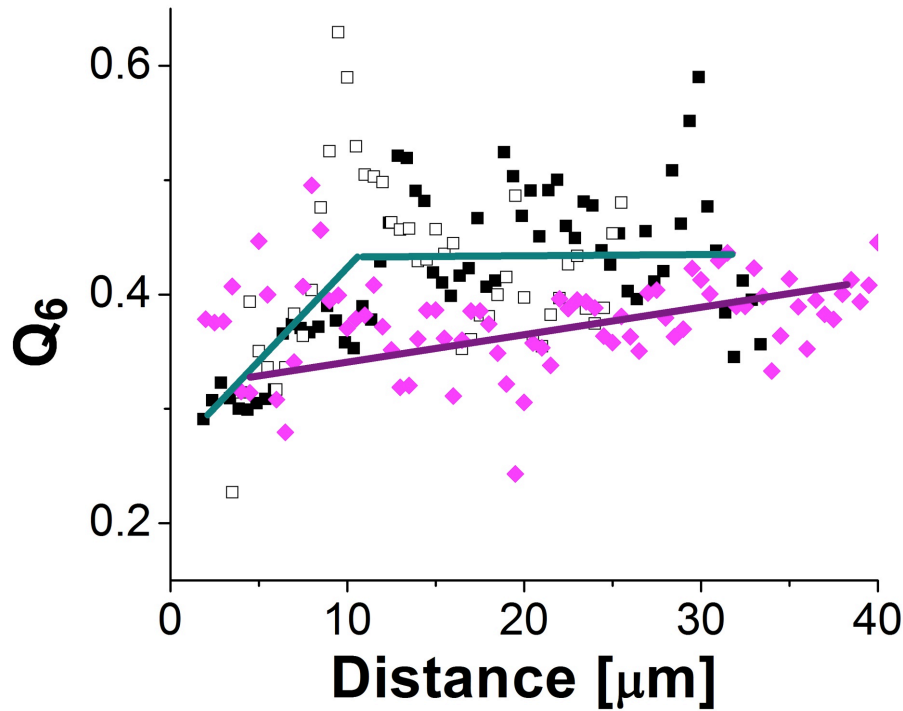
3D confocal sequence

Chain of magnetic particles
in colloidal crystal

Magnetic field gradient



Crystal order vs. distance

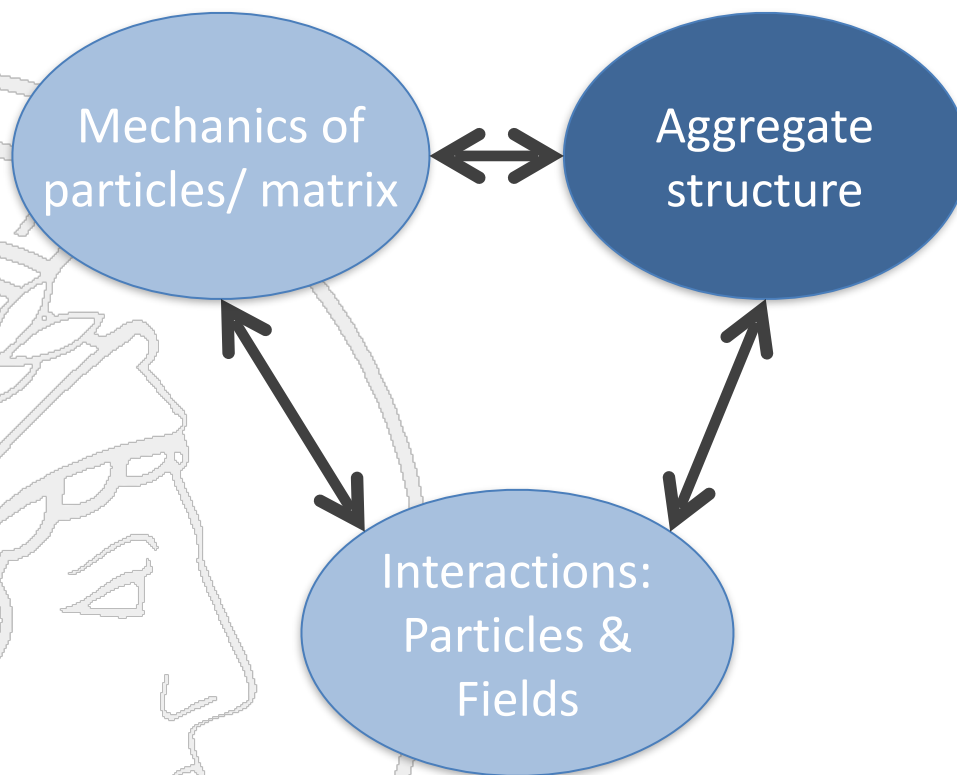


→ Analysis of model atoms

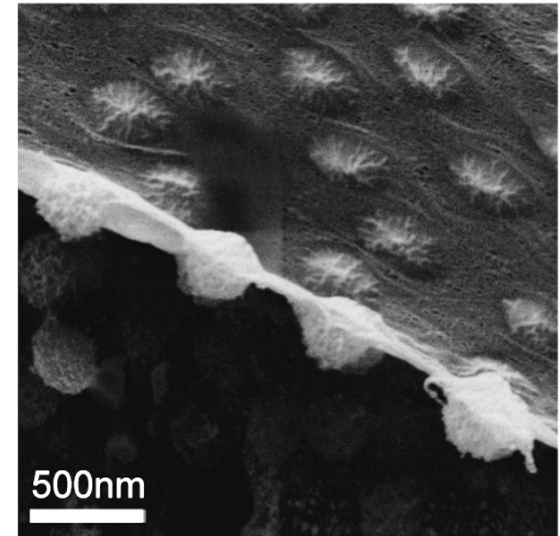
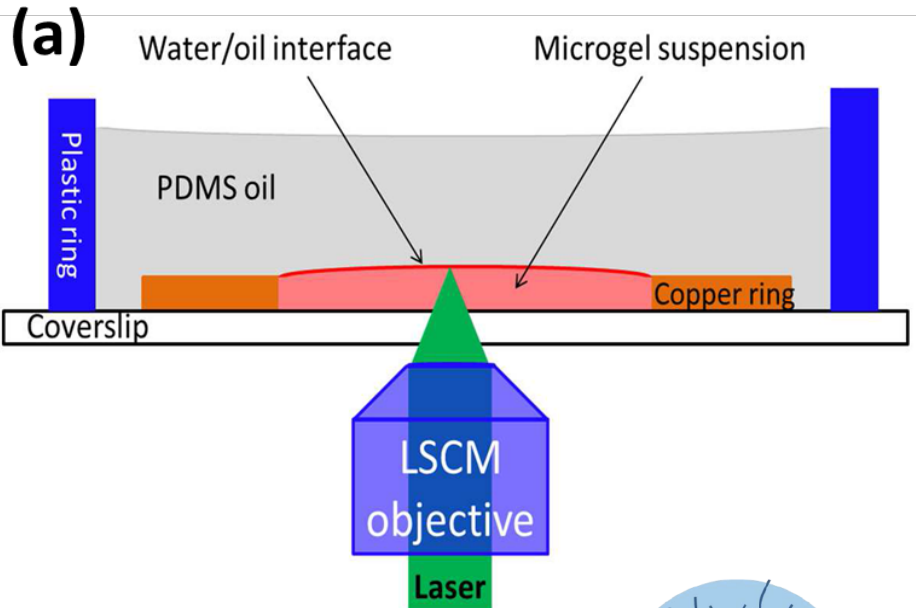
→ Colloidal crystal melts locally and recrystallizes slowly



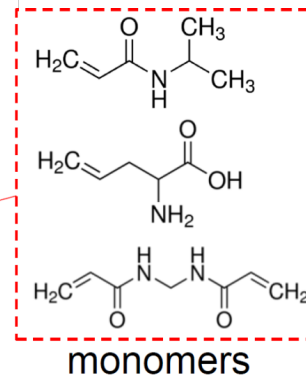
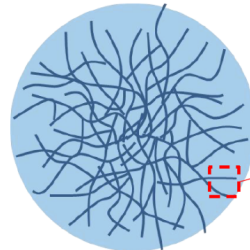
Structure vs. motion 2D



Soft particles at interfaces

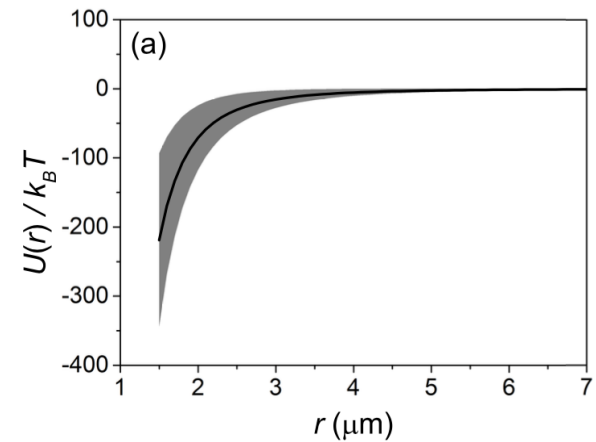
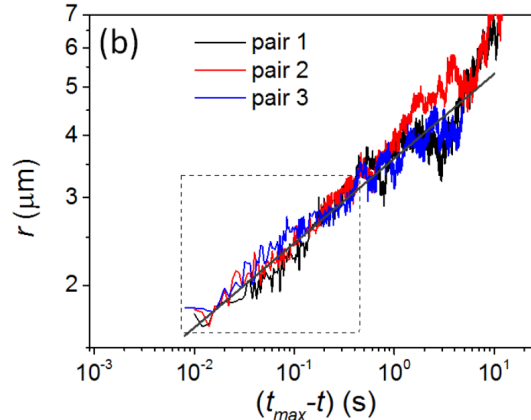
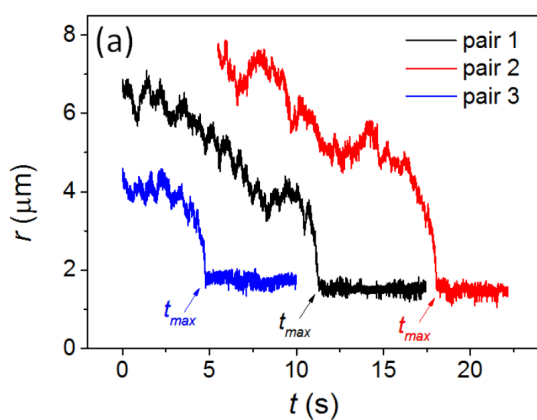
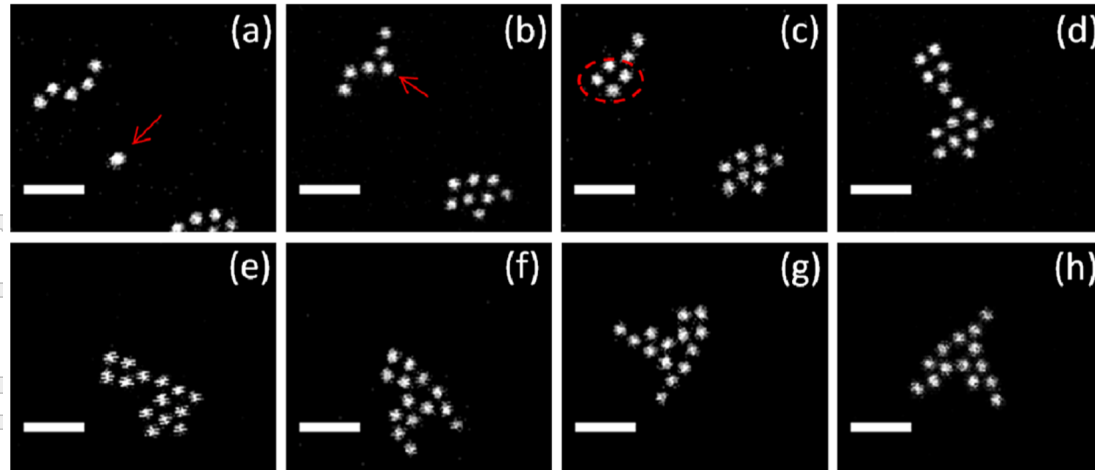


Style, R. W., et al. (2015). "Adsorption of soft particles at fluid interfaces." *Soft Matter* **11**(37): 7412-7419.



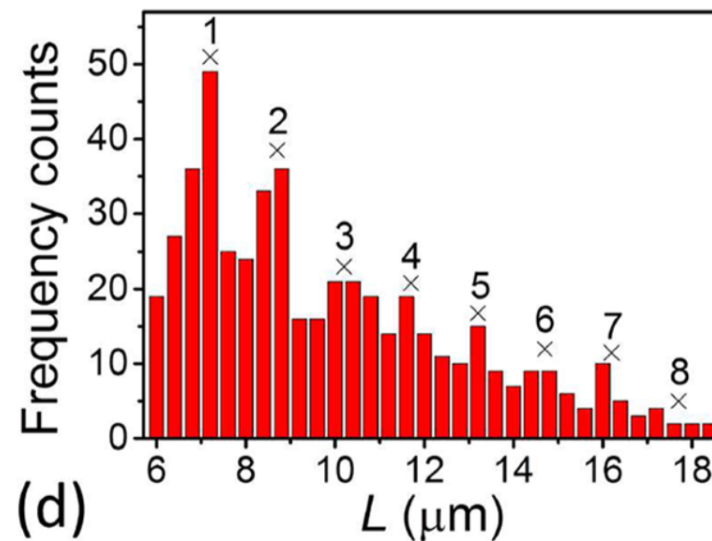
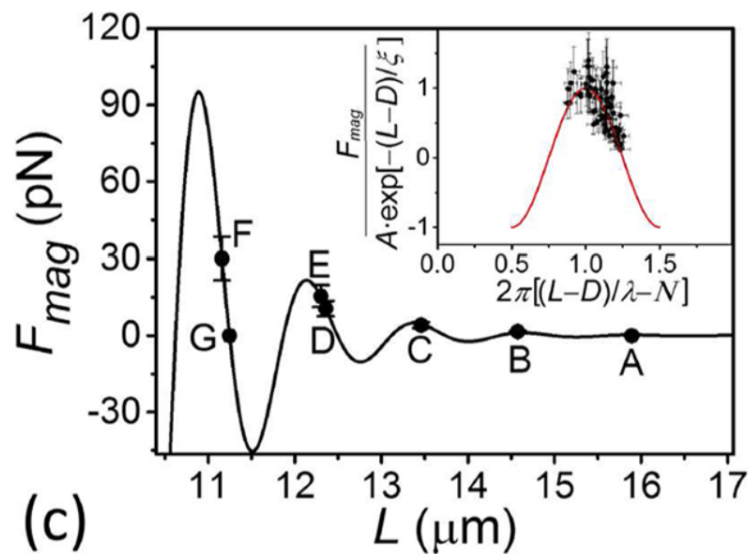
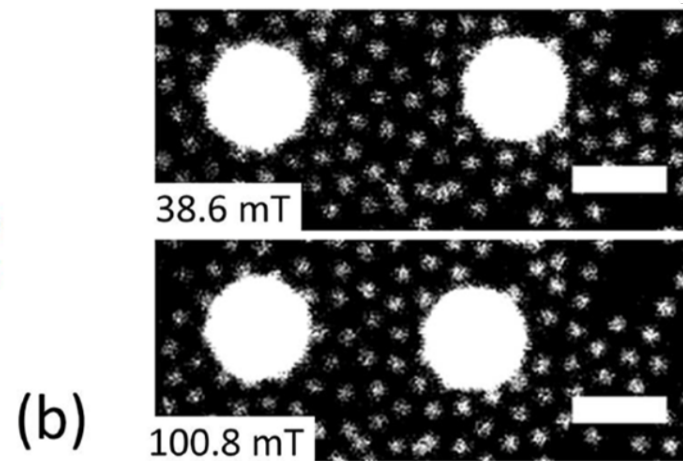
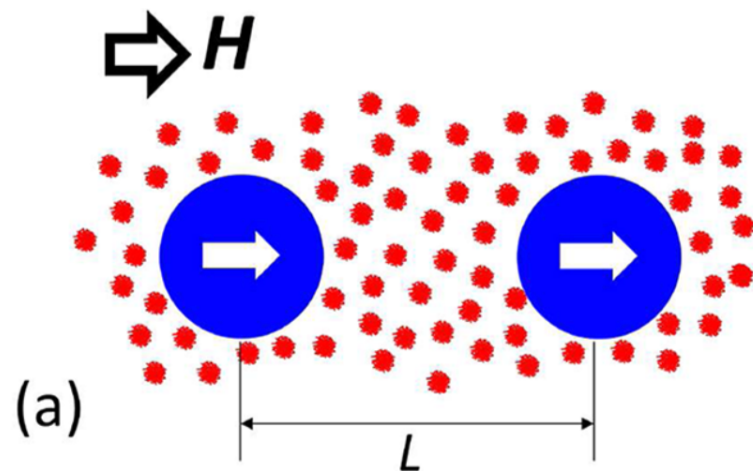
Huang, S., K. Gawlitza, R. von Klitzing, L. Gilson, J. Nowak, S. Odenbach, W. Steffen and G. K. Auernhammer (2016). *Langmuir* **32**: 712-722.

Attractive interaction

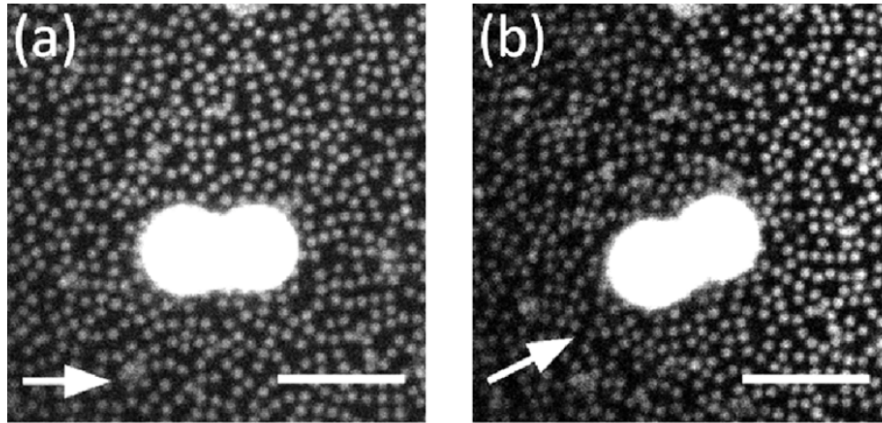


Huang, S., K. Gawlitza, R. von Klitzing, L. Gilson, J. Nowak, S. Odenbach, W. Steffen and G. K. Auernhammer (2016). *Langmuir* **32**: 712-722.

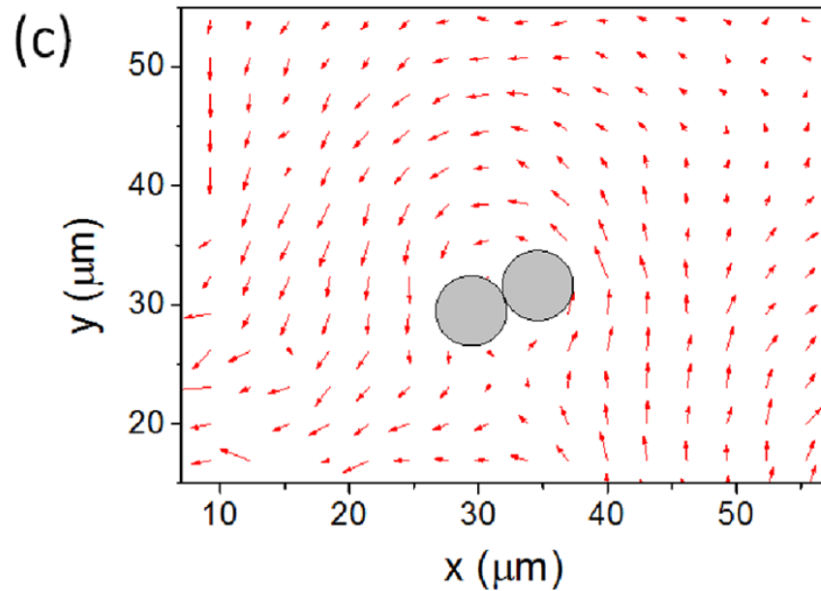
2D compression



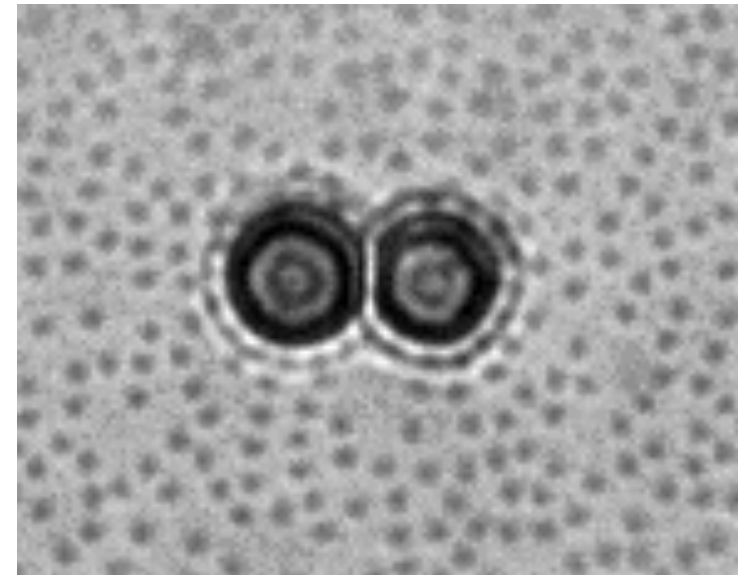
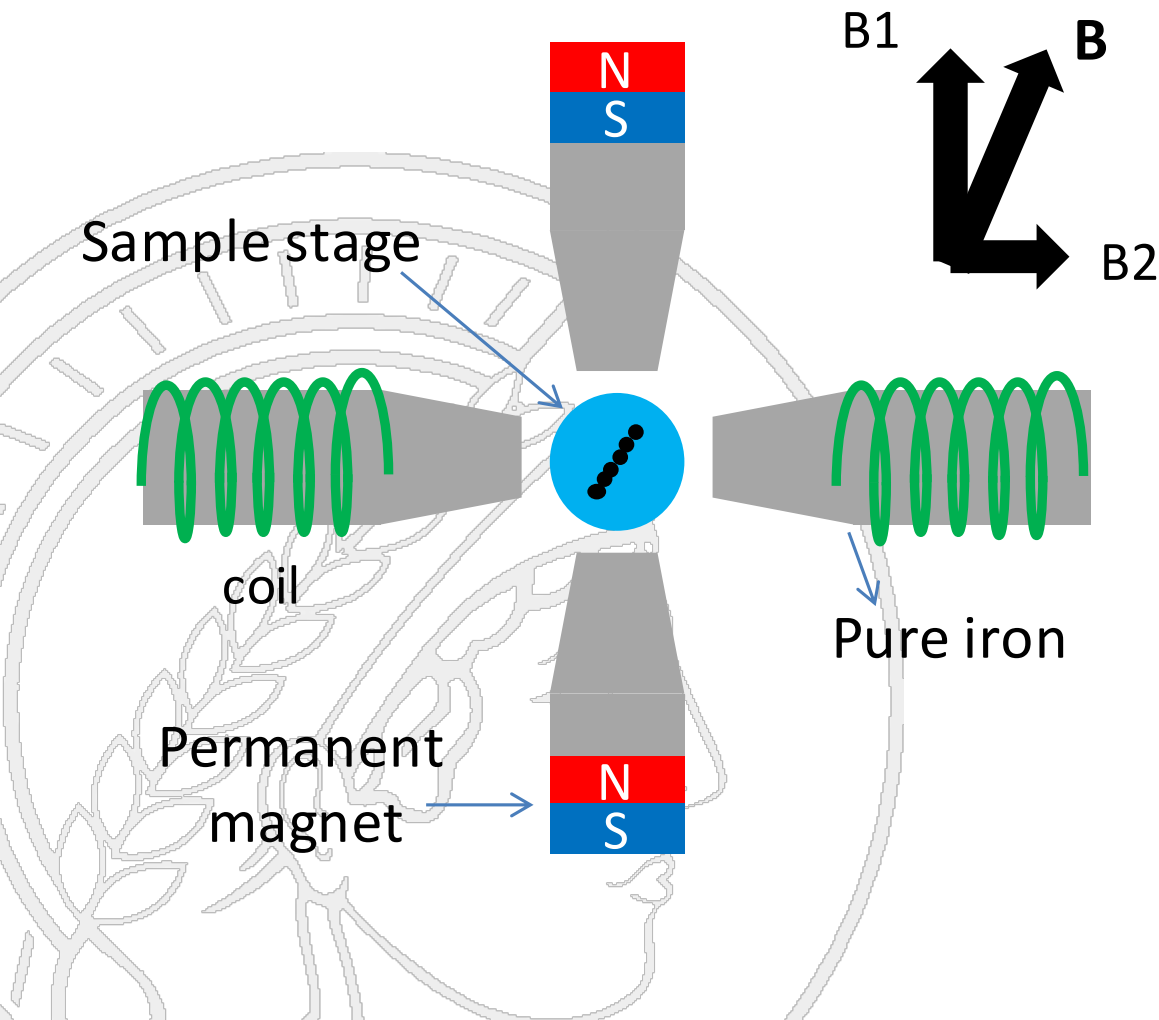
Deformation field in 2D



Step-wise rotation of the magnetic field.

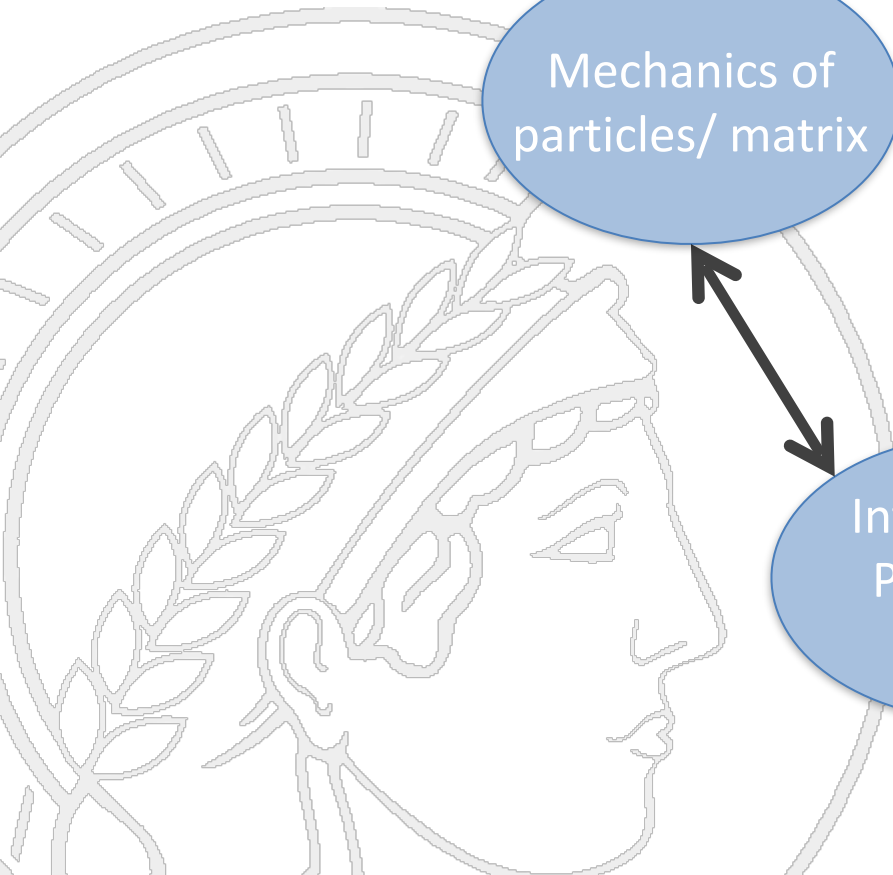
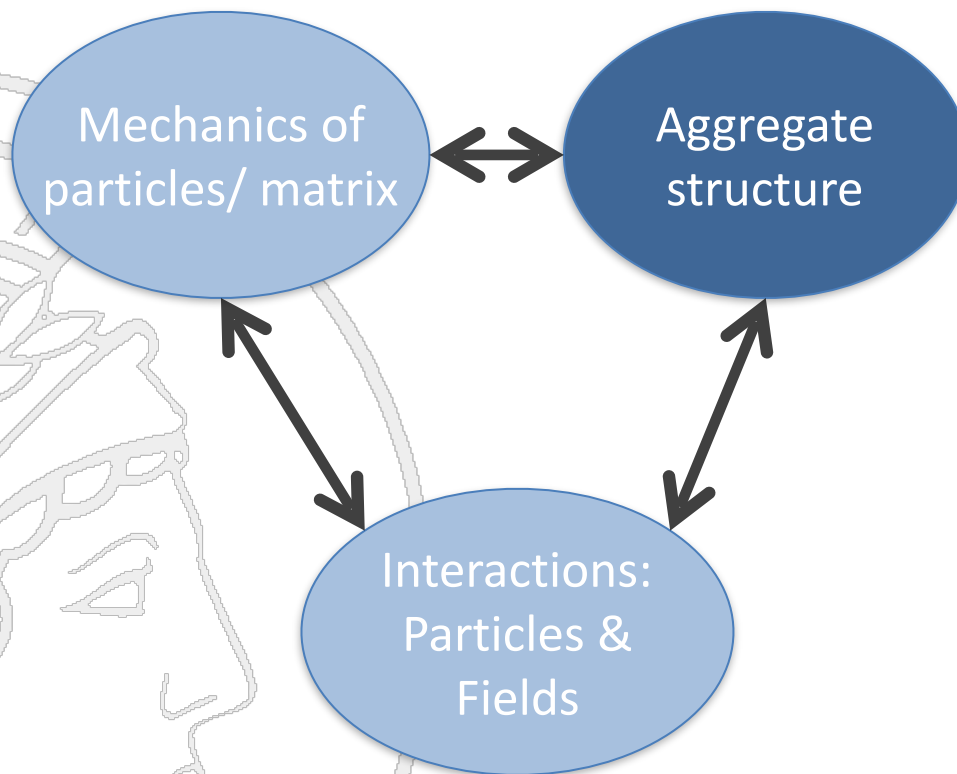


Dynamic measurements (in progress)





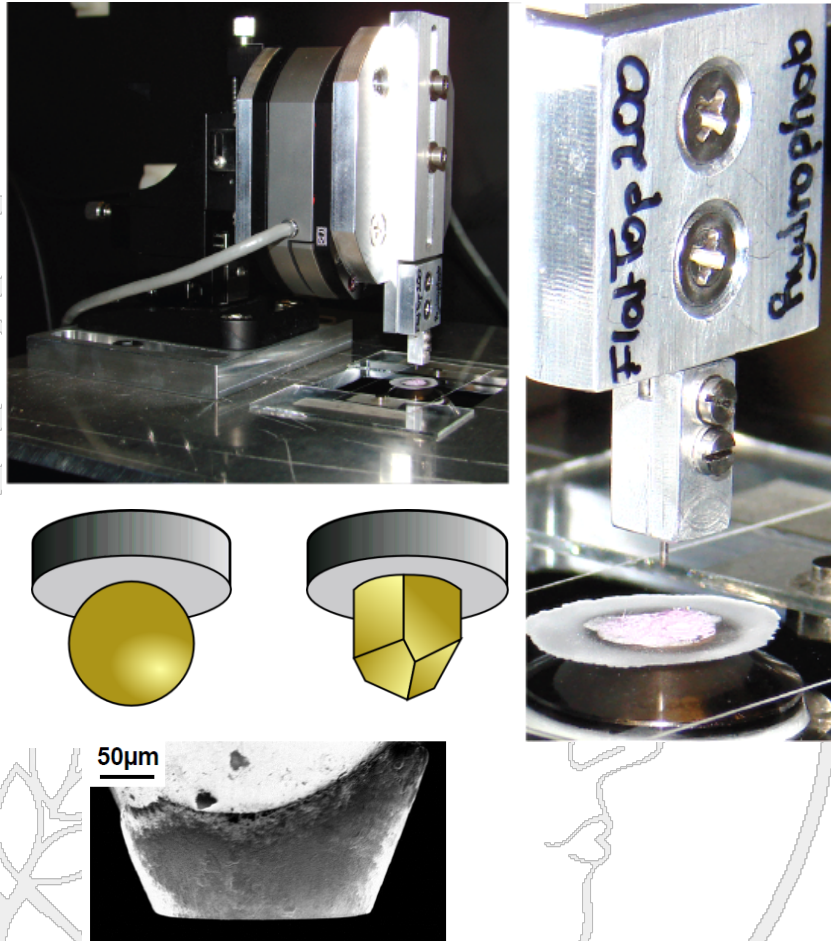
Structure vs. motion



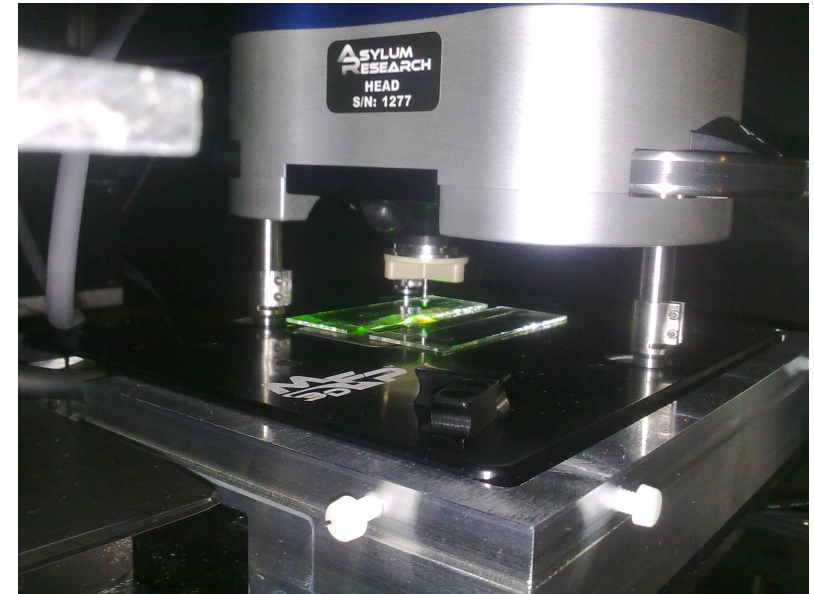
Nano-indentation



Displacement controlled



Force controlled

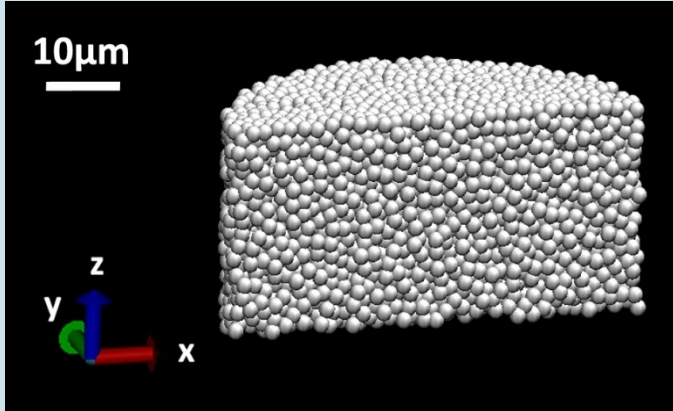
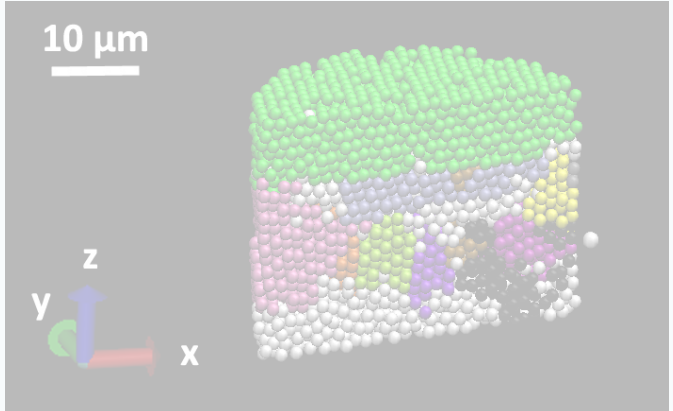


Displacement of the indenter:

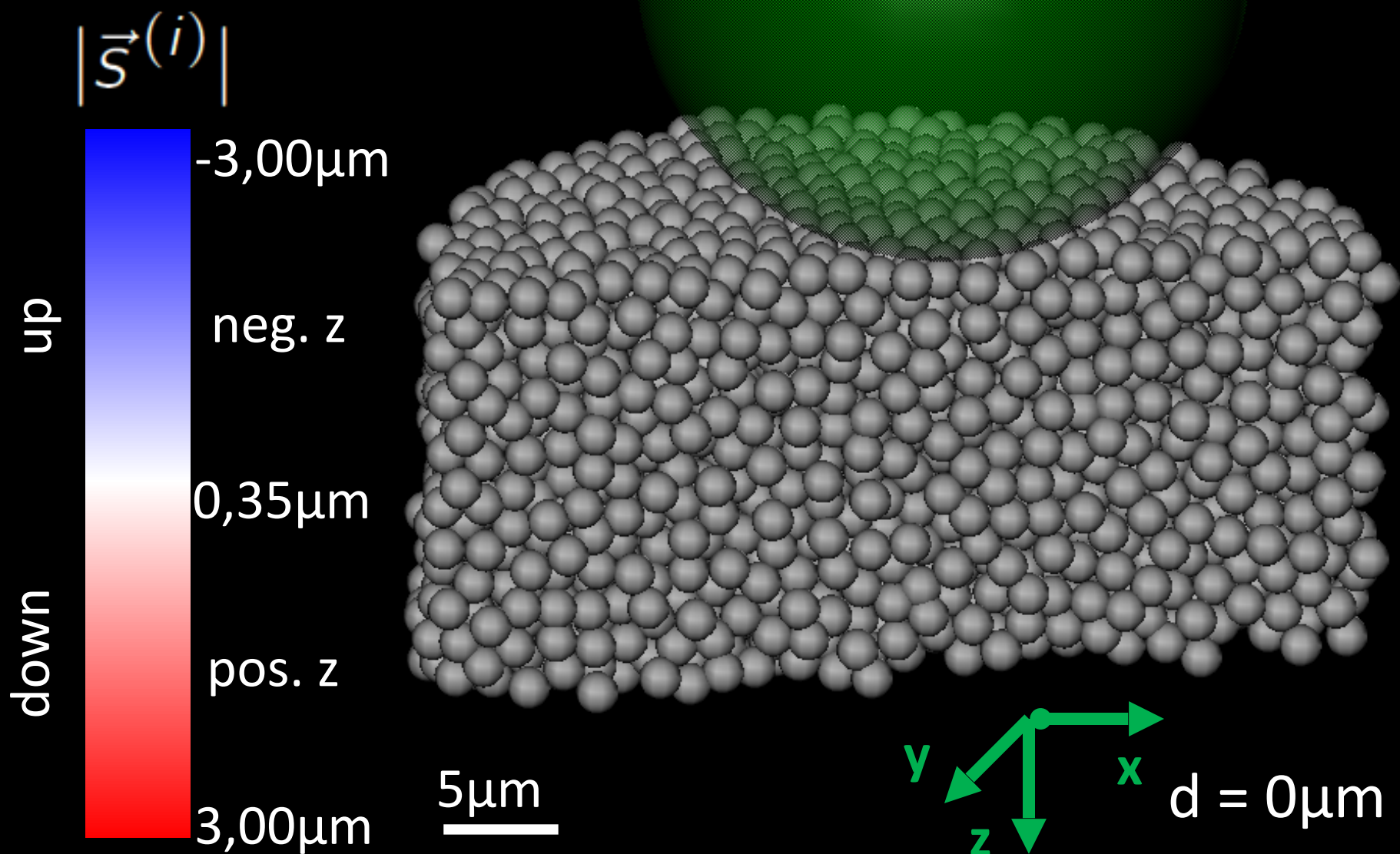
- In x-direction (screch)
- In z-direction (indentation)

Samples

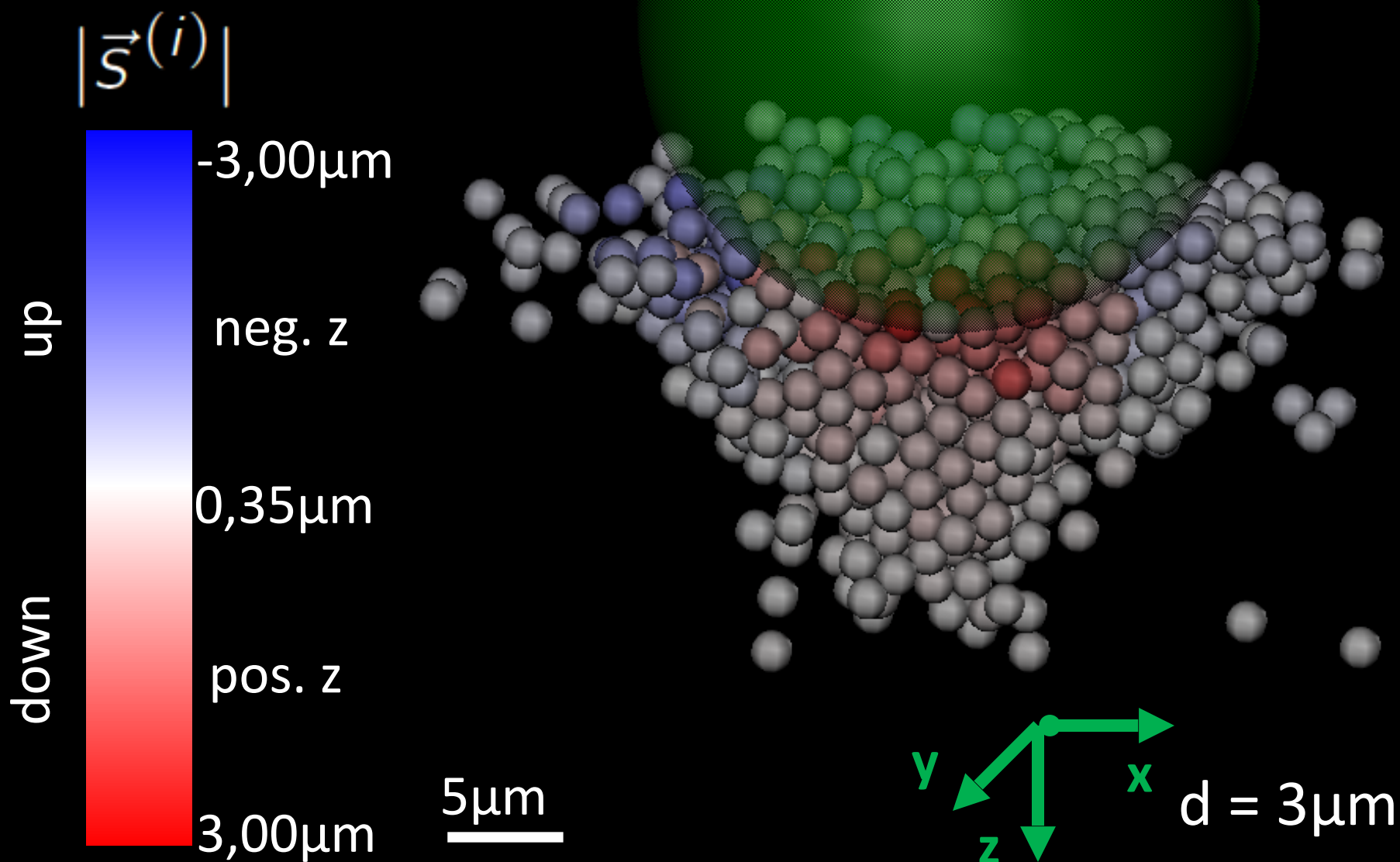


Process	Properties	Reconstruction
<ul style="list-style-type: none"> • PMMA (1.6 μm) • Fast drying from hexane @ 50°C • Slightly cohesive 	<ul style="list-style-type: none"> • Amorphous • Random closed packing (63%) • Film thickness: 30μm 	 <p data-bbox="846 793 1843 829"><i>Auernhammer, Kwade et al. Eur. Phys. J. E (2012) 35, 124;</i></p>
<ul style="list-style-type: none"> • PMMA (1.6 μm) • Slow drying from hexane @ RT • Slightly cohesive 	<ul style="list-style-type: none"> • Crystalline (66%) • Crystalline closed packing (74%) • Film thickness: 30μm 	 <p data-bbox="687 1300 1843 1336"><i>Auernhammer, Kwade et al. Chem. Lett. 41 (10), 1110-1112 (2012).</i></p>

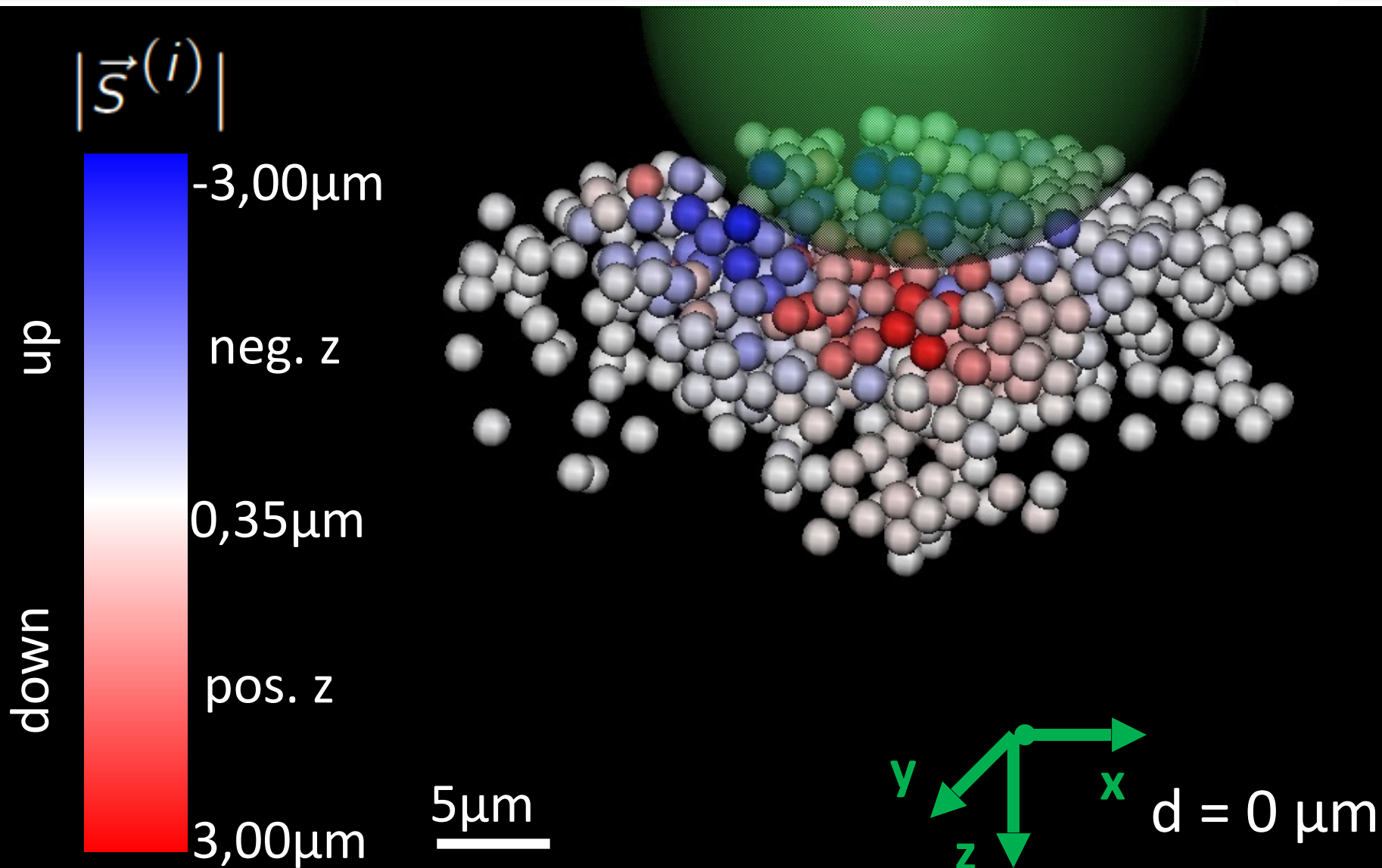
Imaging indentation



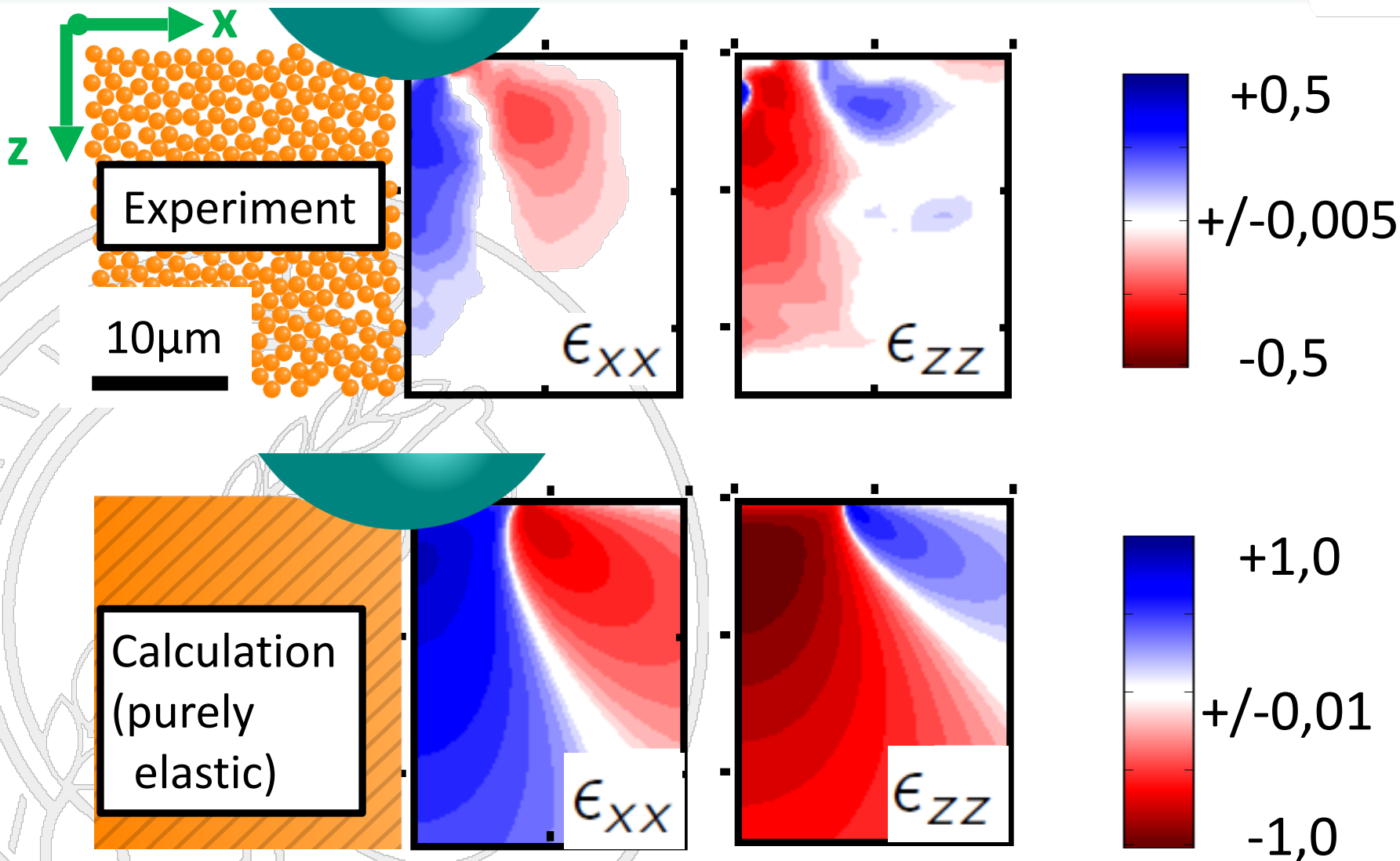
Imaging indentation



Imaging indentation



Comparison to continuum mechanics



Comparison to continuum mechanics

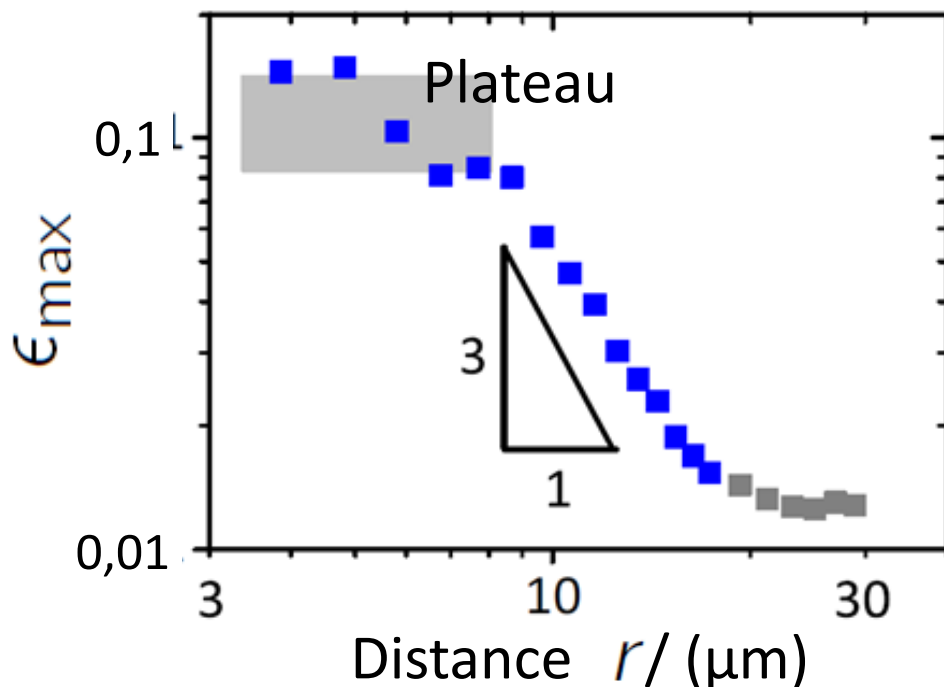
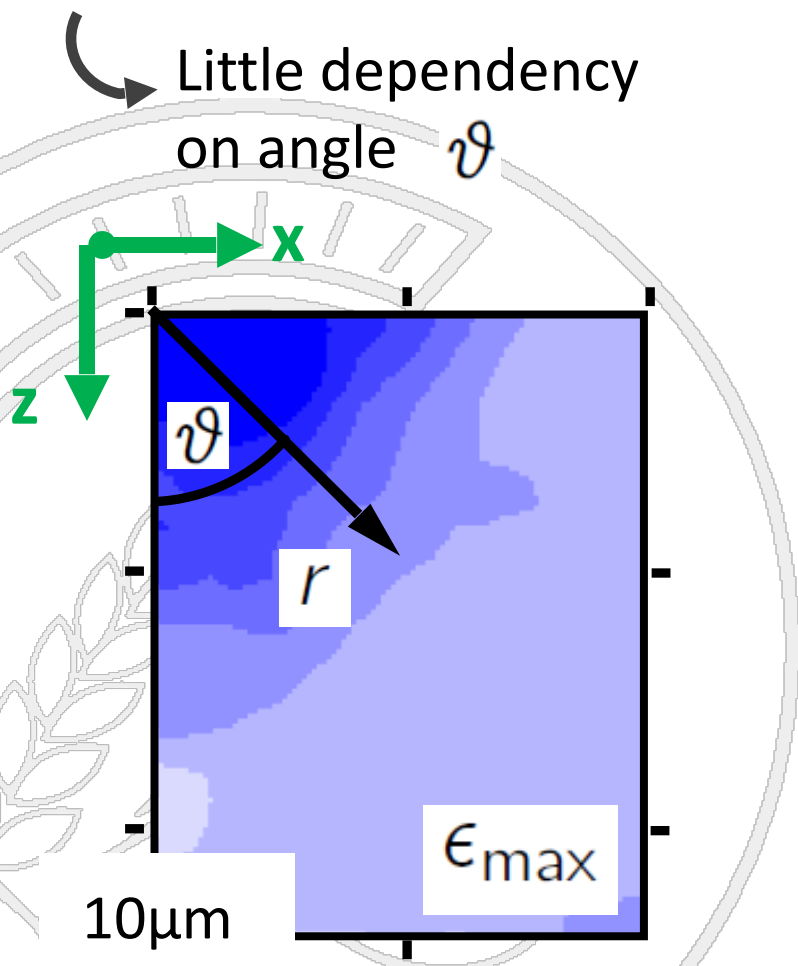


Plastic vs. elastic deformation

$$\epsilon_{max} = \sqrt{(\epsilon_{xx} - \epsilon_{zz})^2 / 4 + \epsilon_{xz}^2}$$

Little dependency on angle ϑ

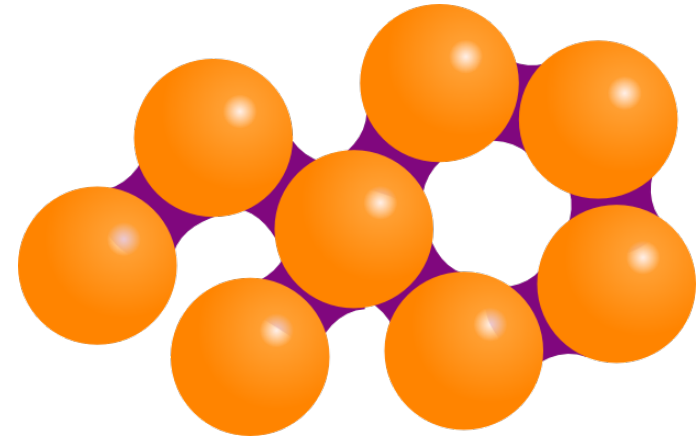
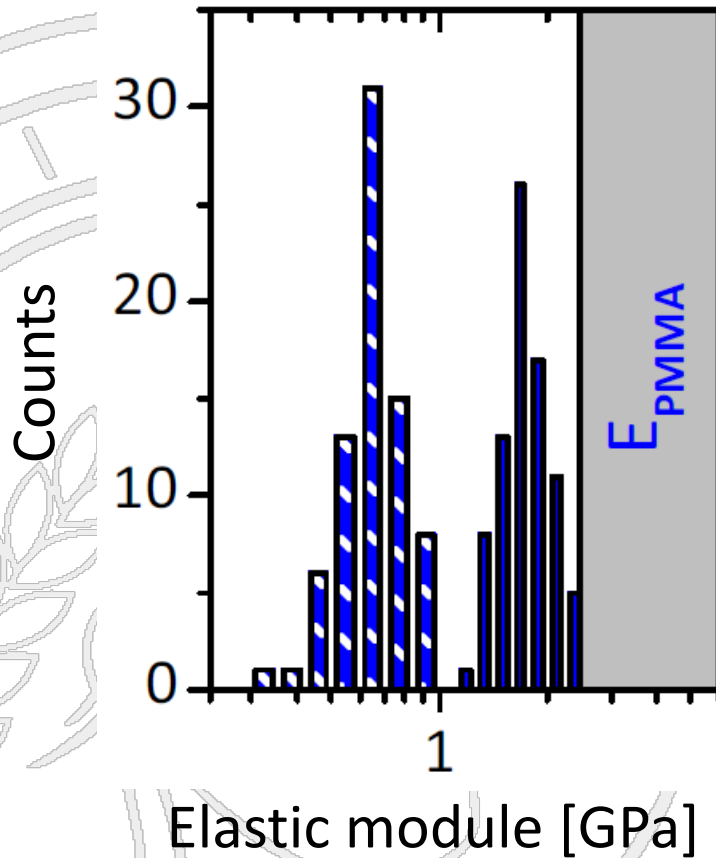
Elastic deformation for „large“ distances to indenter



Effect of added polymer

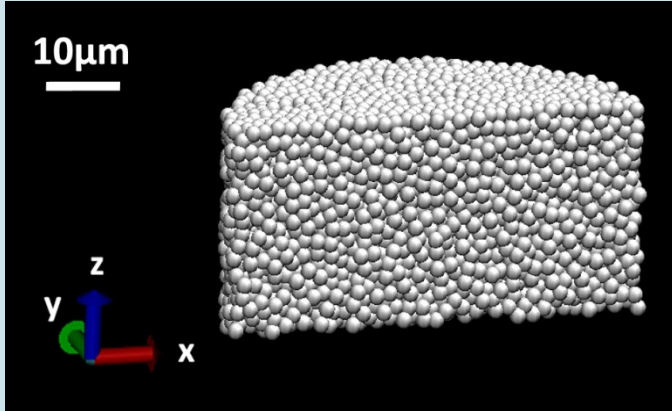
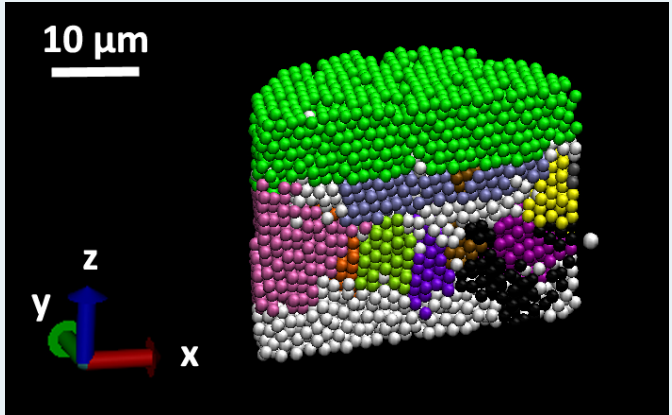


Introducing solid bridges between polymer colloids

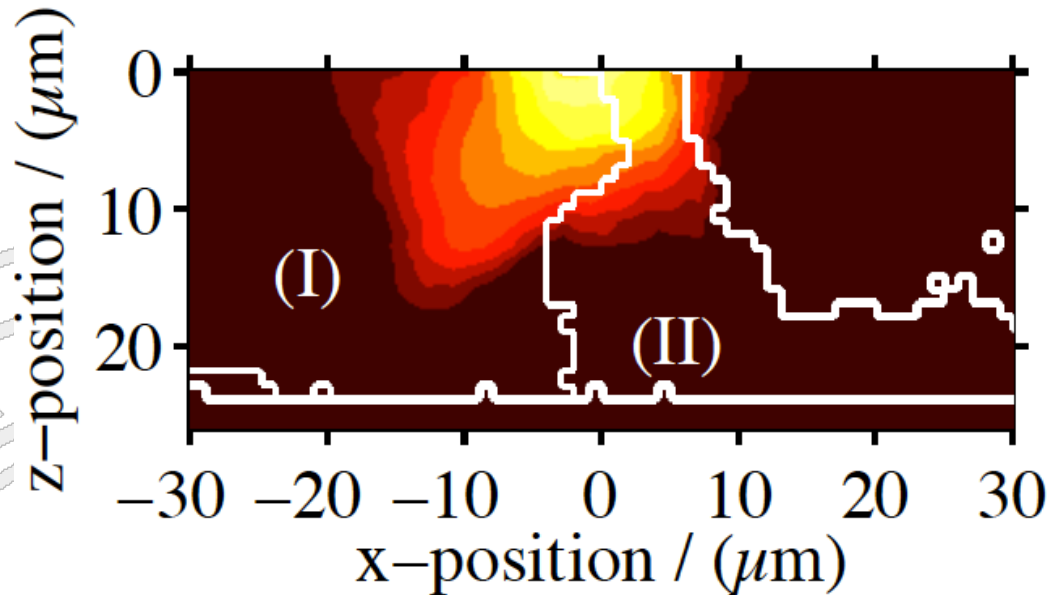


Samples



Process	Properties	Reconstruction
<ul style="list-style-type: none"> • PMMA (1.6 μm) • • Fast drying from hexane @ 50°C • • Dominant effect: Hydrodynamics 	<ul style="list-style-type: none"> • Amorphous • • Random closed packing (63%) • • Film thickness: 30μm 	 <p style="text-align: right;"><i>Auernhammer, Kwade et al. Eur. Phys. J. E (2012) 35, 124;</i></p>
<ul style="list-style-type: none"> • PMMA (1.6 μm) • • Slow drying from hexane @ RT • • Dominant effect: Sedimentation 	<ul style="list-style-type: none"> • Crystalline (66%) • • Crystalline closed packing (74%) • • Film thickness: 30μm 	 <p style="text-align: right;"><i>Auernhammer, Kwade et al. Chem. Lett. 41 (10), 1110-1112 (2012).</i></p>

Effect of crystallinity



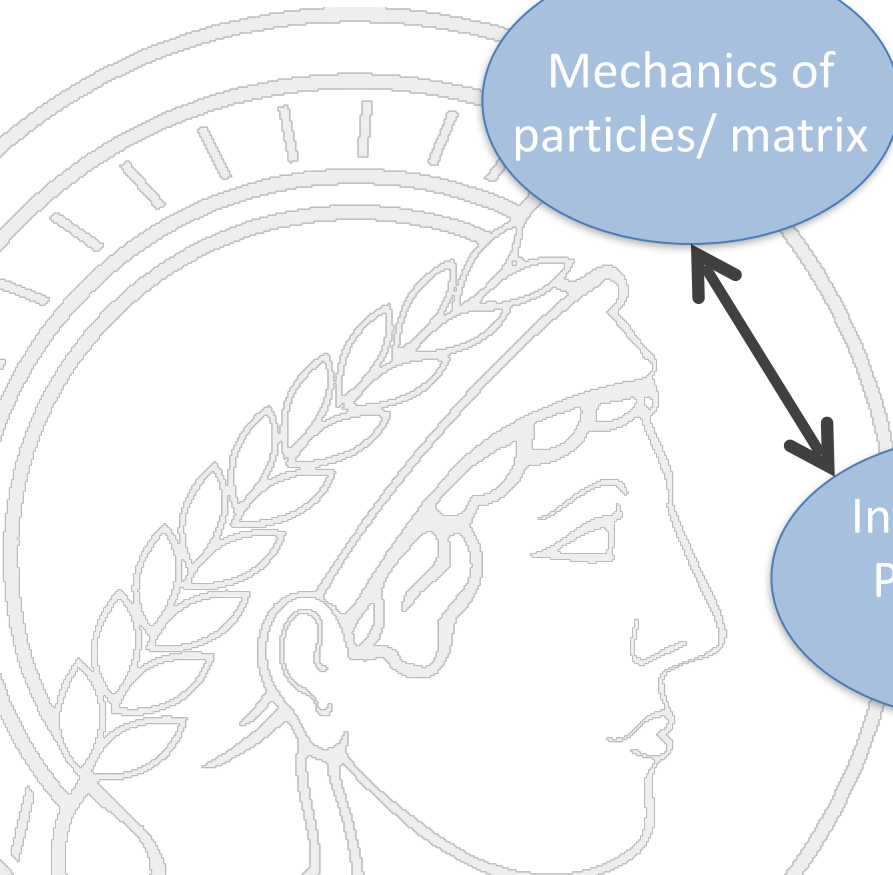
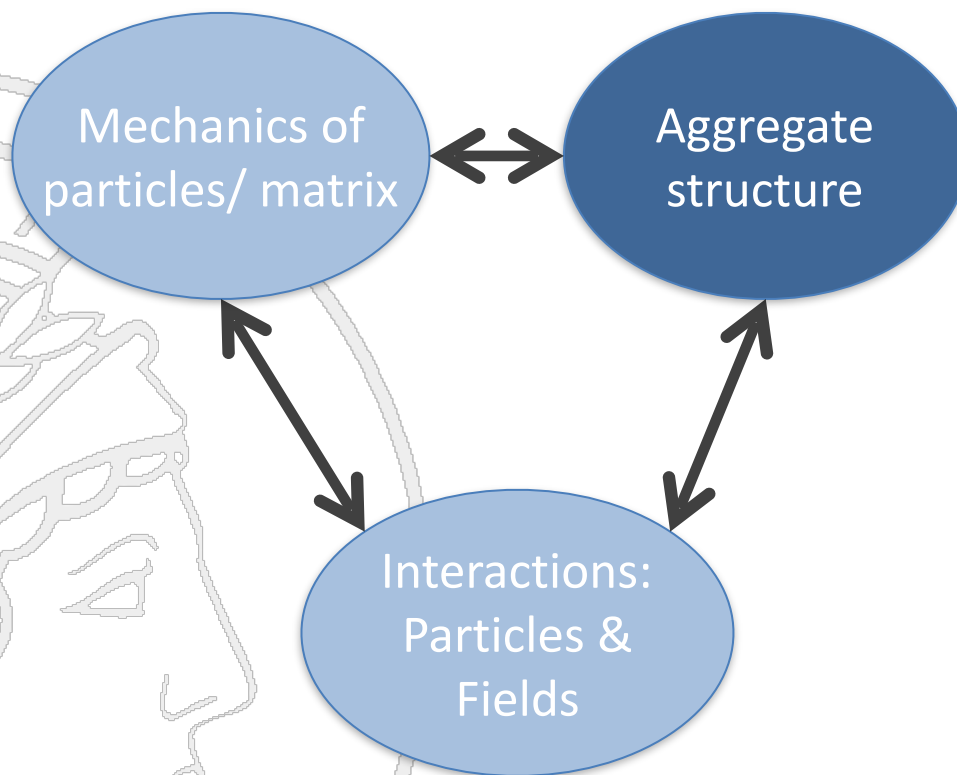
- Deformation longer ranged in crystalline region (I)
- Stronger reorganization (plastic deformation) in amorphous region (II)
- Largest microscopic structure is crystallite size

➔ Particle systems as model for amorphous/semi-crystalline solids

➔ Additives may dominate behavior



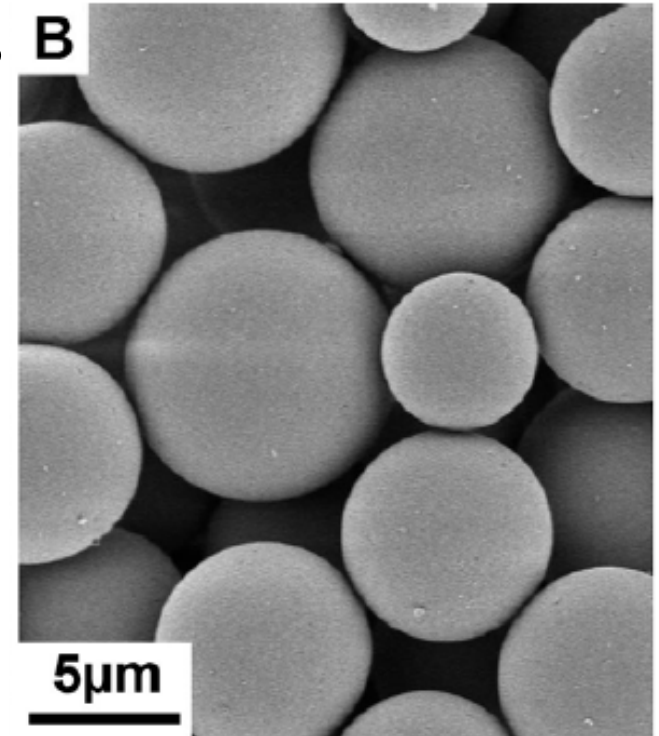
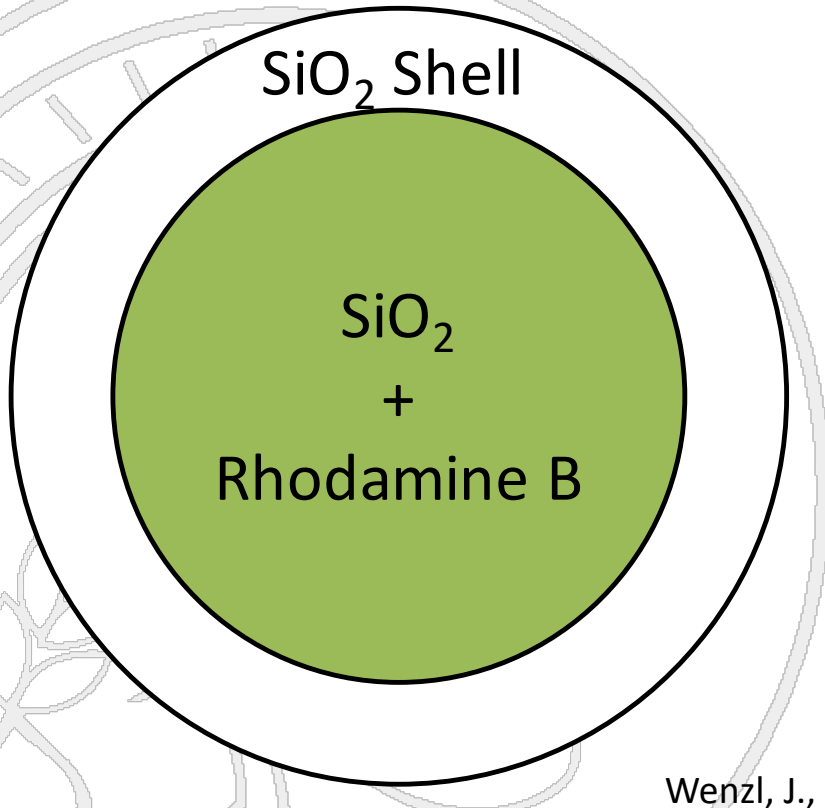
Shearing model sand



Model granular particles

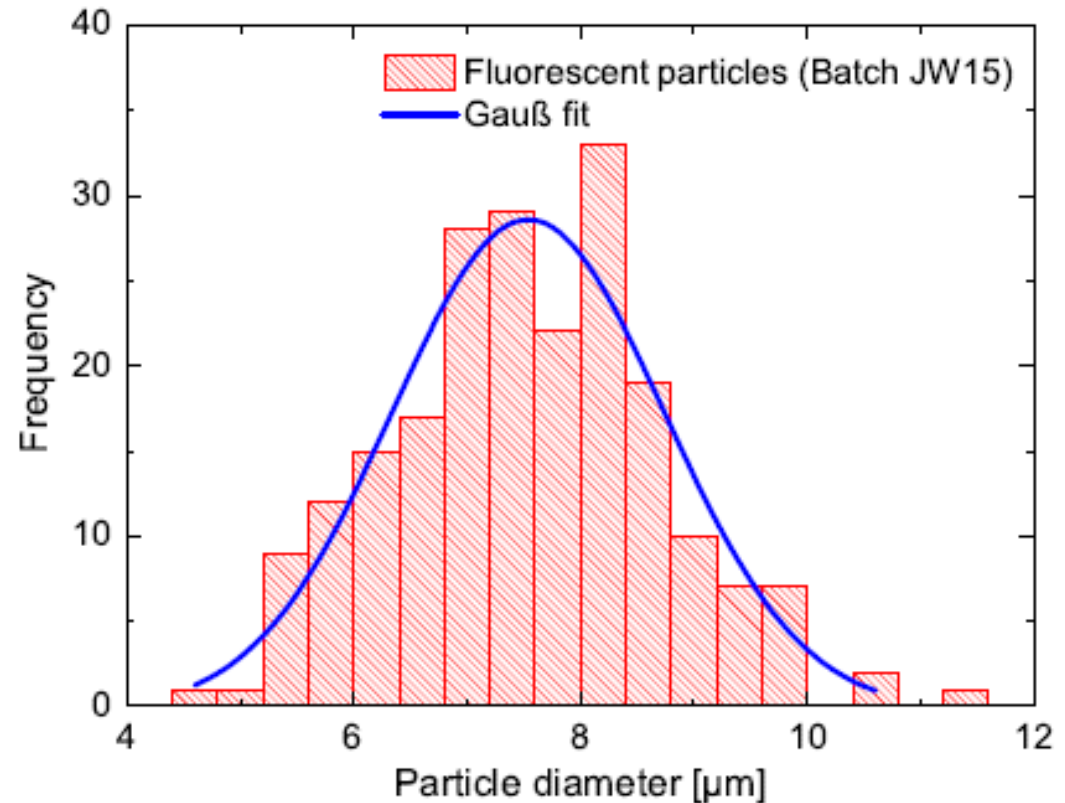
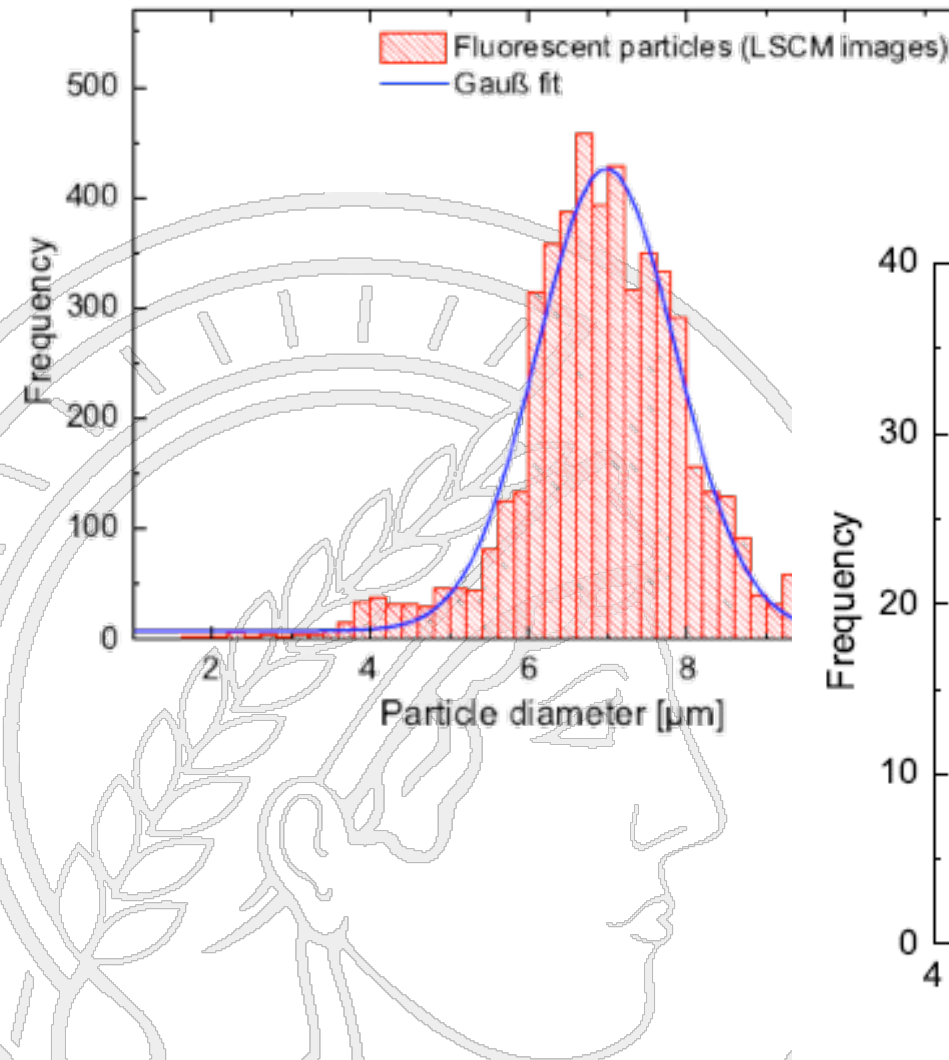


- 7 μm spherical silica particles
- Core-shell particles

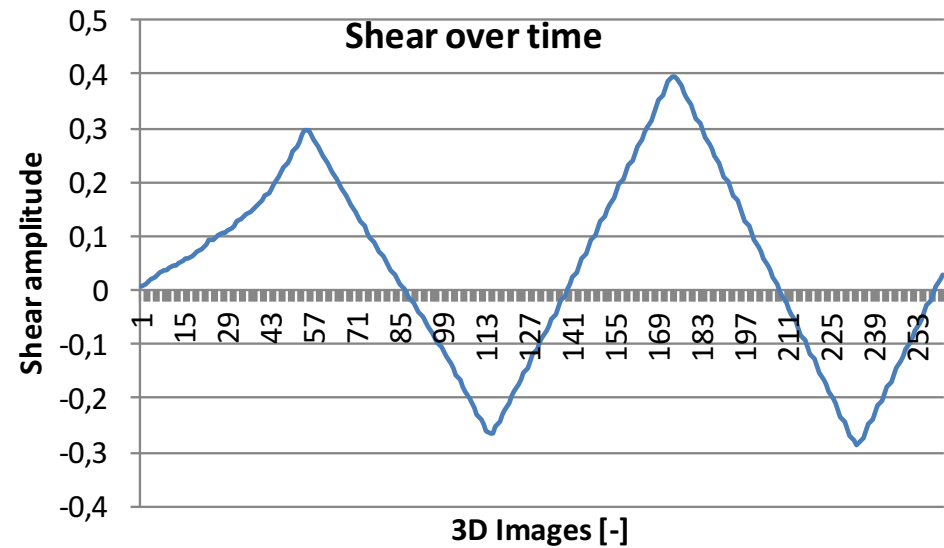
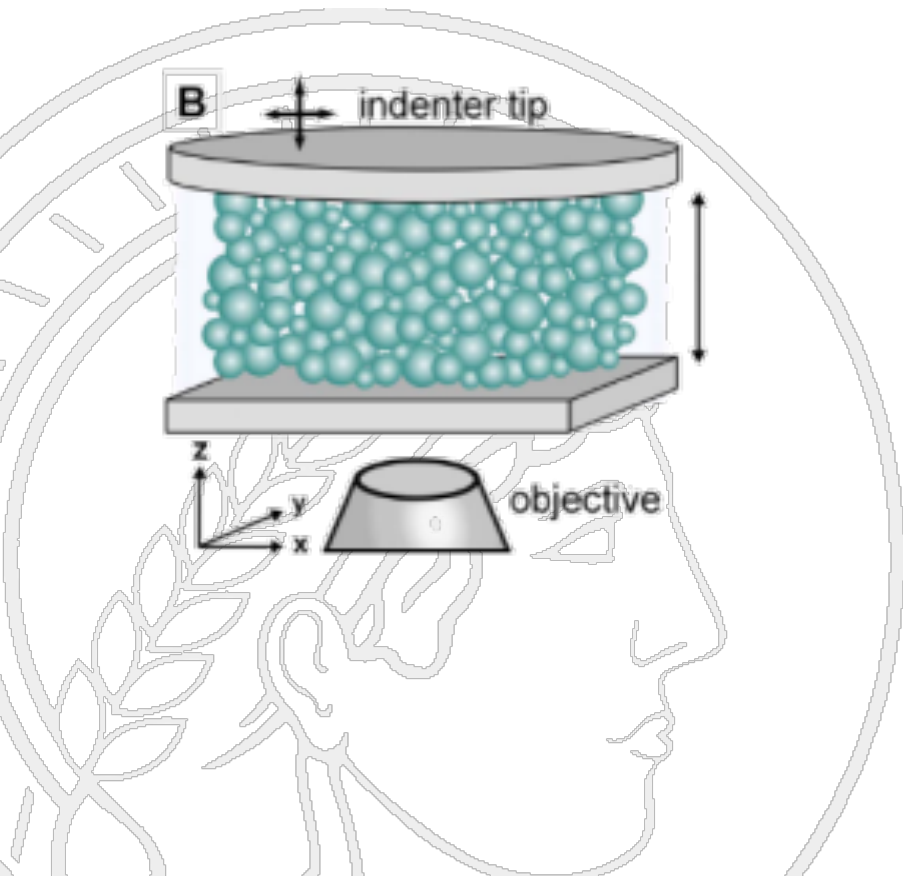


Wenzl, J., R. Seto, M. Roth, H.-J. Butt and G. Auernhammer (2013). *Granular Matter* **15**: 391-400.

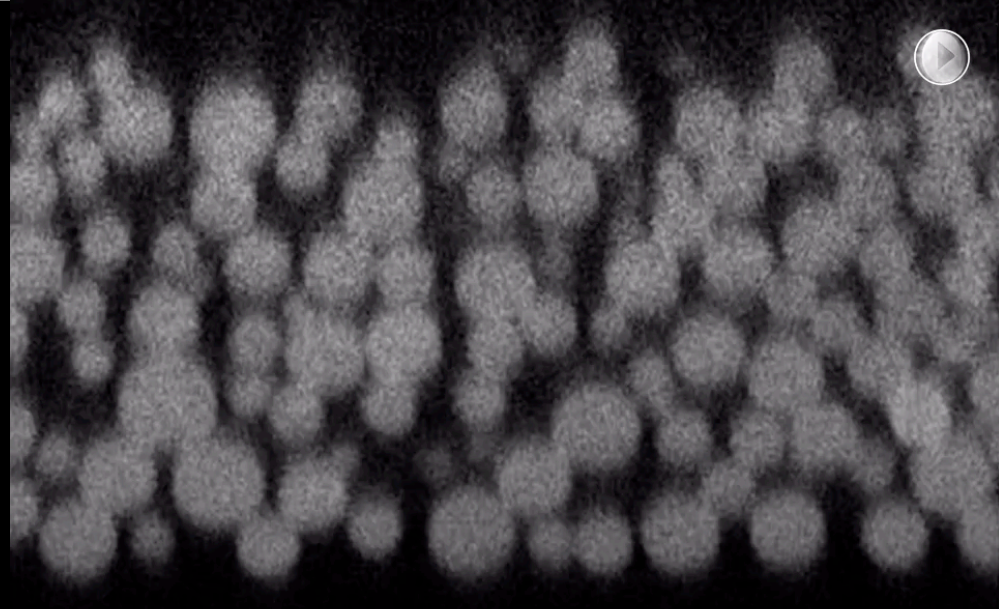
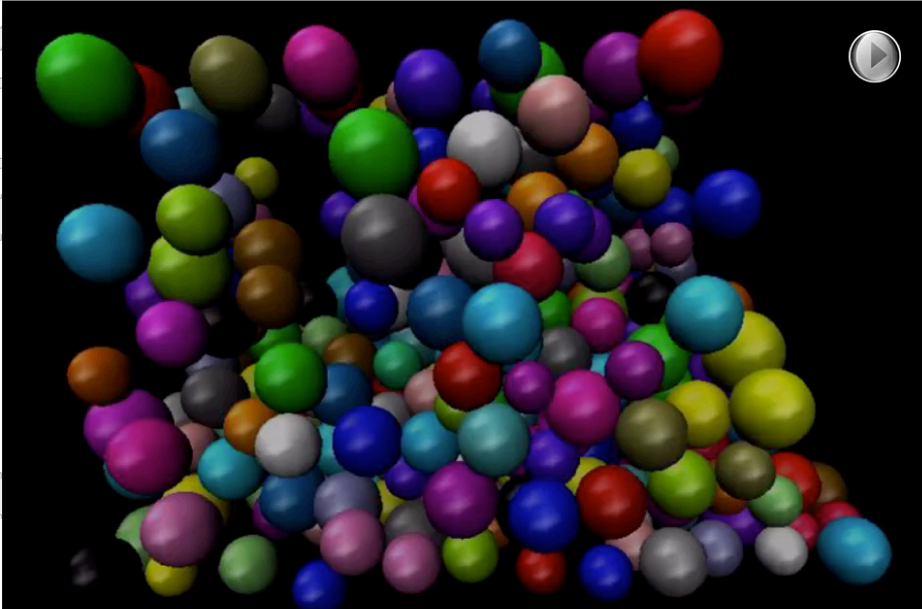
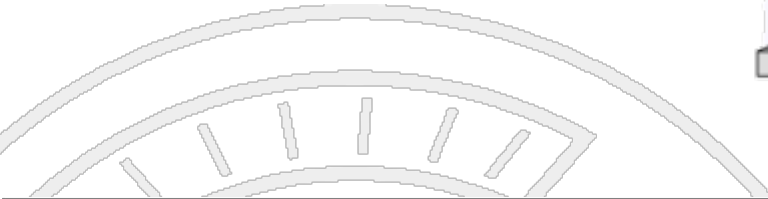
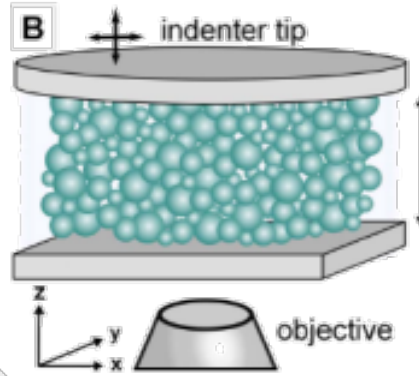
Polydisperse silica spheres



Typical shear experiment



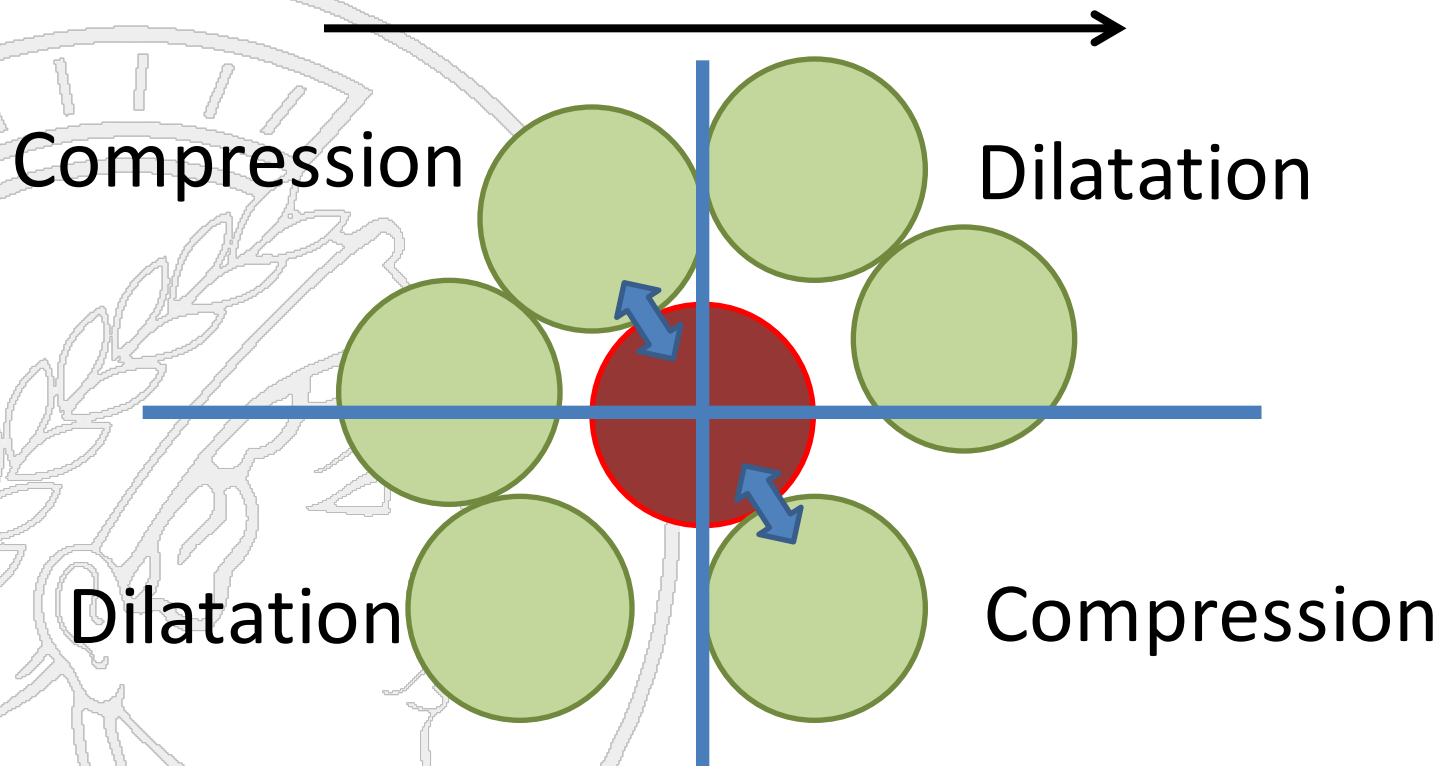
Trajectories of all particles



Contact network



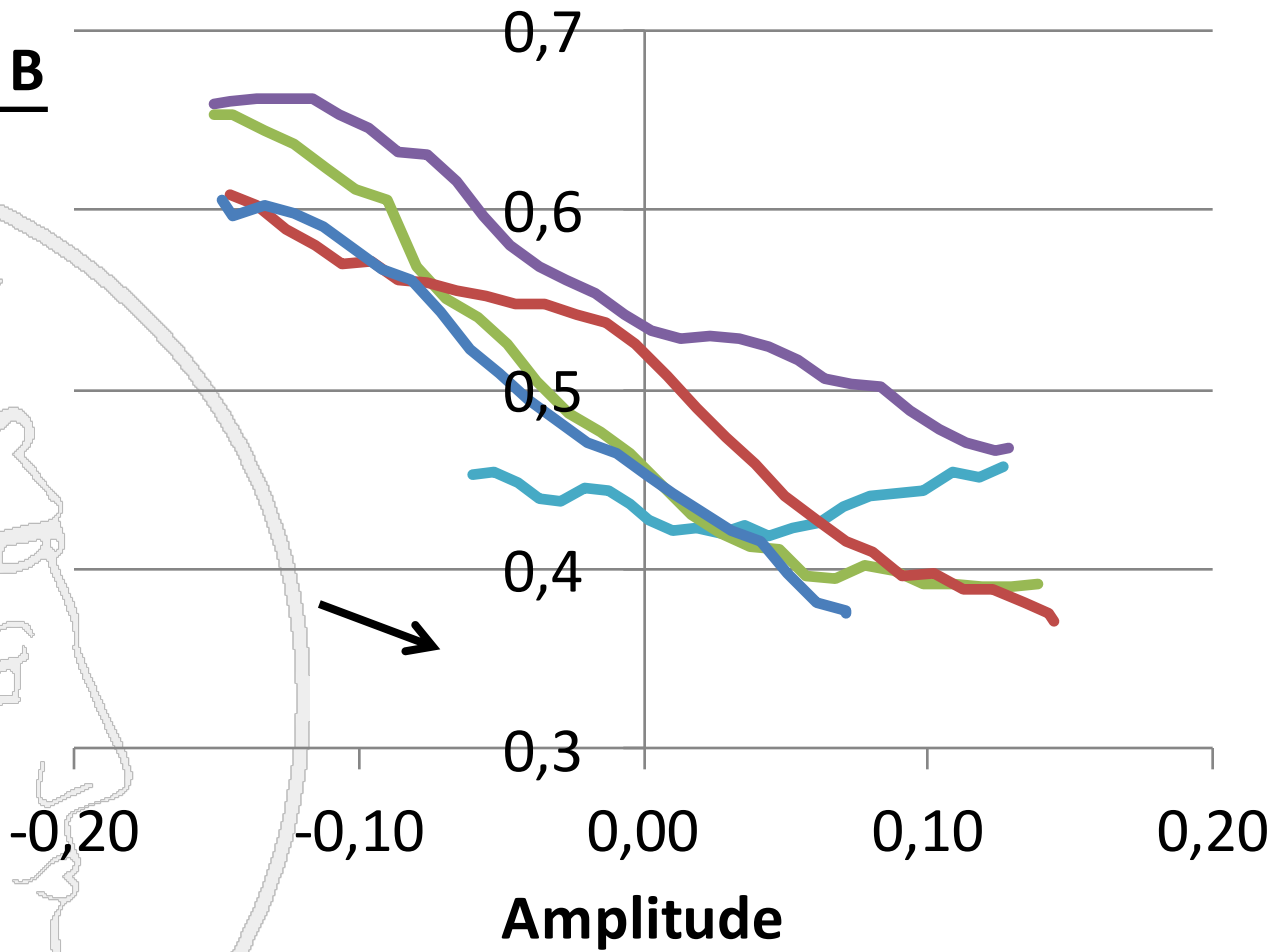
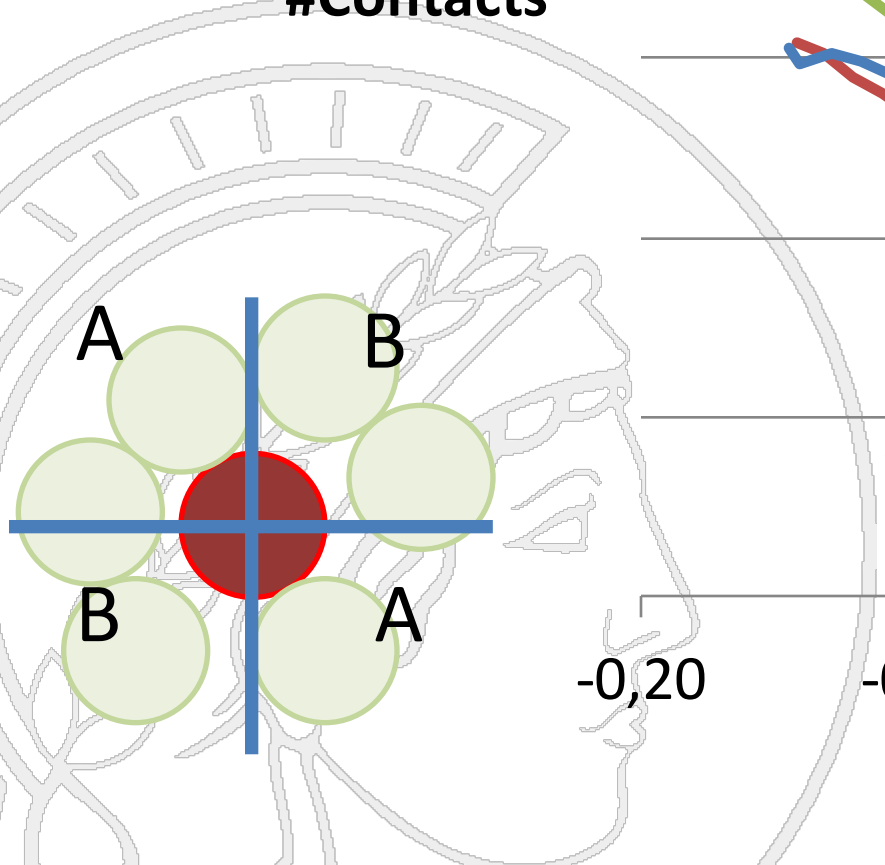
- Anisotropic contact network
- Force chain network



Contact network (small amplitude)



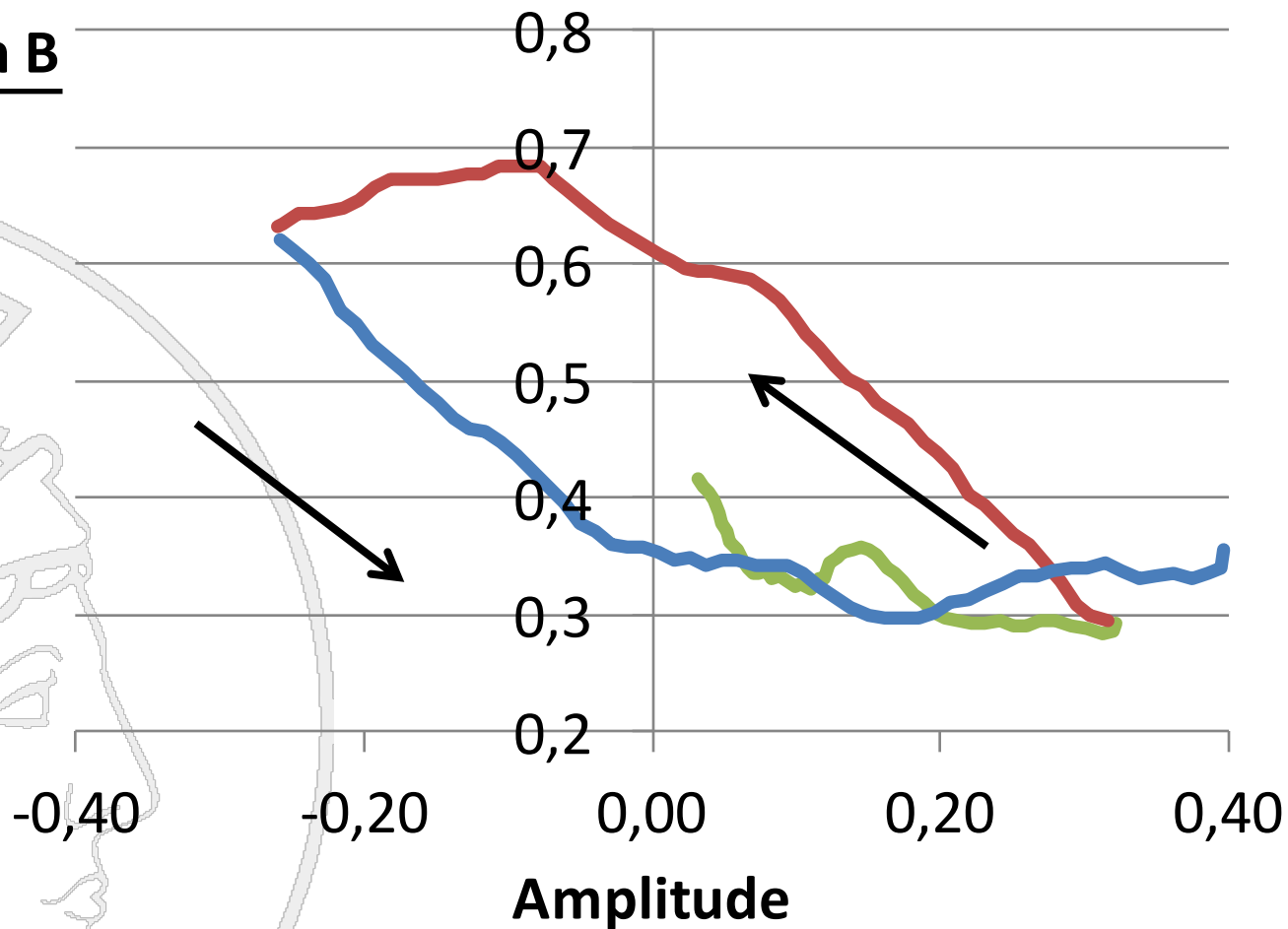
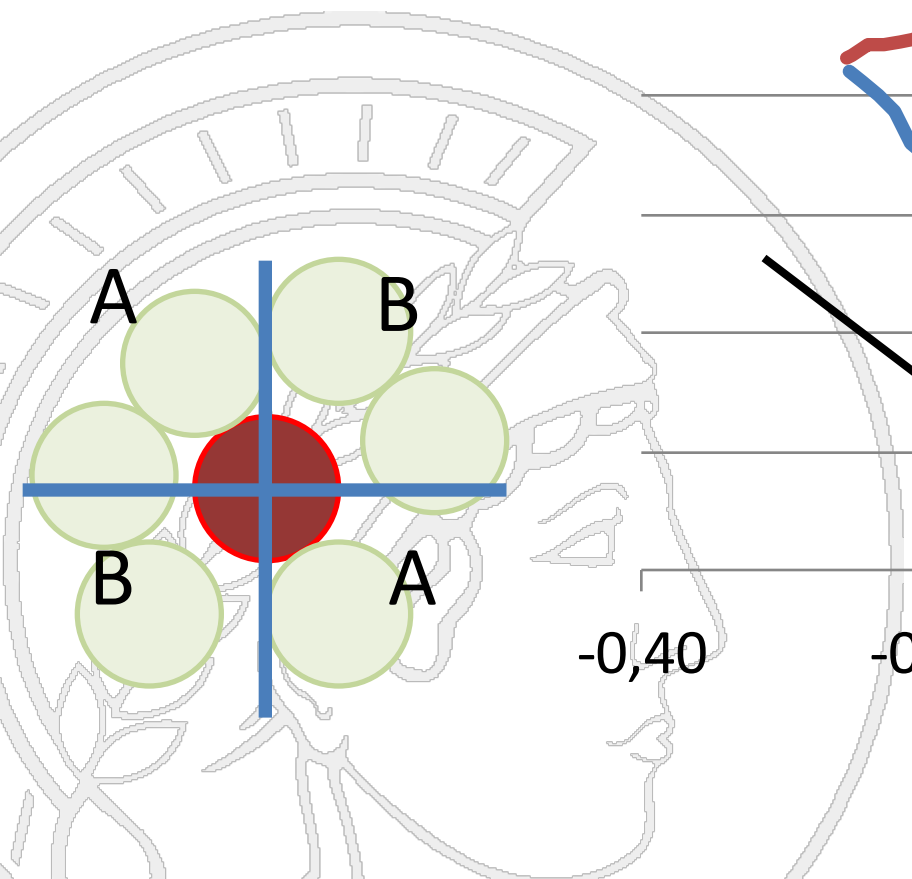
#Direction B
#Contacts



Contact network (large amplitude)

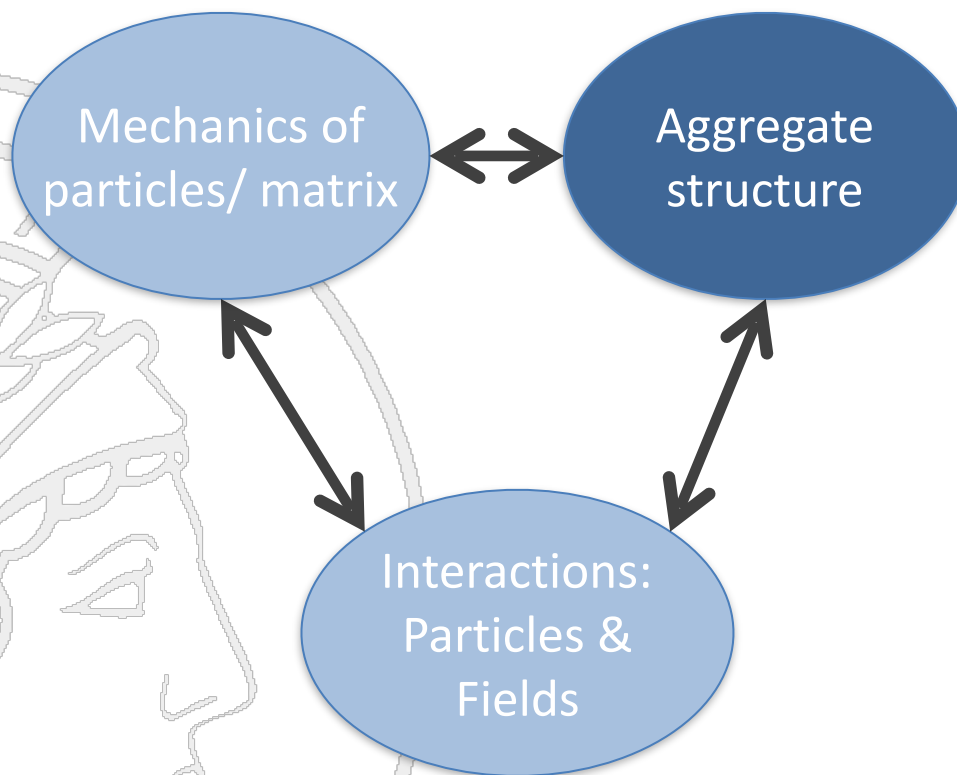


#Direction B
#Contacts





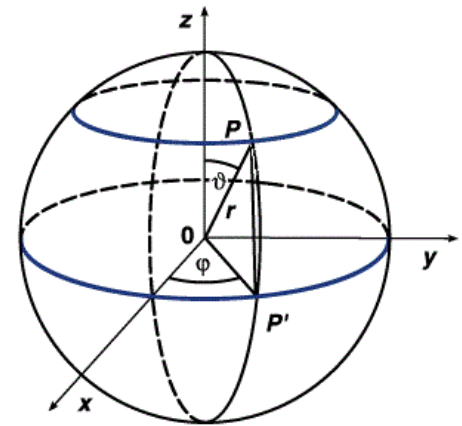
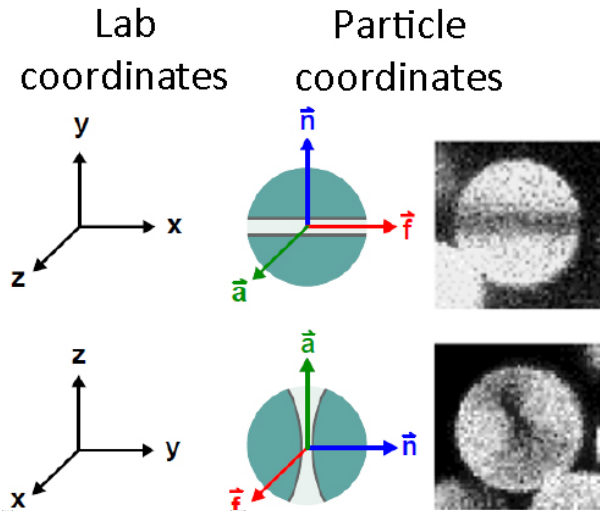
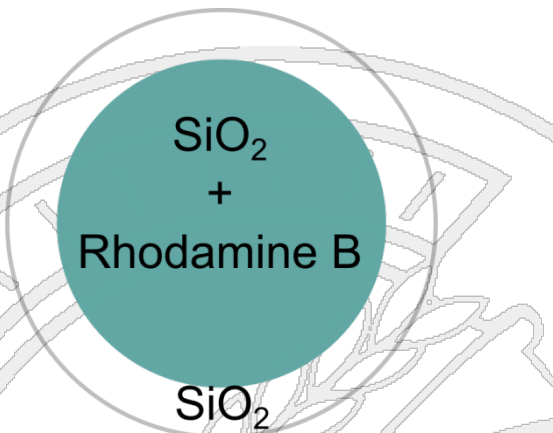
Orientation of spheres



Particle rotation



Optically anisotropic particles: Photobleaching of internal structure

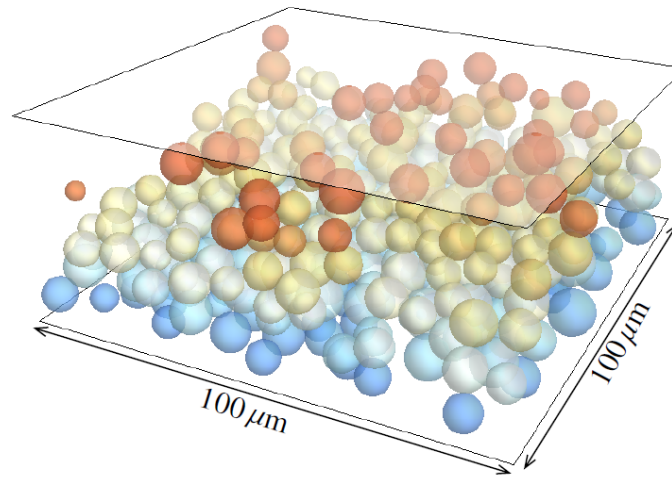
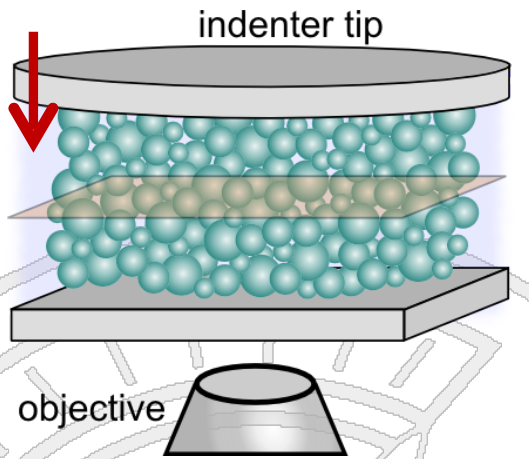


- Core-shell particles: Isotropic interaction
- Bleaching generates plane in particle
- Particle coordinate system
- Rotates in lab system

Set of 3 independant angles, describes particle orientation in 3D

Wenzl, et al., Granul Matt (2012) DOI 10.1007/s10035-012-0383-7

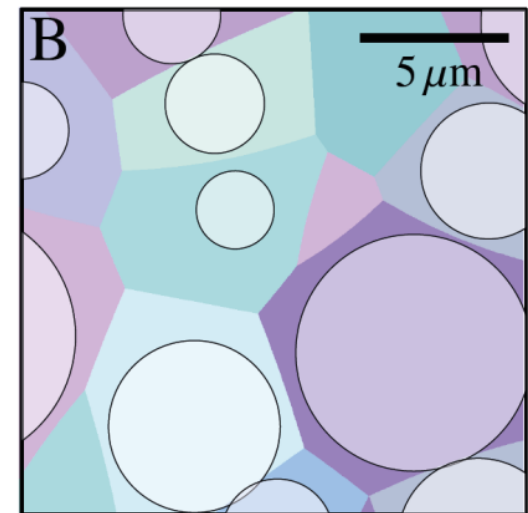
Compression measurement



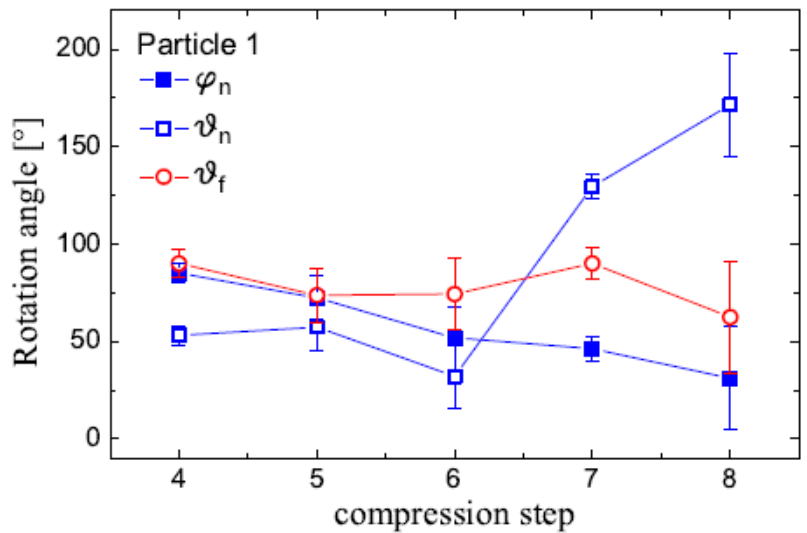
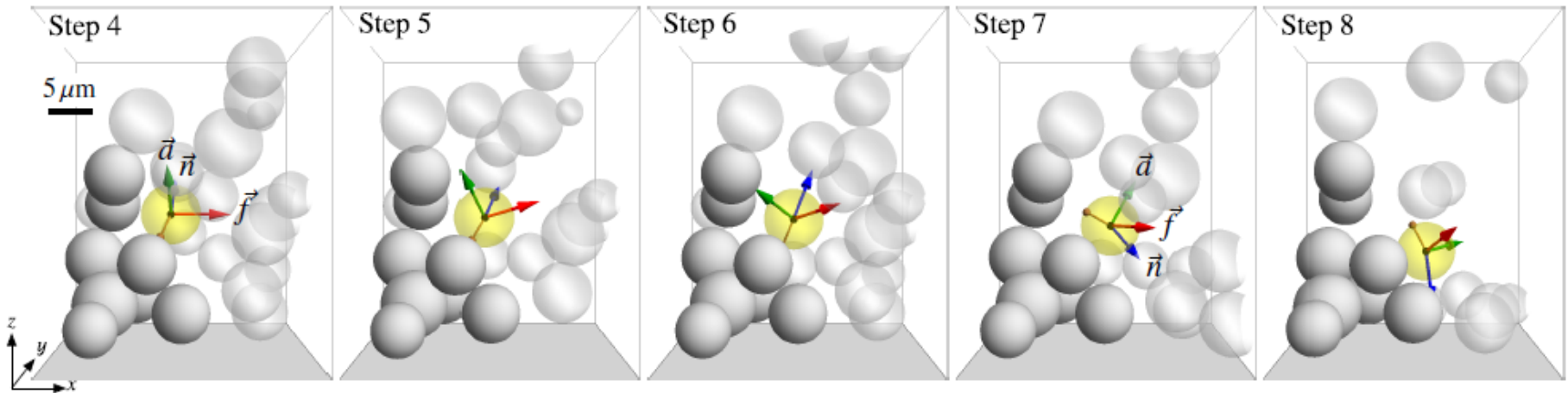
- Starting configuration:
- sedimented bed
 - Volume fraction $\phi \sim 45\%$
 - 8 compression steps

For each compression step:

- Position (coordinates) of particles
- Voronoi cell volume
- Trajectories of anisotropic particles



3D reconstruction



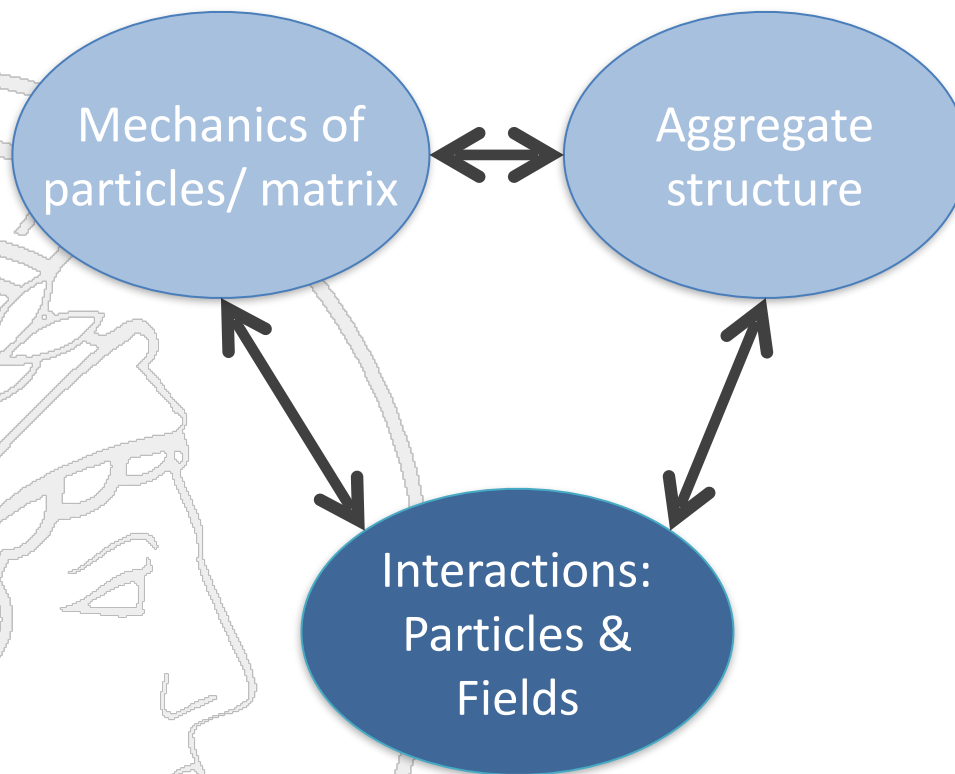
- Step 4 - 6: Only rotation
- Step 6/7: Bond breaking
- Step 7 & 8: Additional sliding motion

Full description needs translation and rotation

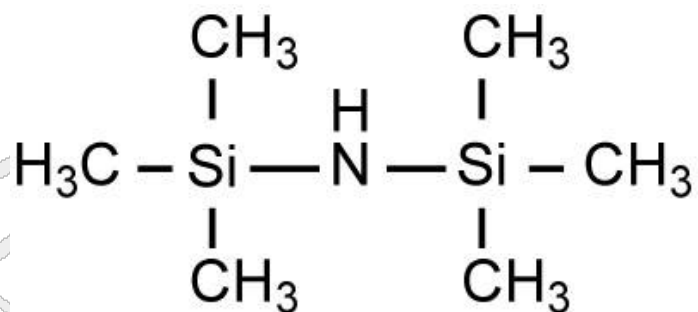
Now: Automatic orientation detection



Attractive colloids

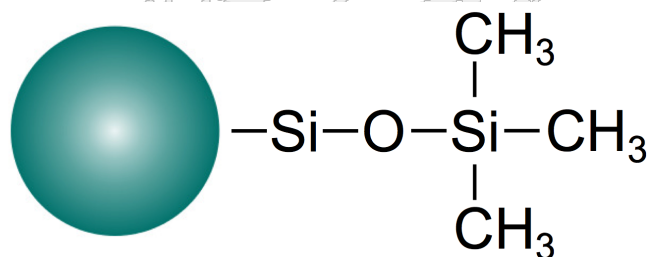


Hydrophobization of silica



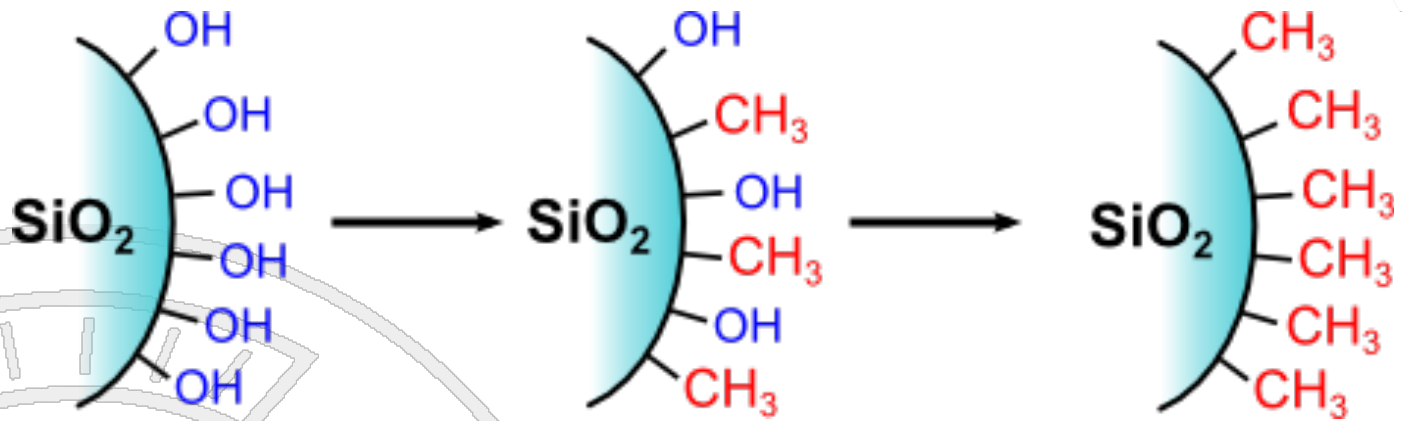
HDMS
Hexamethyldisilazane

- Slow binding



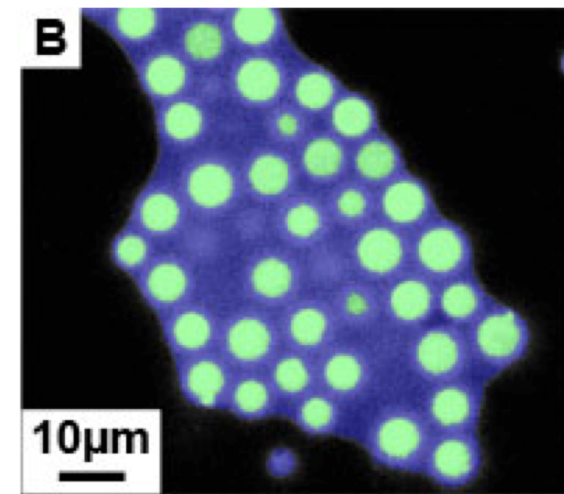
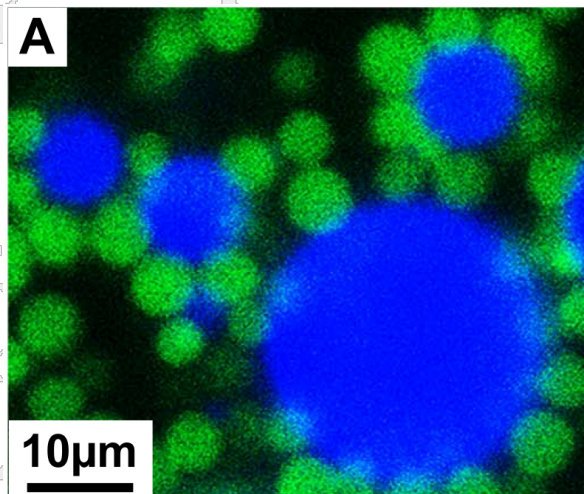
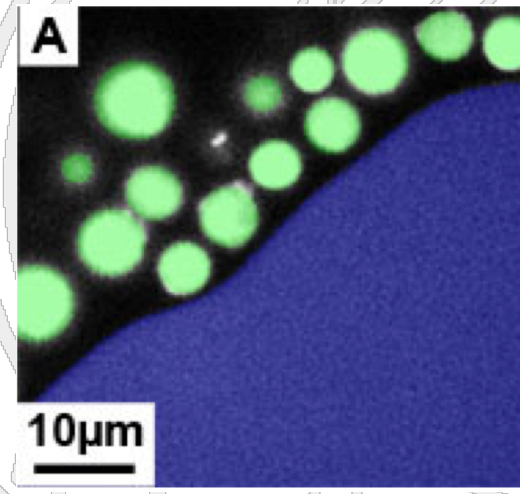
Hair, M. L. and W. Hertl (1971). *J. Phys. Chem.* **75**(14): 2181-2185.

Wet model granulates



Phase separation

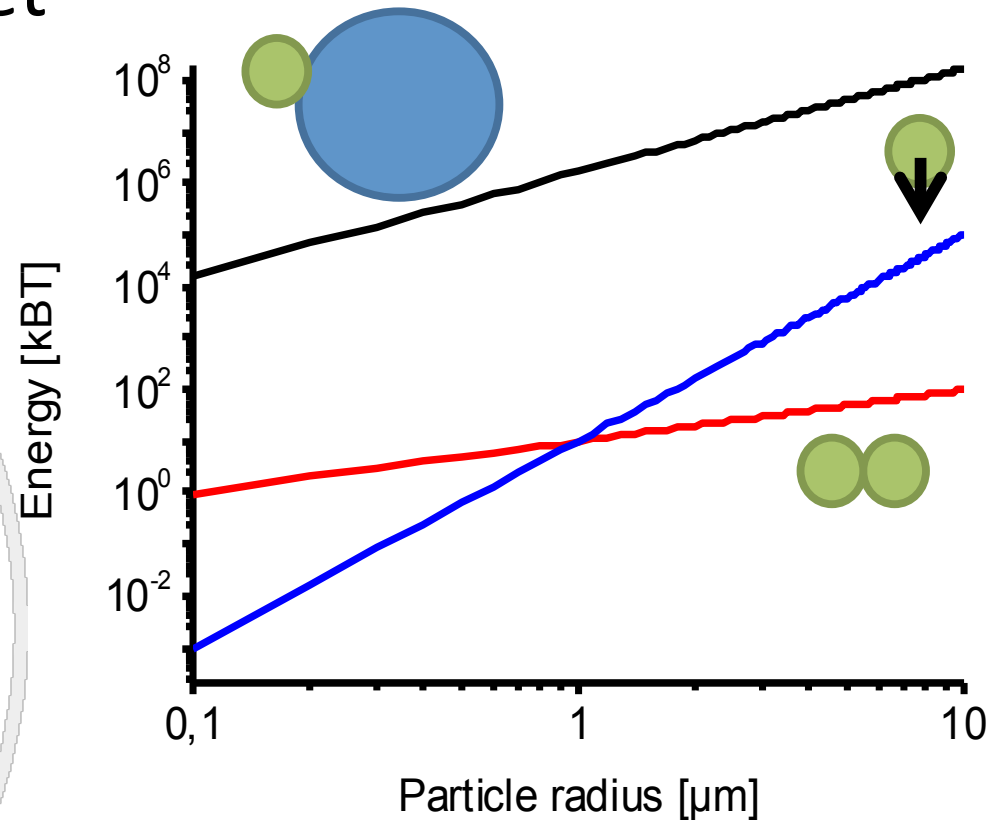
Phase separation



Energy scales

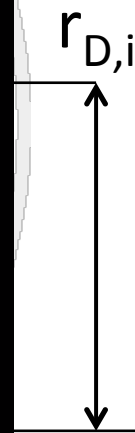
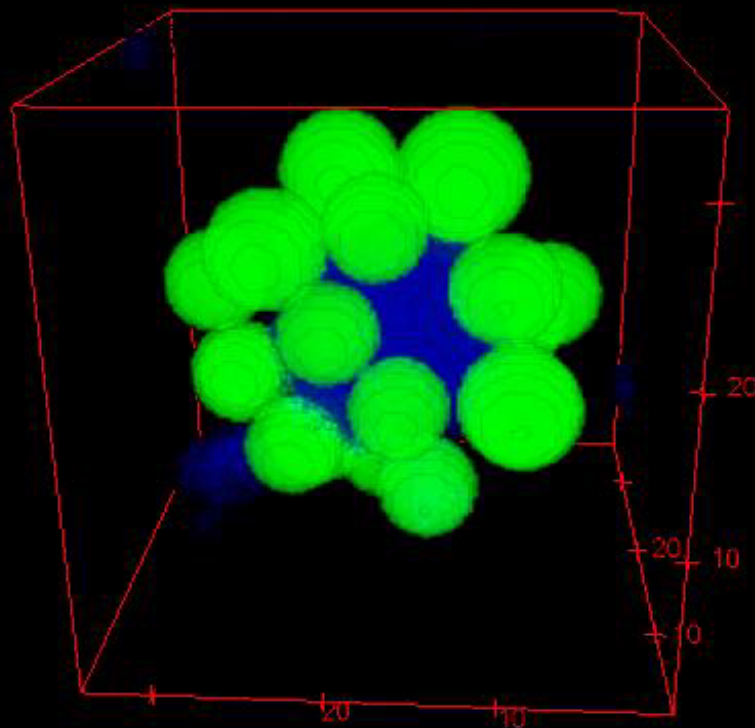
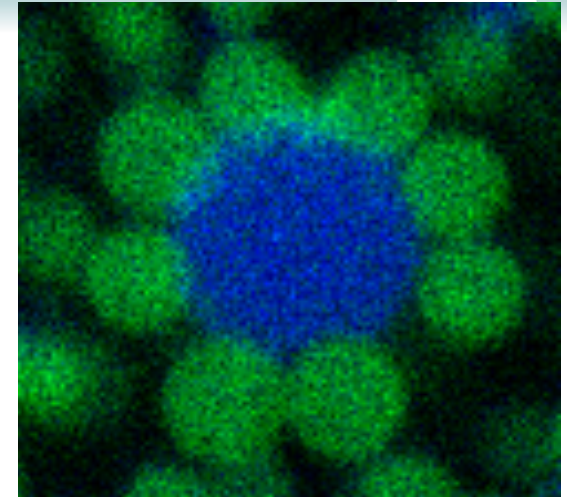


- Model system for wet granular matter
- Single particle resolution
- Binding force dominant
- Droplet – particles ensembles



Drop sizing and deformation

- Particle fixed to interface
- Drop position and diameter



$r_{p,i}$: Radius particle i

$r_{D,i}$: Radius drop i

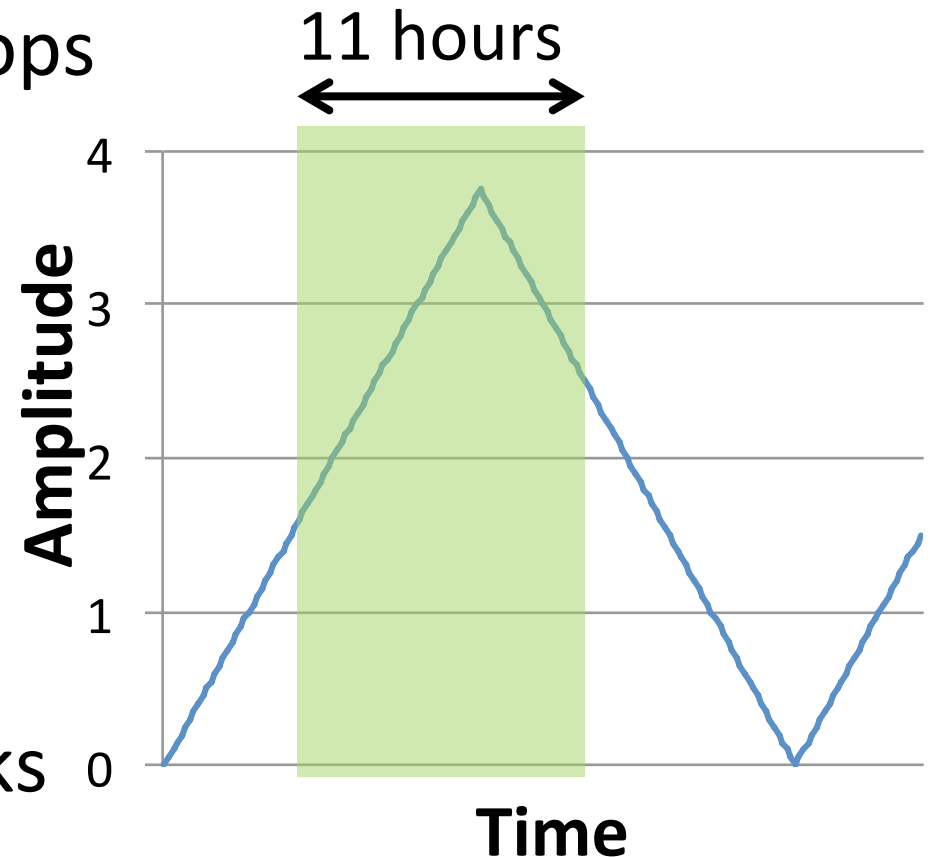
$sub_{p,i}$: submerged part i

Minimise $StdDev(r_{D,i})$

Long shear experiment

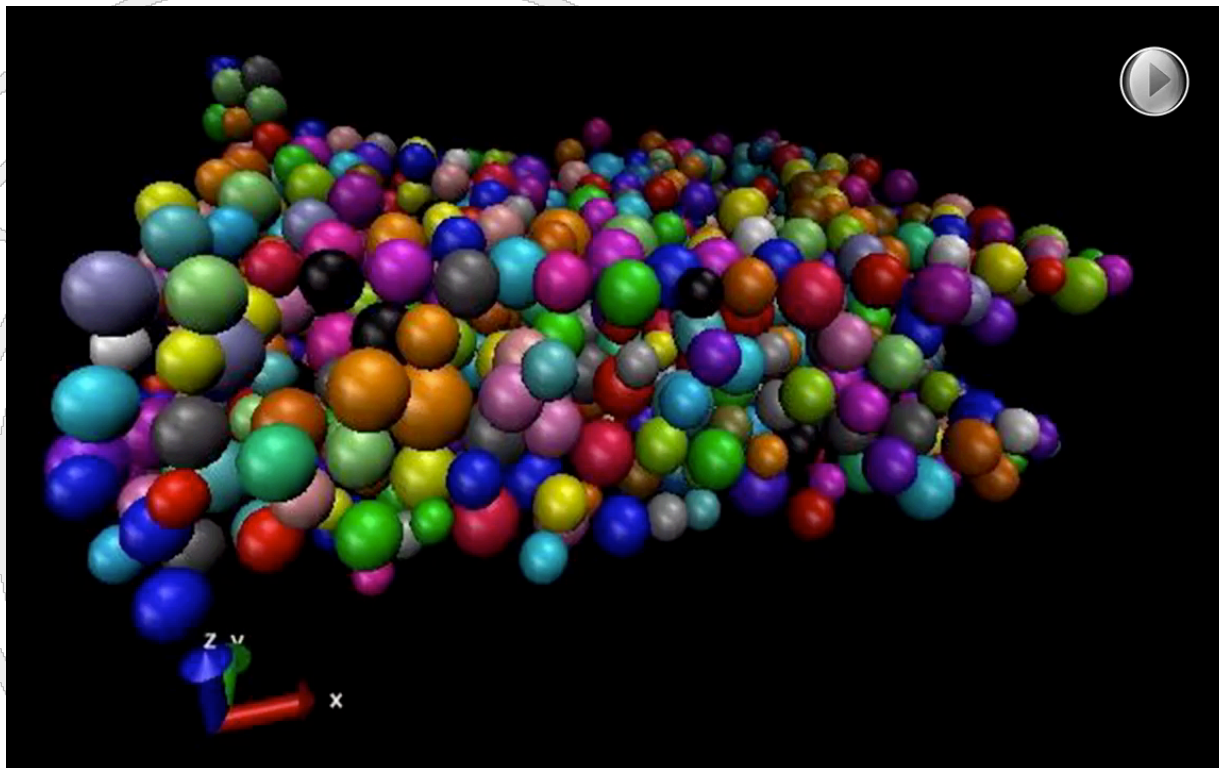
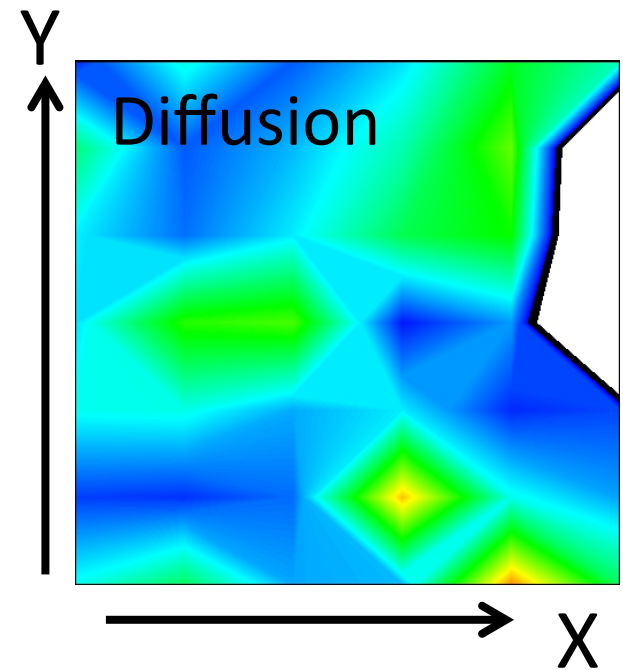
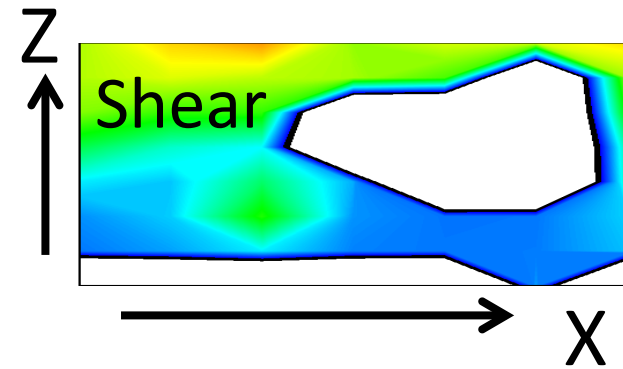


- 150x150x50 μm volume
~1400 particles, 87 drops
- Oscillatory shear
- 3.75 shear amplitude
- Observed:
1.75 – max – 2.5
- 320 3D images
- Position, size and tracks



Results

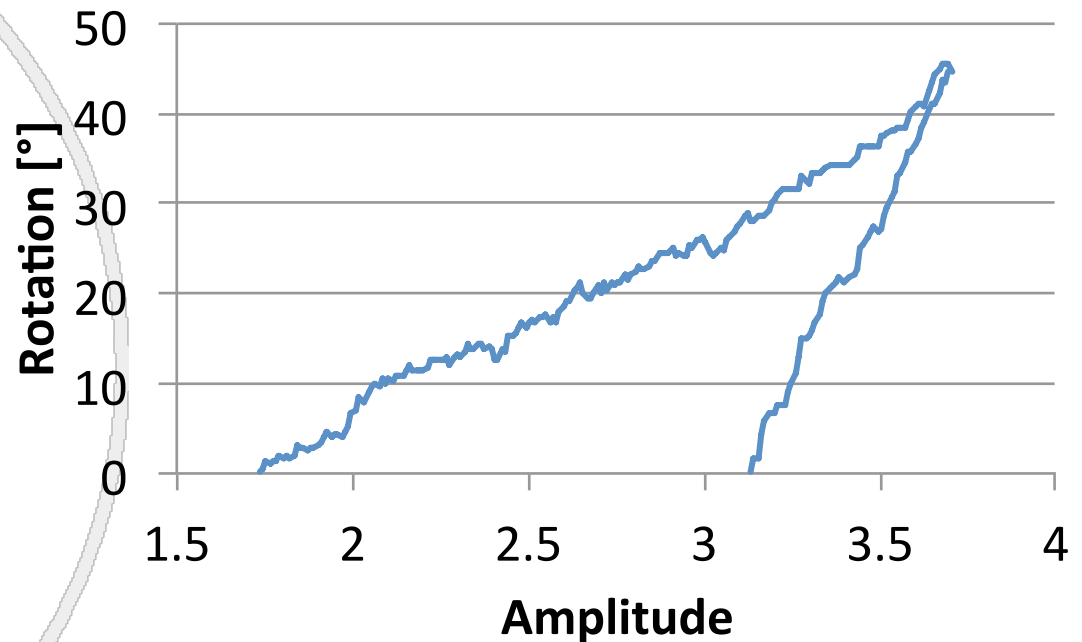
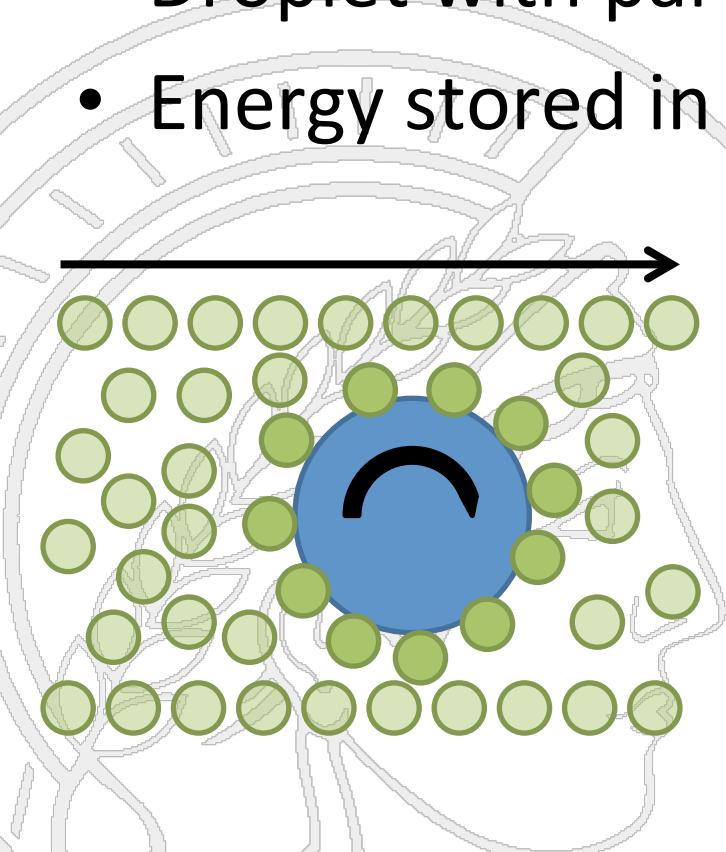
- Temporal and spatial resolved movement in 3D



Rotation of drops

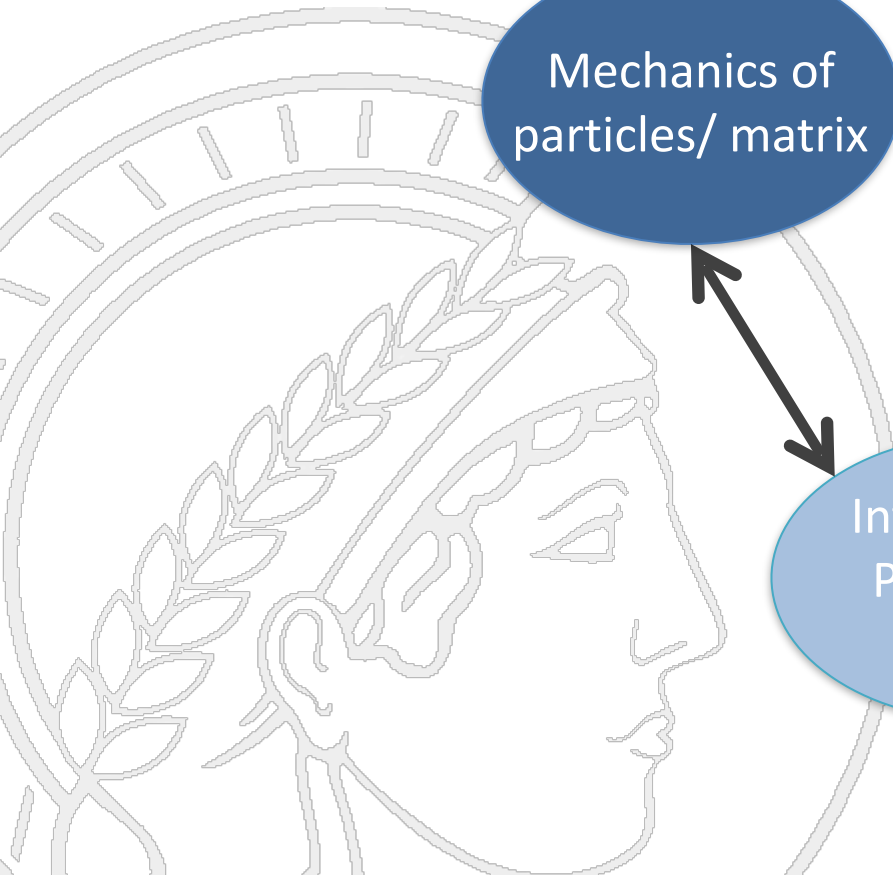
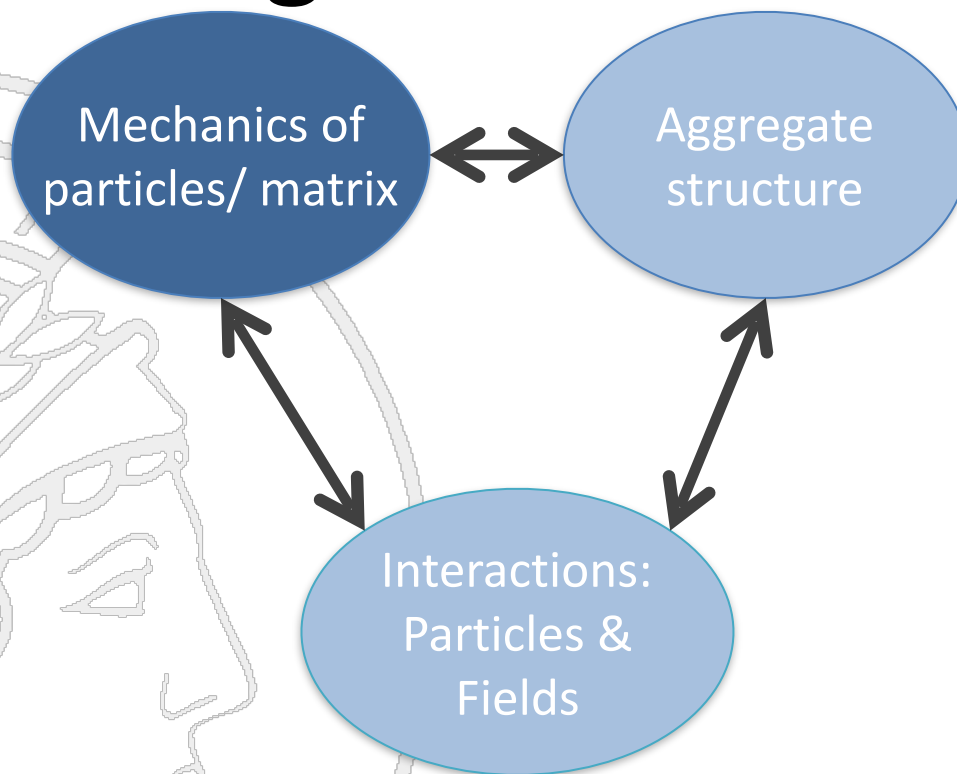


- Particles trapped at surface
- Droplet with particles forms one entity
- Energy stored in droplet

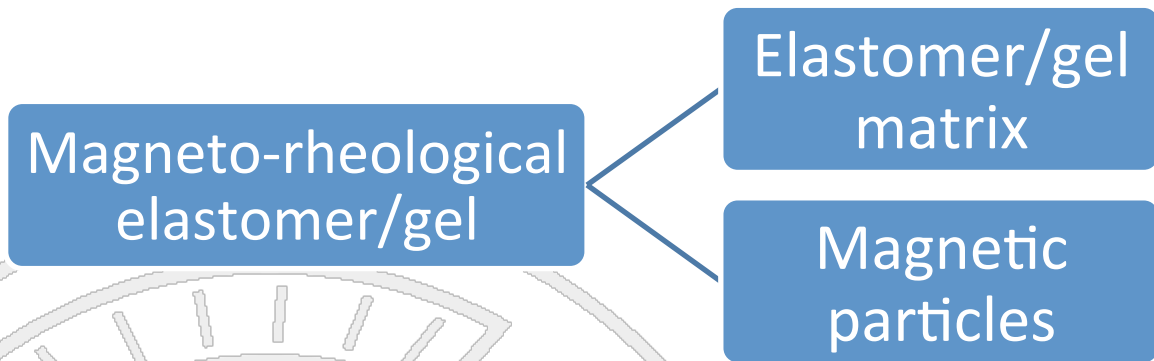




Magnetic fields



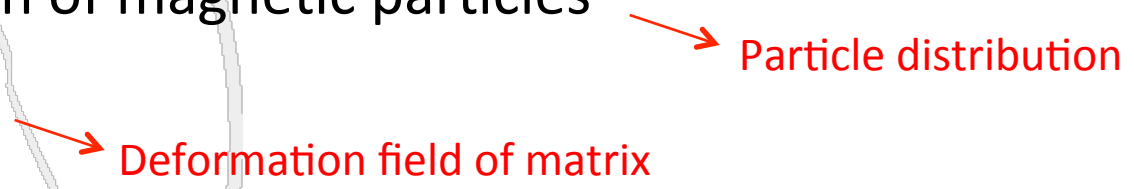
Magnetic gels



Damper
Actuator
...

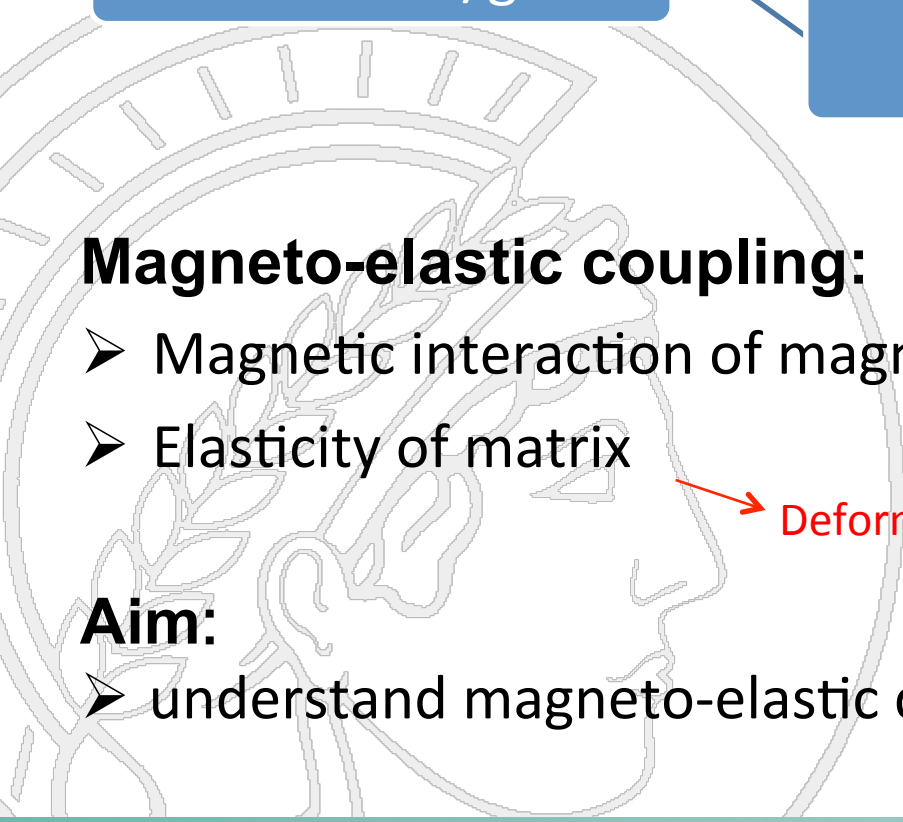
Magneto-elastic coupling:

- Magnetic interaction of magnetic particles
- Elasticity of matrix

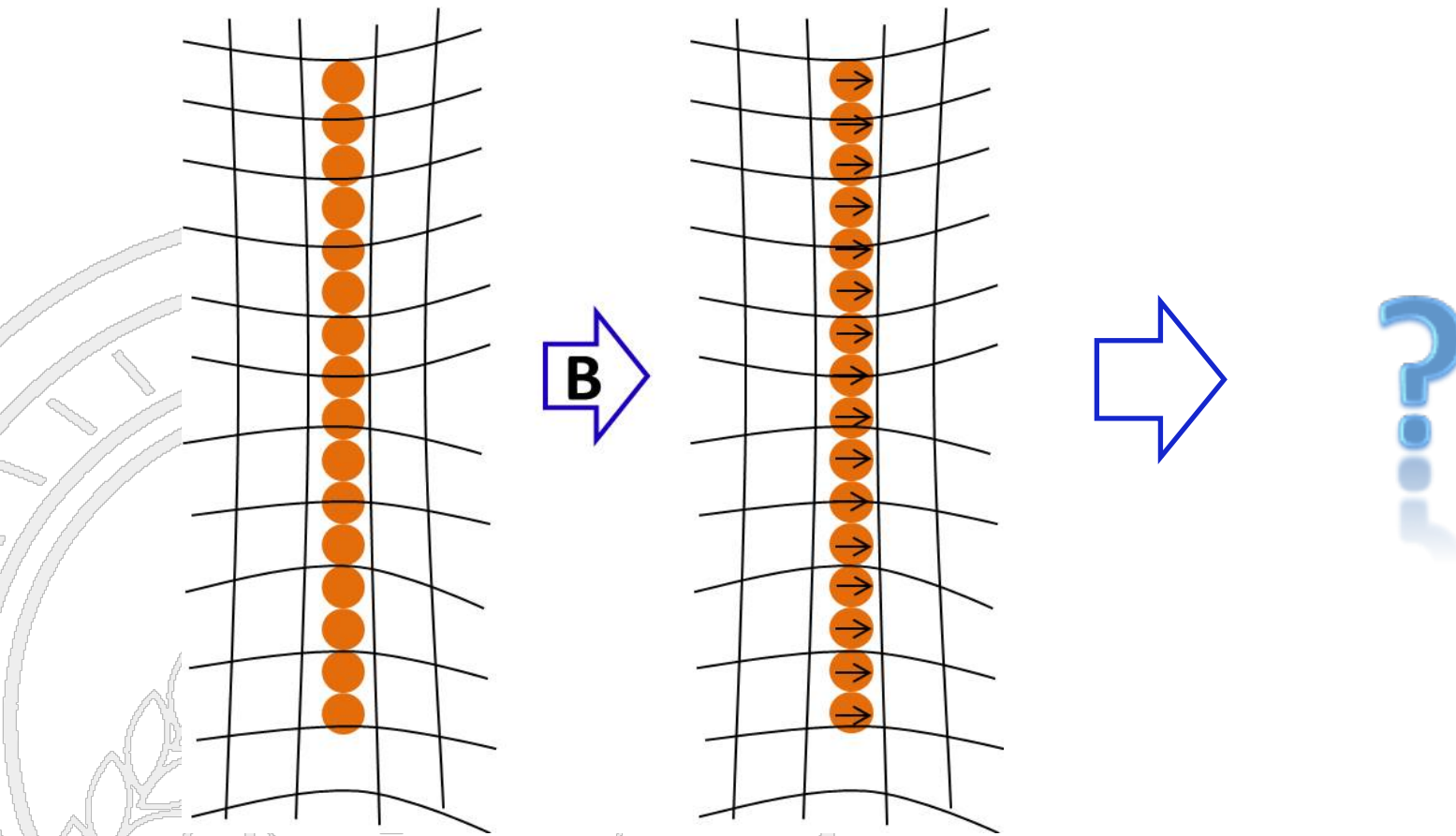


Aim:

- understand magneto-elastic coupling at *microscopic level*

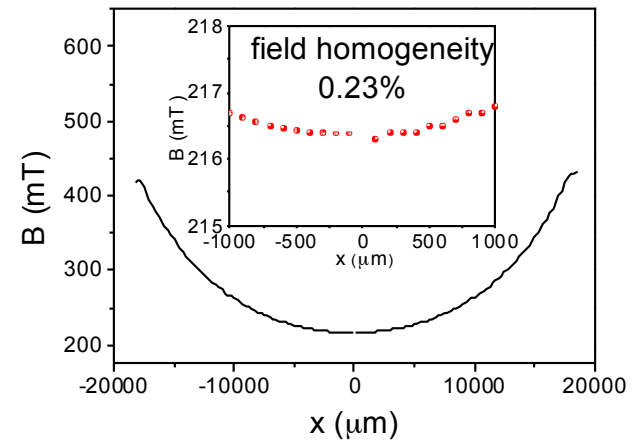
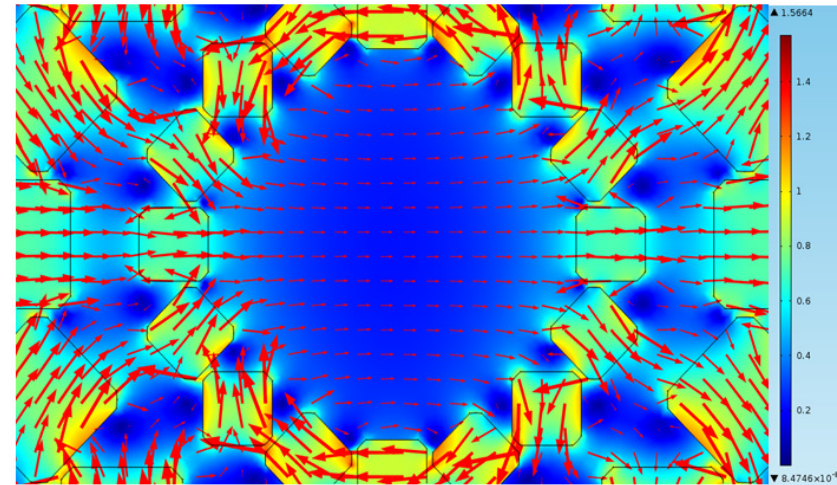
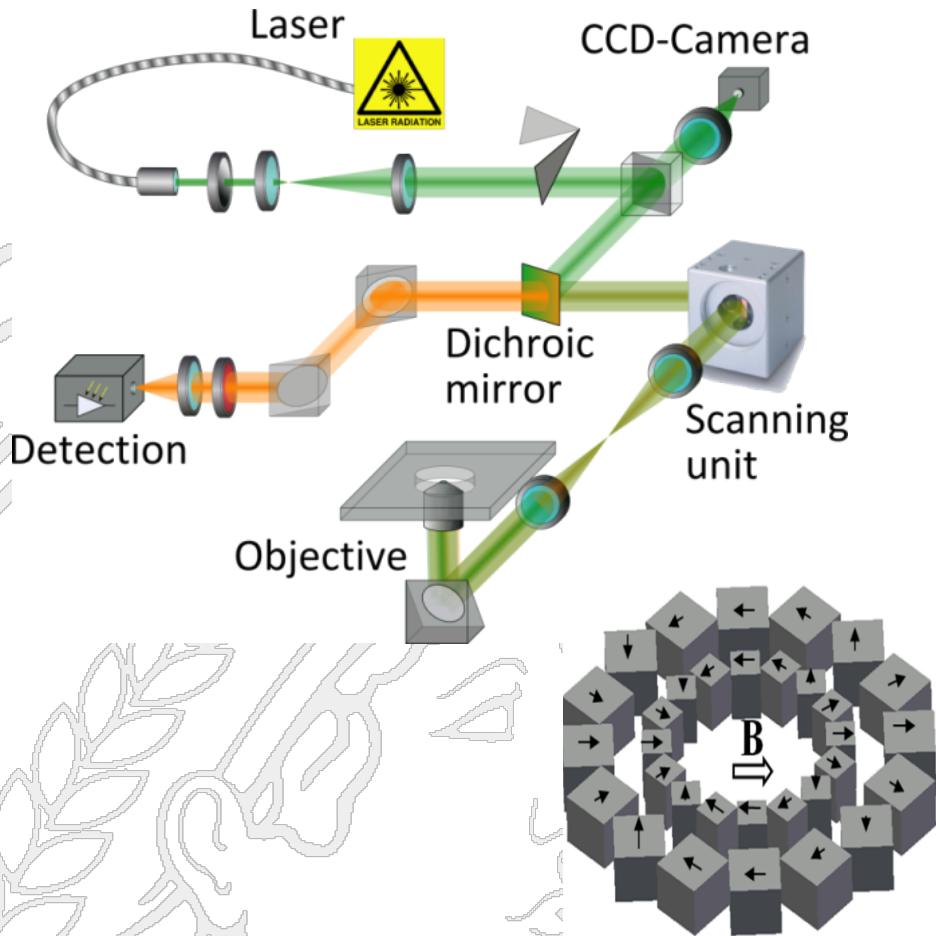


Basic question



Single chain behavior

Magnetic fields

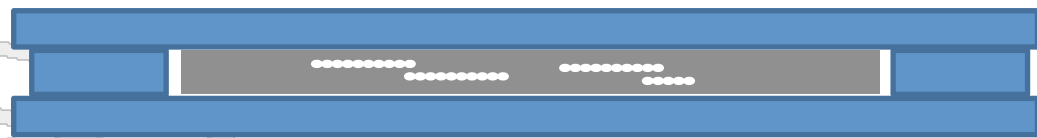


Huang, S., G. Pessot, P. Cremer, R. Weeber, C. Holm, J. Nowak, S. Odenbach, A.

Sample geometry

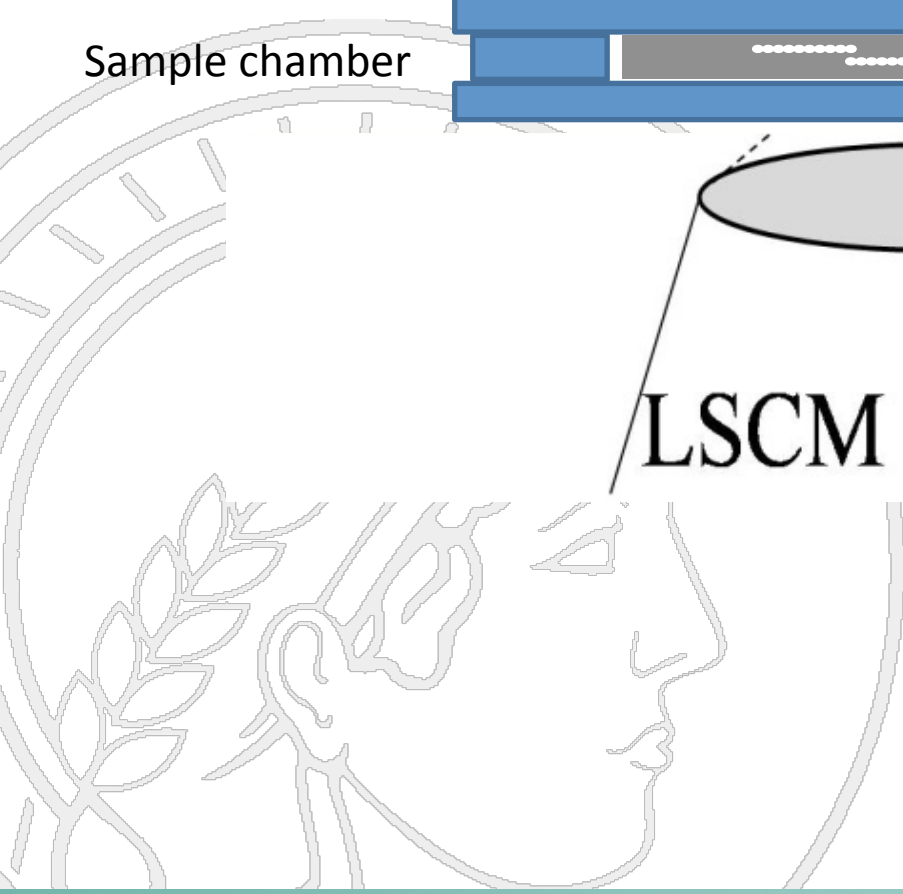


Sample chamber



No.1 standard coverslip
 $\approx 160 \mu\text{m}$

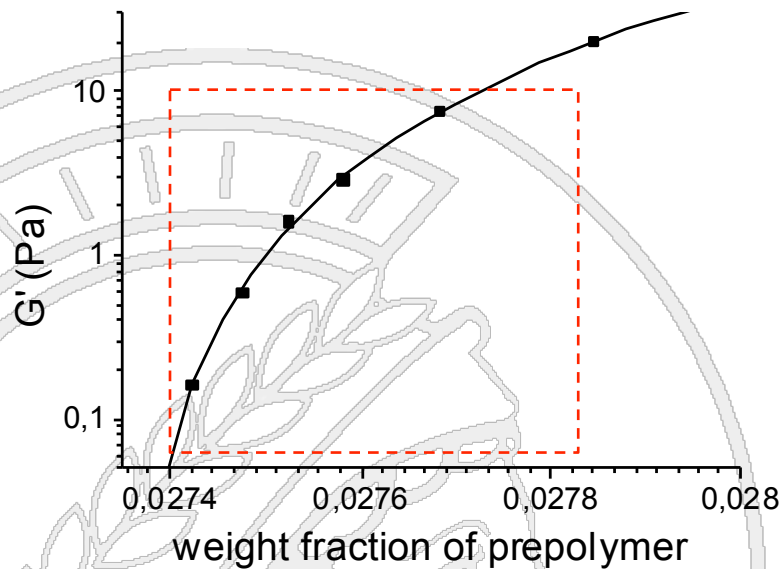
LSCM objective



Sample

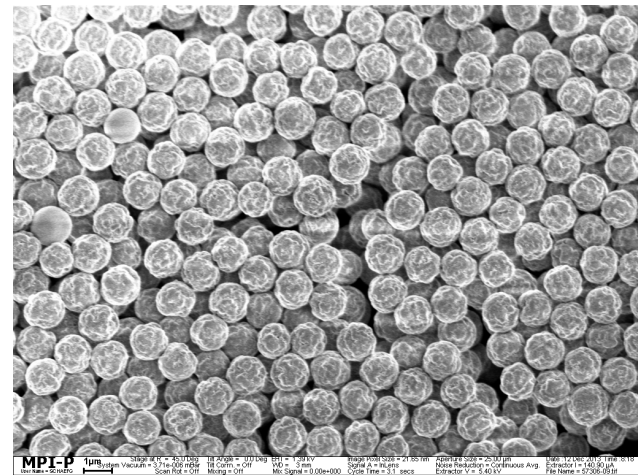


Matrix: PDMS gel



Crosslinking:
Hydrosilation with
platinum catalyst

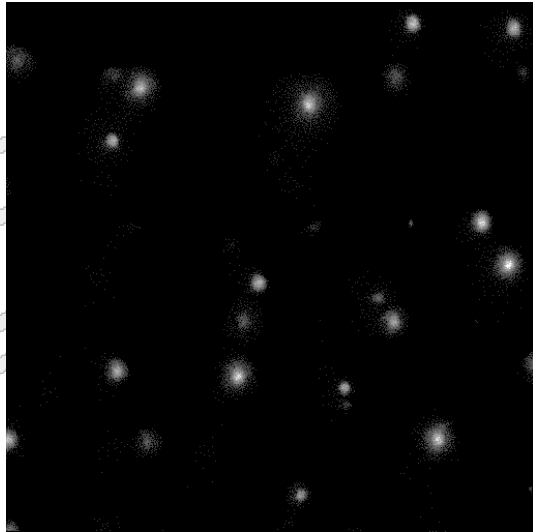
Magnetic particles



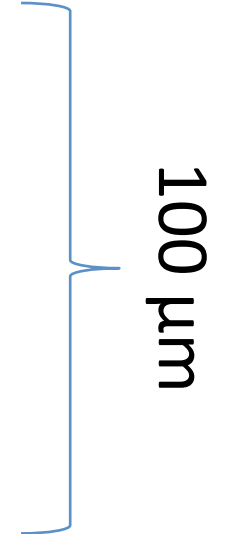
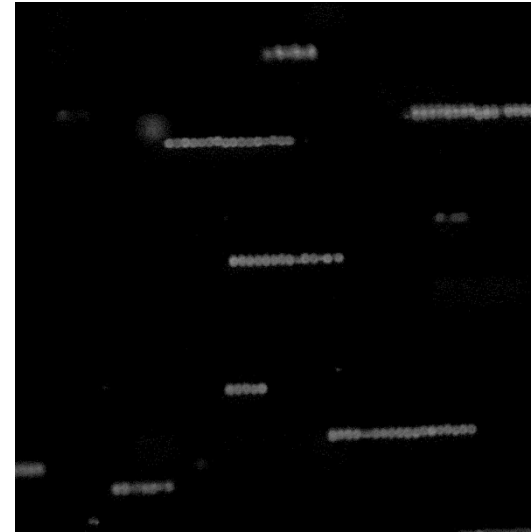
Superparamagnetic
Fluorescent-labeled

0.1 wt%

Confocal images



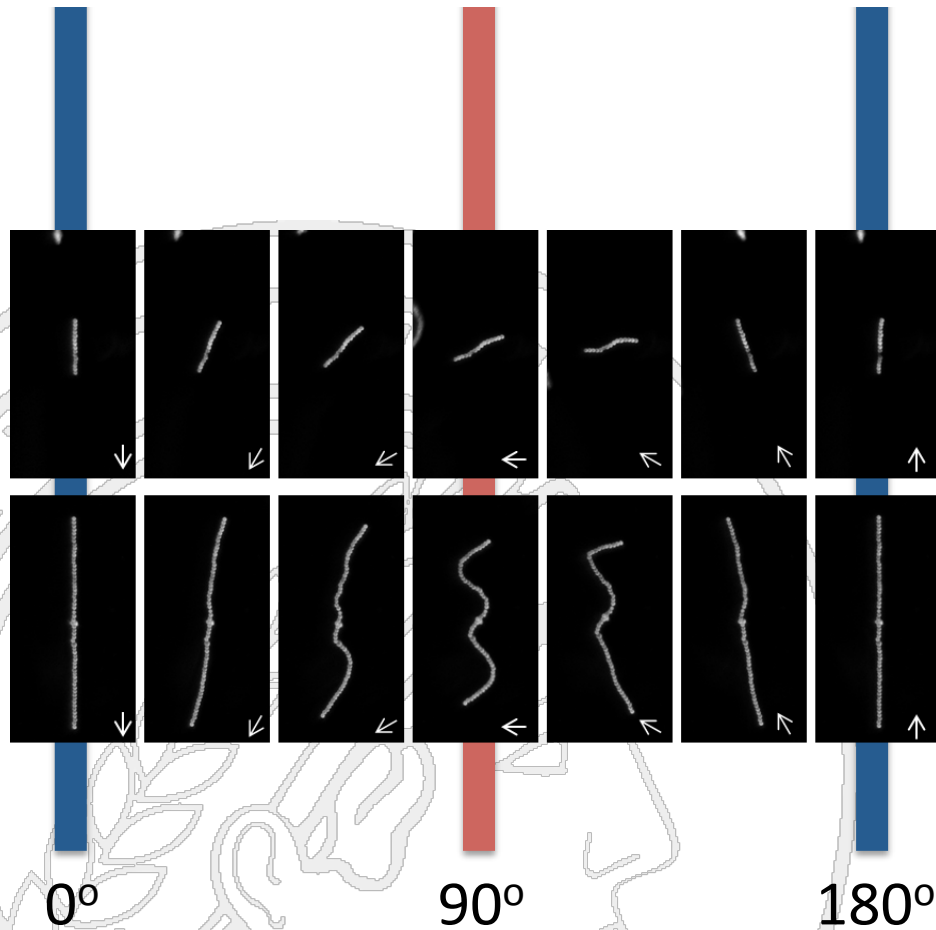
Crosslinked without
applied field



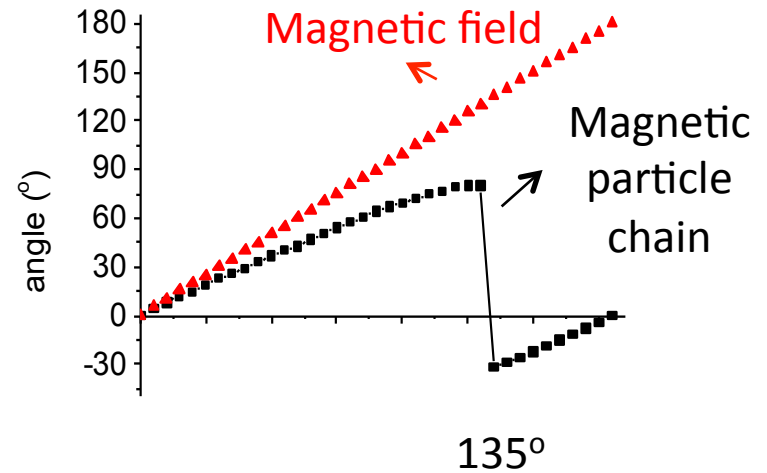
100 μm

Crosslinked with
applied field

Rotating field



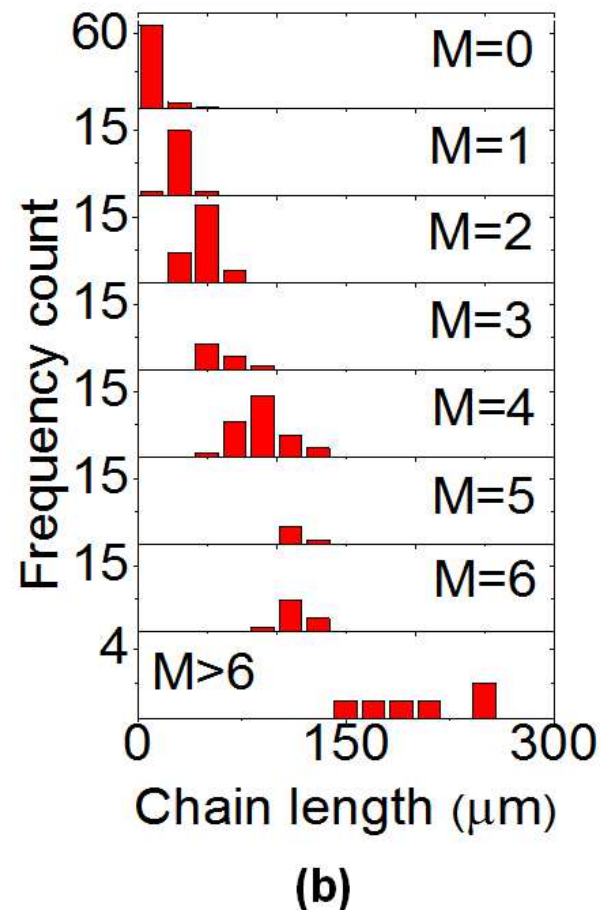
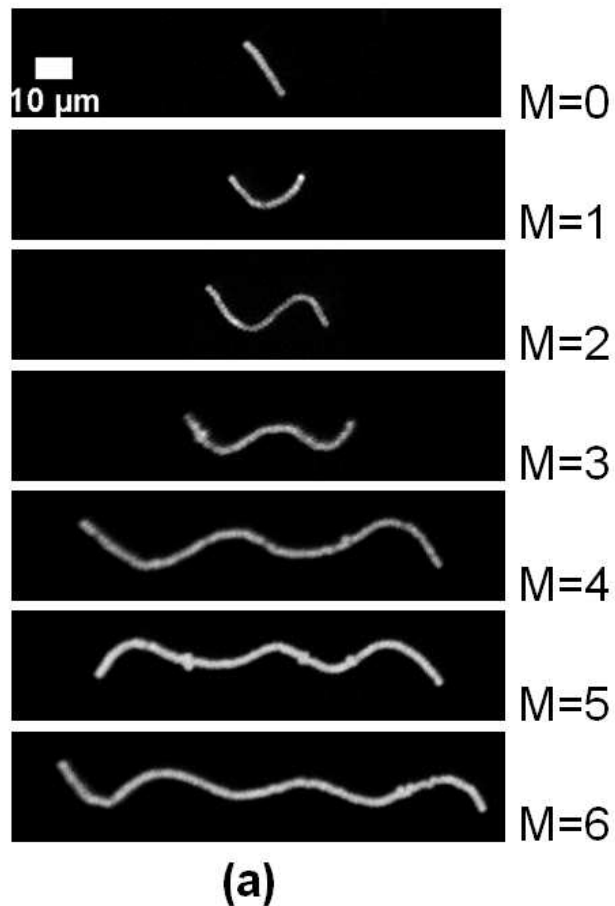
- Short chains rotate
Long chains buckle
- Hysteresis



$G' \approx 0.2 \text{ Pa}$, $B = 216 \text{ mT}$

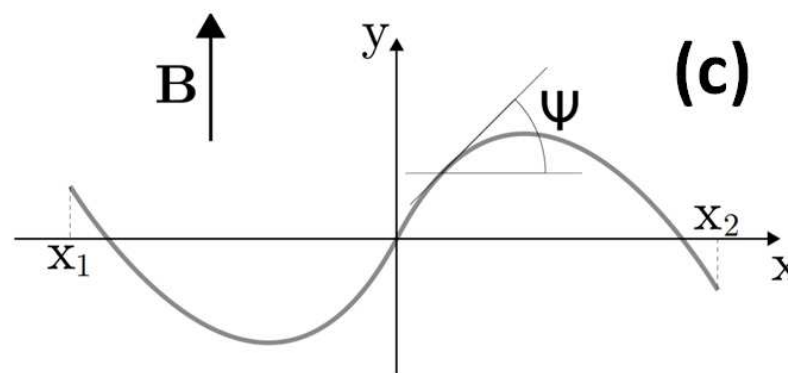
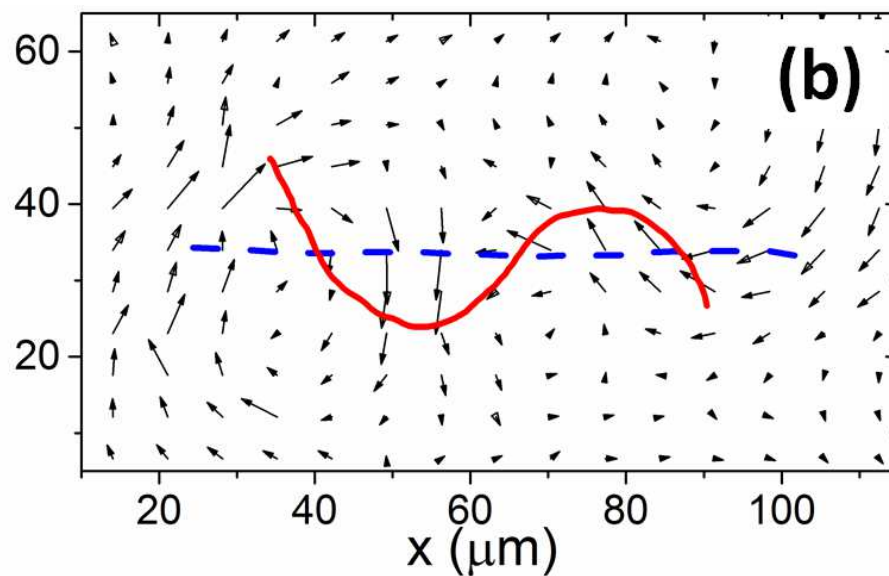
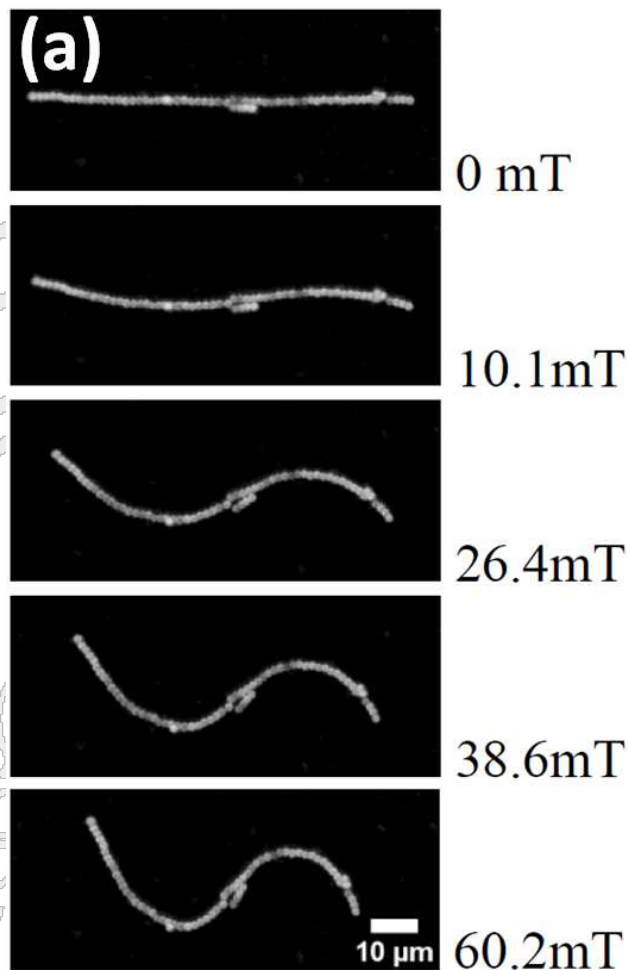
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Modes of deformation



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Deformation field



Huang, S., G. Pessot, P. Cremer, R. Weeber, C. Holm, J. Nowak, S. Odenbach, A.

Modelling



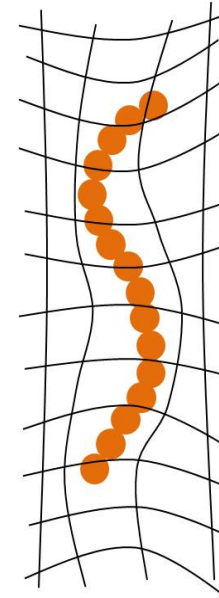
Magnetic chain tends to align to magnetic field direction to decrease magnetic energy.



buckling



Surrounding elastic network increases elastic energy due to deformation.

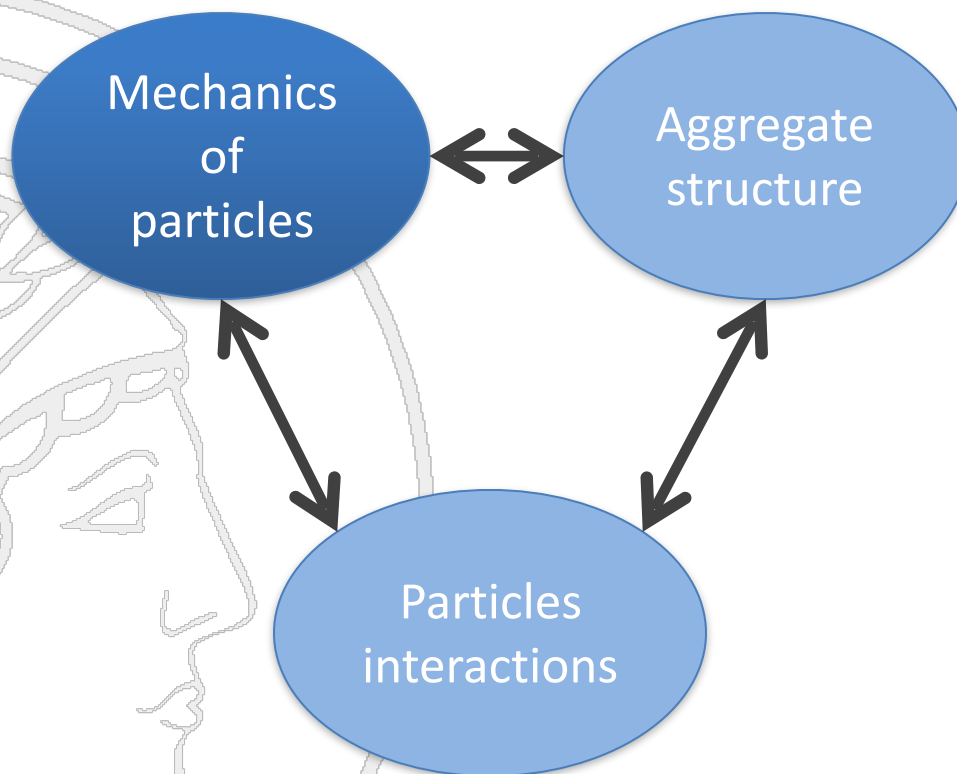


Wave length = $f(G', H_0, l_0)$
Amplitude = $g(G', H_0, l_0)$

$$E_{total} = E_{elastic} + E_{magnetic}$$



Soft Colloids



Colloids in liquid crystals



Network formation

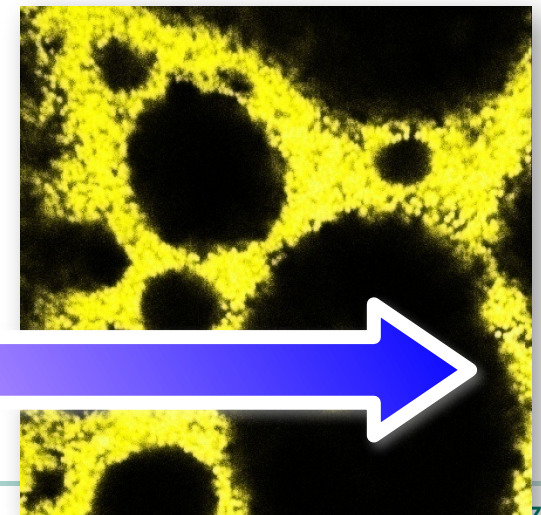
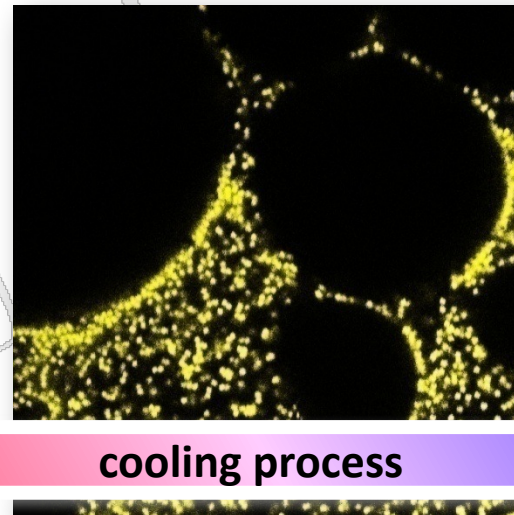
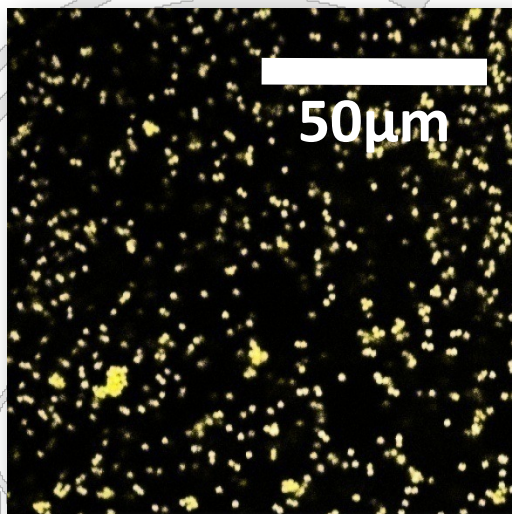
- Colloids dispersed in isotropic phase
- Coexistence of nematic and isotropic phase
- Interaction of nematic liquid crystal, impurities and colloids



Stark, Phys. Rep. (2001) 351, 387

Meeker et al., Phys. Rev. E (2000) 61, R6083

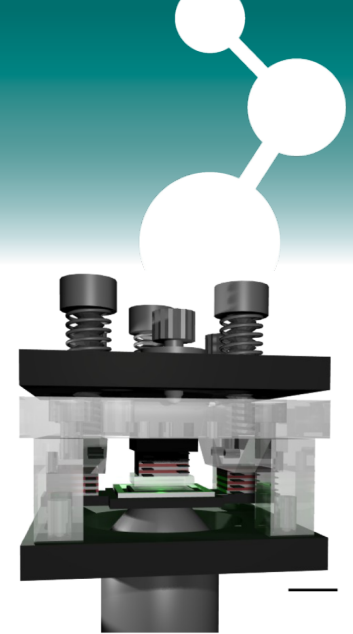
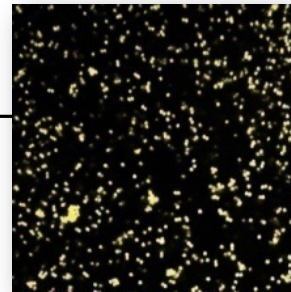
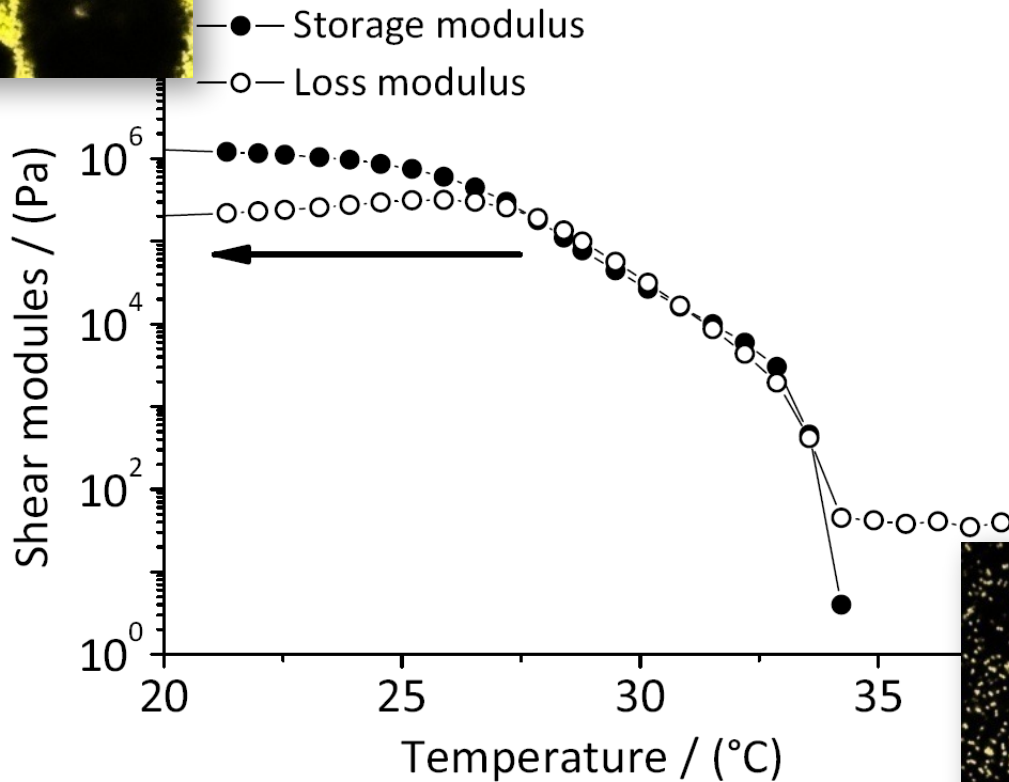
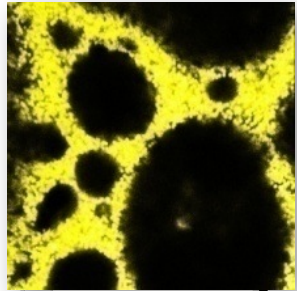
Vollmer et al., Langmuir (2005) 21, 4921



5CB

Rheological properties

Temperature dependency

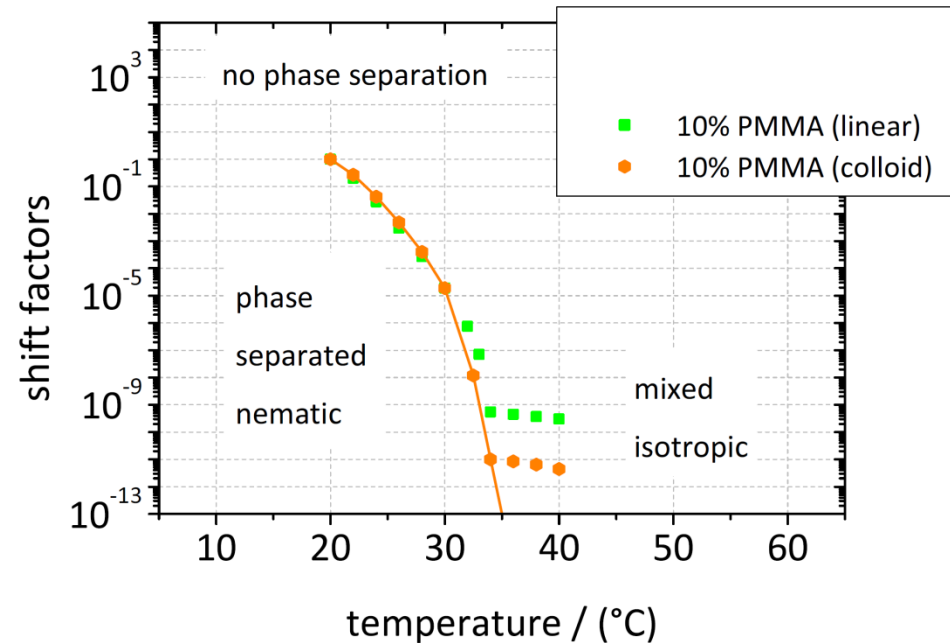
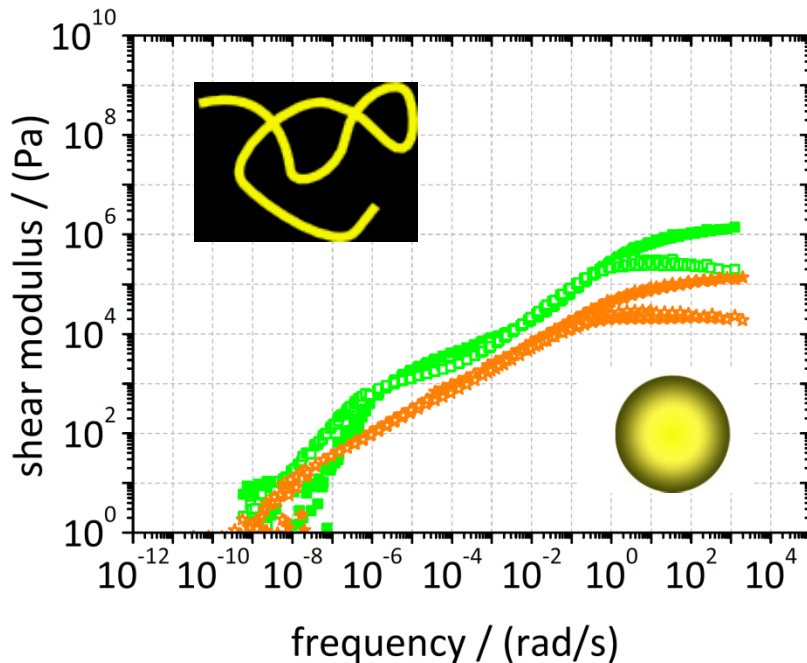


- frequency dependency gives further physical insight

Superposition



Phase separation master curves and shift factors

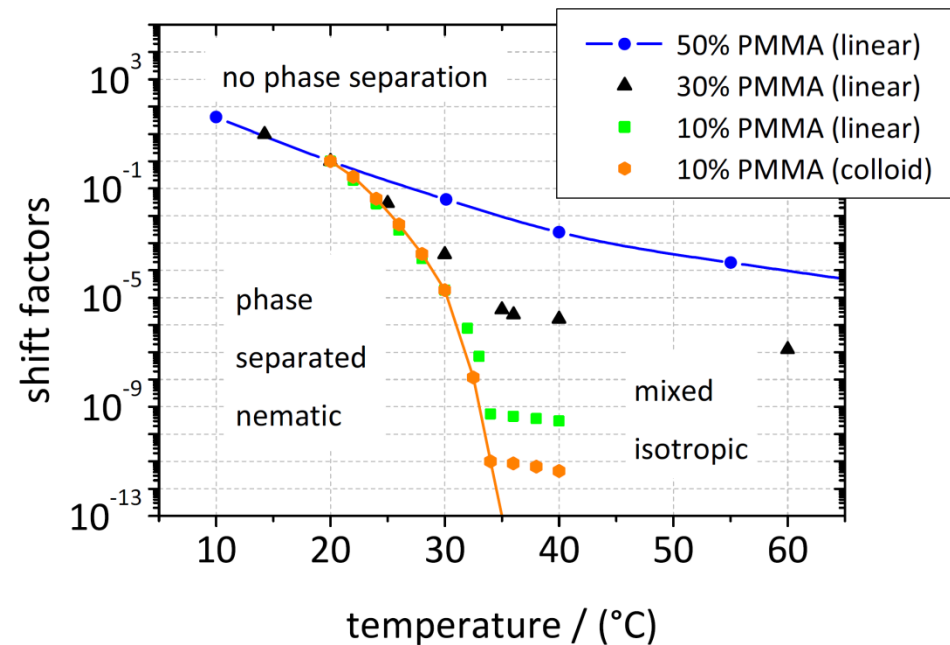
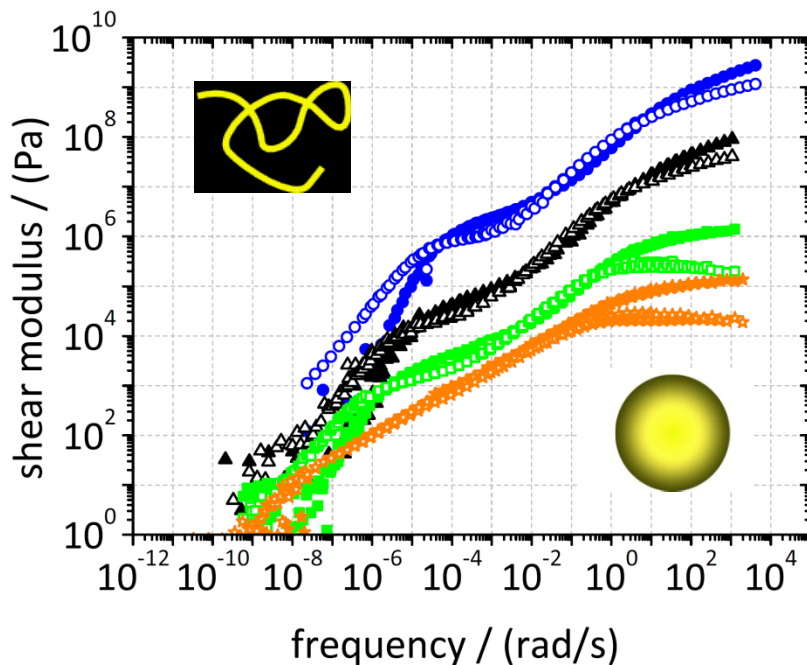


- Hard core reflected in suppressed rubbery plateau
- Shift factors universal, therefore bound to the phase diagram
- Rheology dominated by plastized PMMA, compare polymer dispersed LC

Superposition



Phase separation master curves and shift factors



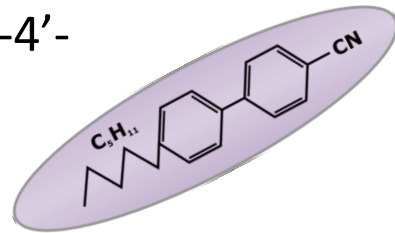
- Hard core reflected in suppressed rubbery plateau
- Shift factors universal, therefore bound to the phase diagram
- Rheology dominated by plastized PMMA

Experimental: Sample system

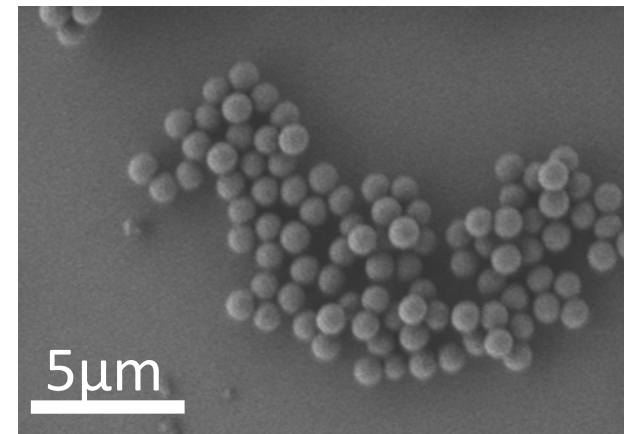
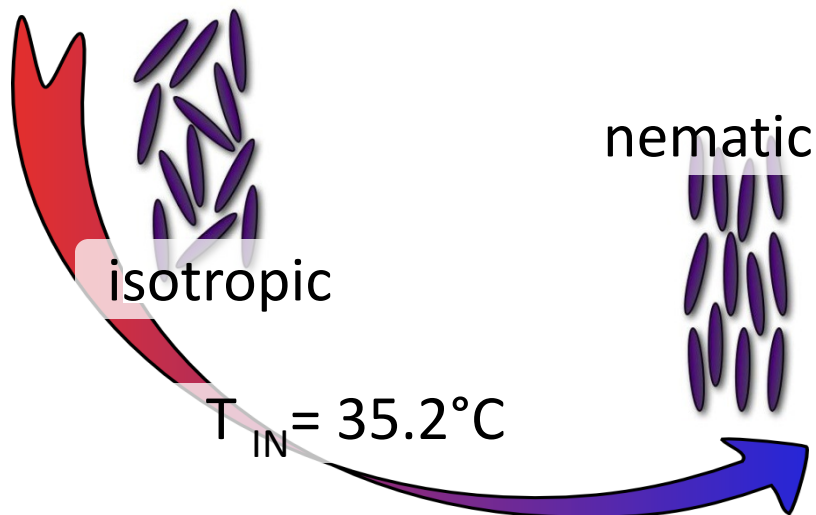


Composition

- Thermotropic liquid crystal
5CB (4-n-pentyl-4'-cyanobiphenyl)



- PMMA– particles, partially crosslinked, sterically stabilized (poly 12-hydroxy stearic acid)

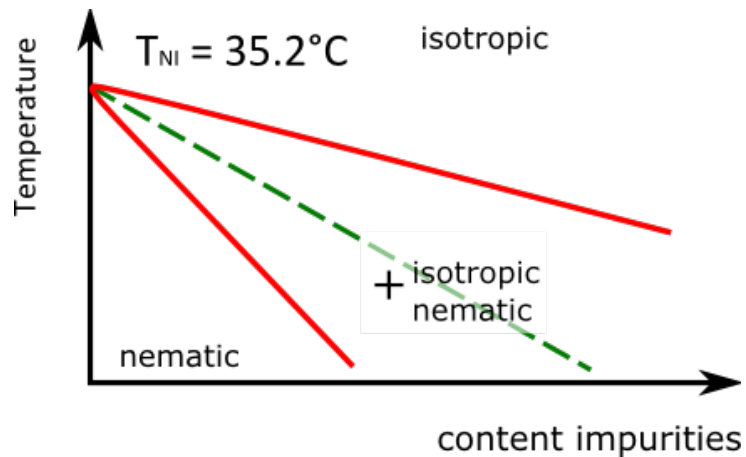


Colloids at NI Interface

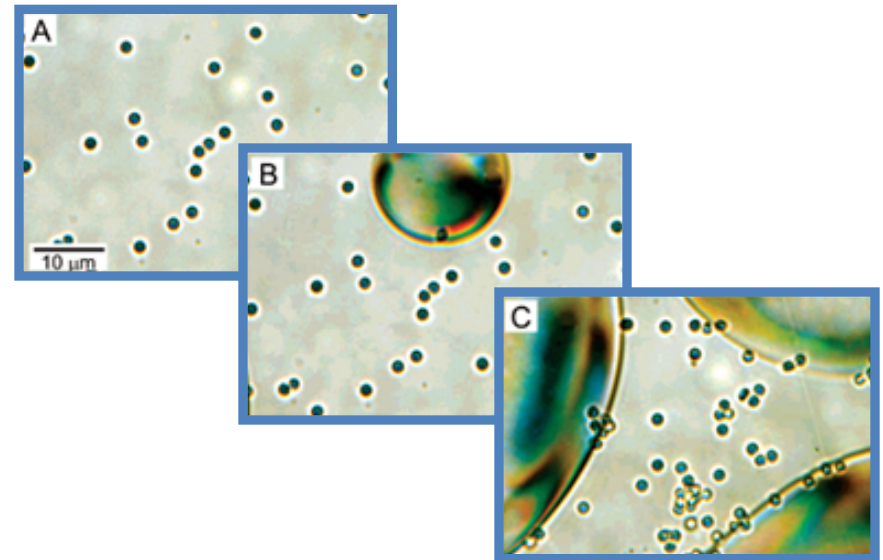


- Kinetics and mechanisms at the phase transition

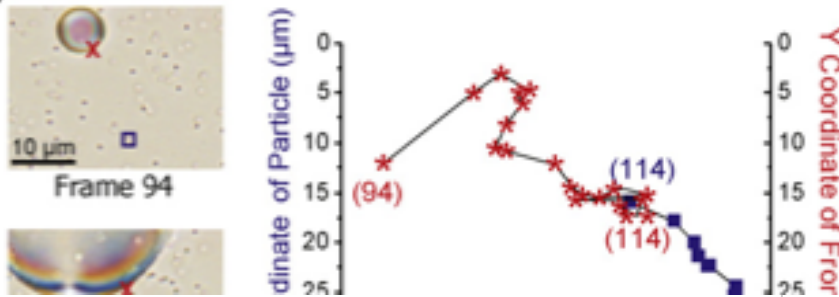
→ Coexistence of nematic and isotropic phase due to impurities



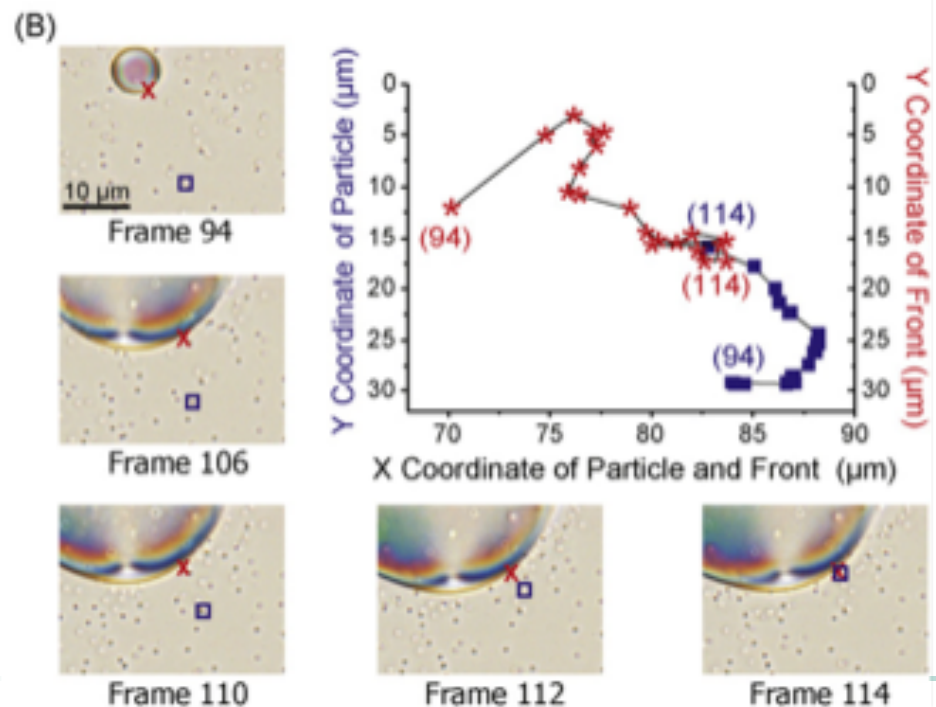
→ Interaction with NI interface and impurities



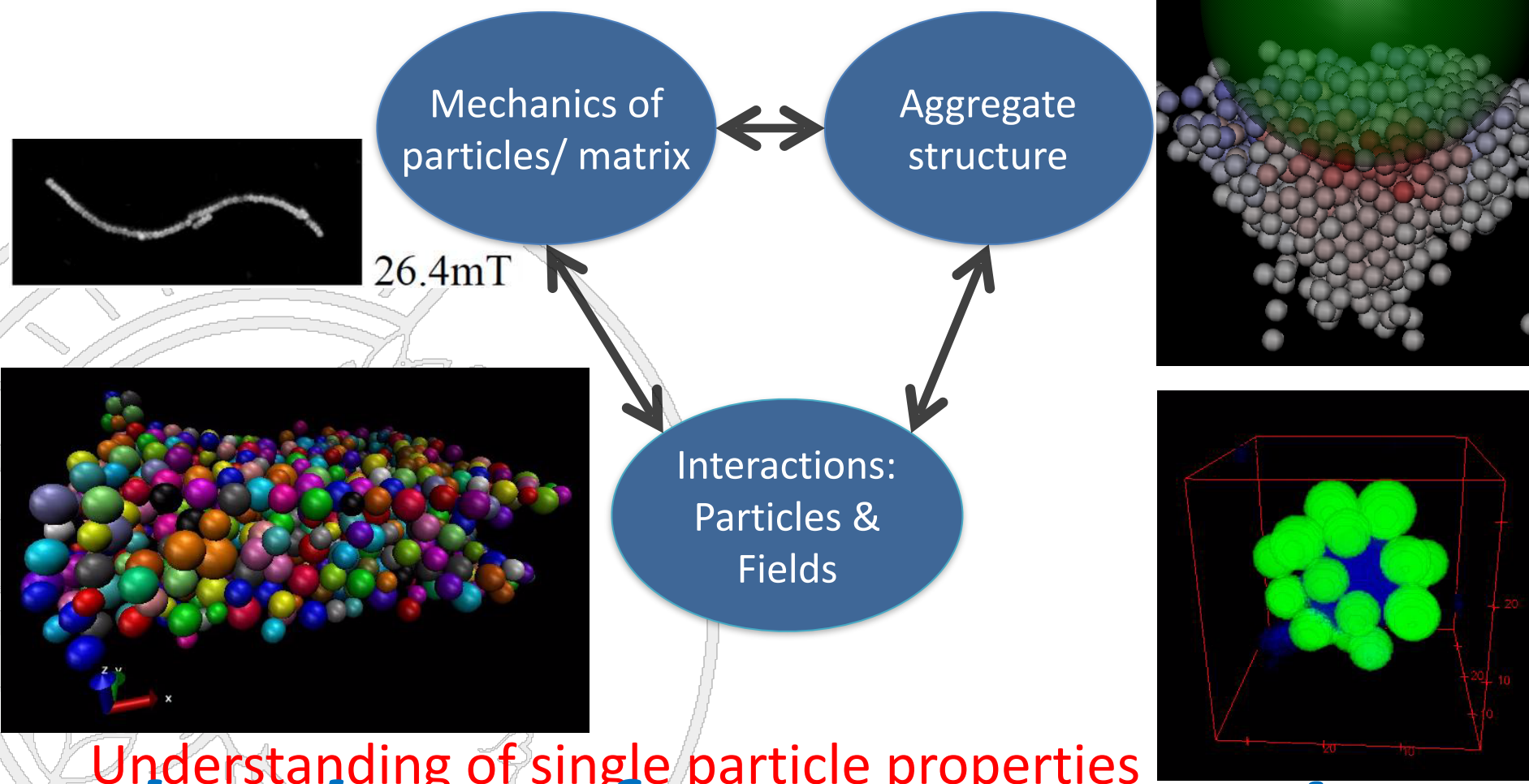
Colloids at NI Interface



Interaction with the interface?



Summary



Understanding of single particle properties
Thank you for your attention
=> tunable macroscopic properties